

# **“How to mine a complex spectrum”**

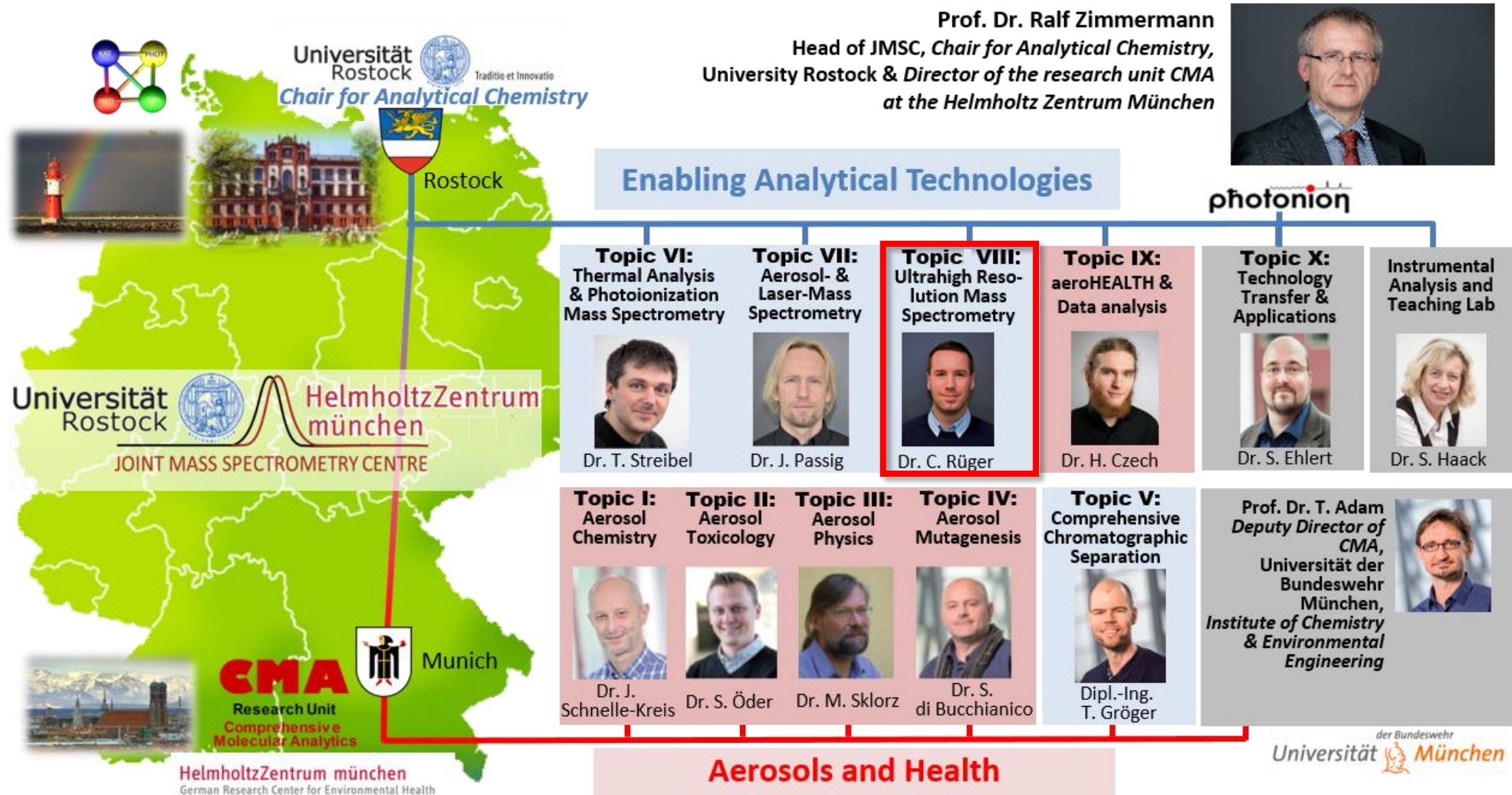
or

## **Dealing with Complex Mass Spectrometric Data in the Framework of the EU\_FT-ICR\_MS Project**

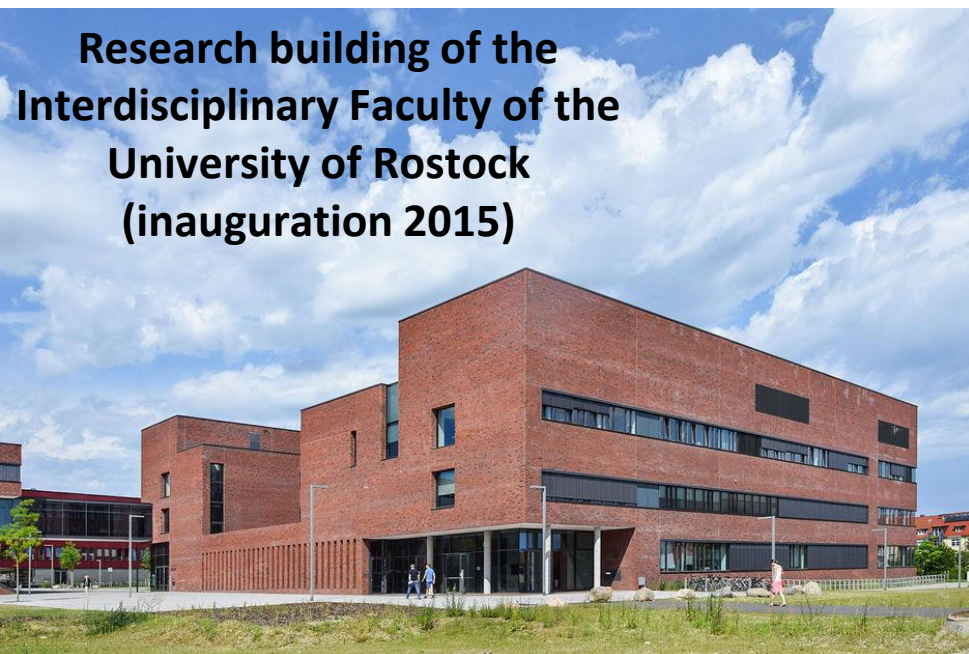
**Christopher P. Rüger**  
University of Rostock



- **Chemical Complexity and Analytical Capabilities** “What does complexity really means?”
- **Challenges in High-Resolution Mass Spectrometry** “Can we cope with the complexity by FT-ICR MS?”
- **From Transient to Mass Spectrum**
- **Utilizing the Mass Spectrum Data and Feature Detection** “Where are the good data?”
- **From Mass to Chemistry** “How to convert numbers to science?”
- **Handling Attribution Data** “Thousands of Elemental Compositions and now?”
- **Research Examples from our Group** – From Energy Transition to Environmental Sciences
- **Summary and Outlook** “Are we there yet?”







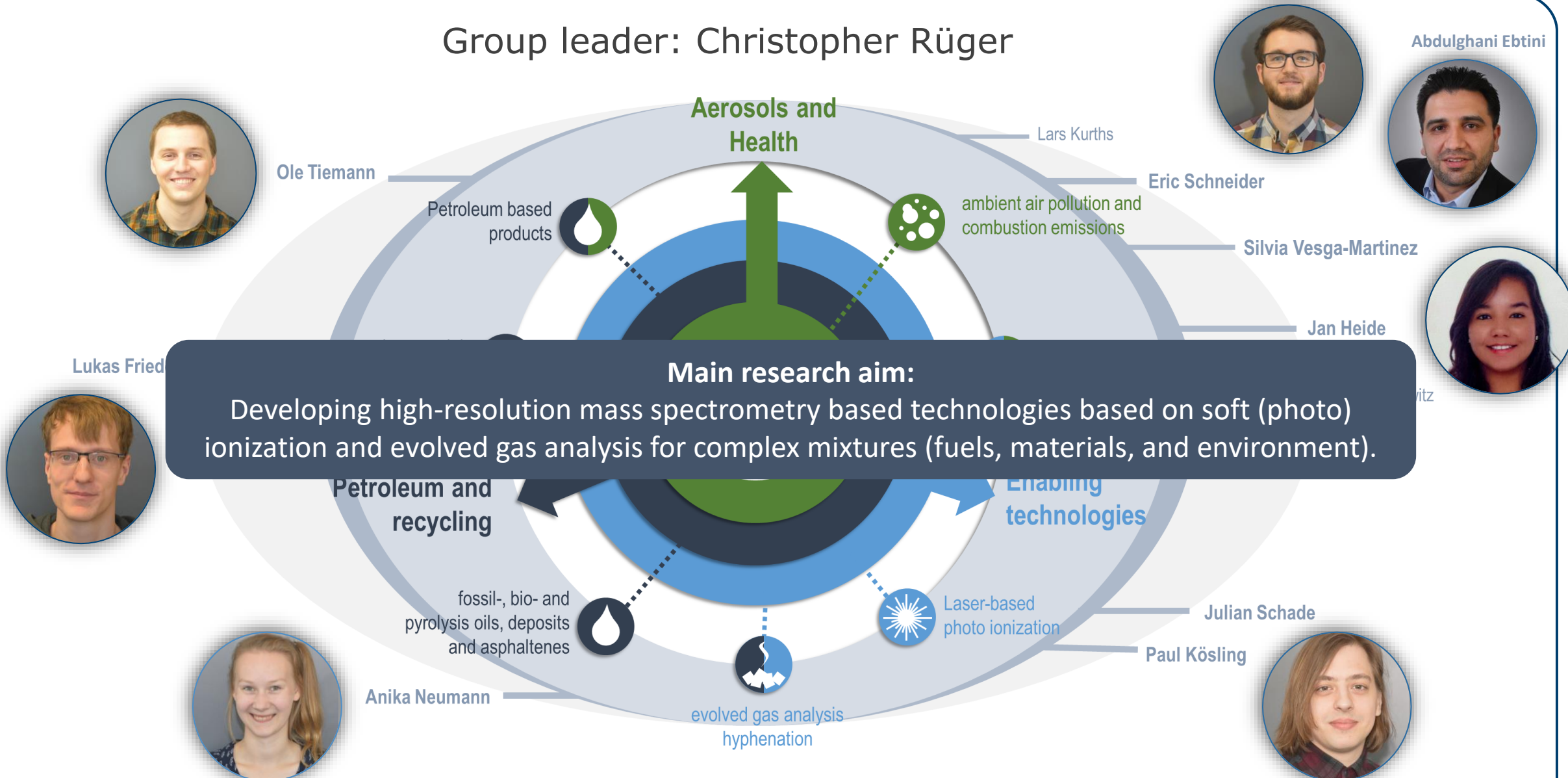
**New extension building  
of the Institute of  
Chemistry of the  
University of Rostock  
(inauguration 2022)**



**Some insights into the mass spectrometry laboratories**

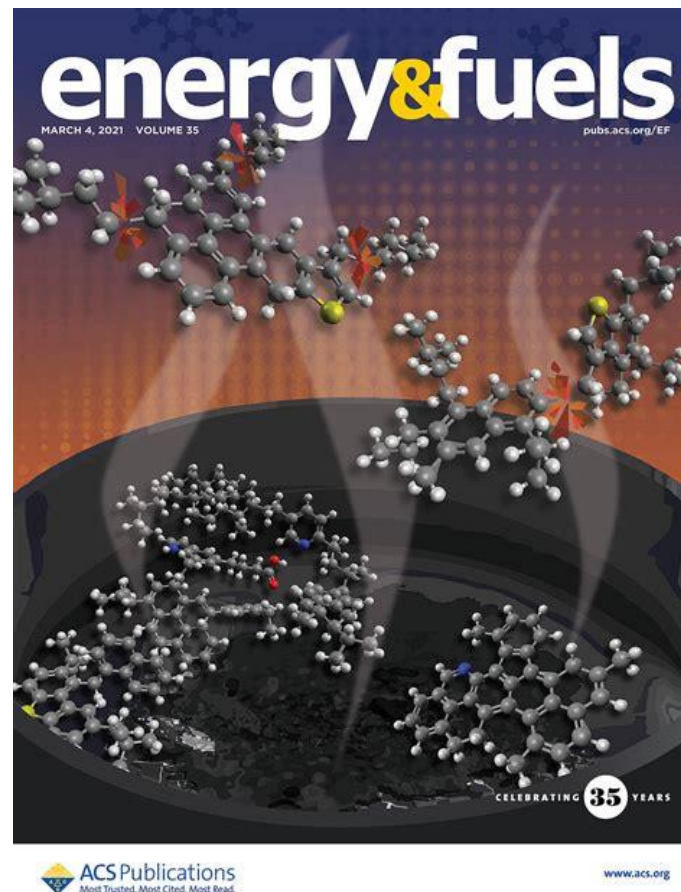


Group leader: Christopher Rüger

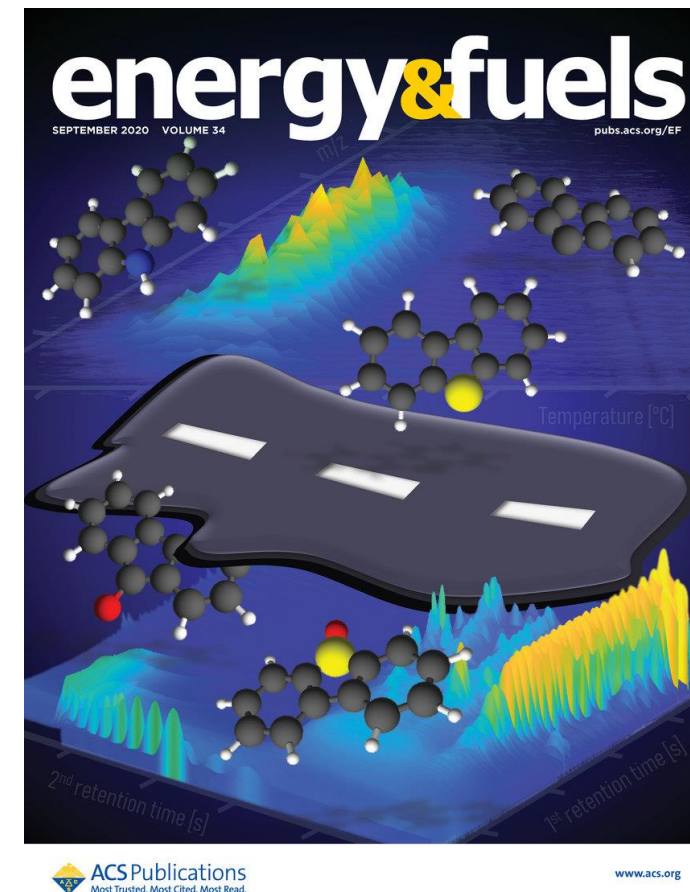




January 2021 – Thesis Defense Anika Neumann (COVID-19 affected)



Neumann et al., Energy and Fuels 2021, 35, 5, 3808–3824



Neumann et al., Energy and Fuels 2020, 34, 9, 10641–10654

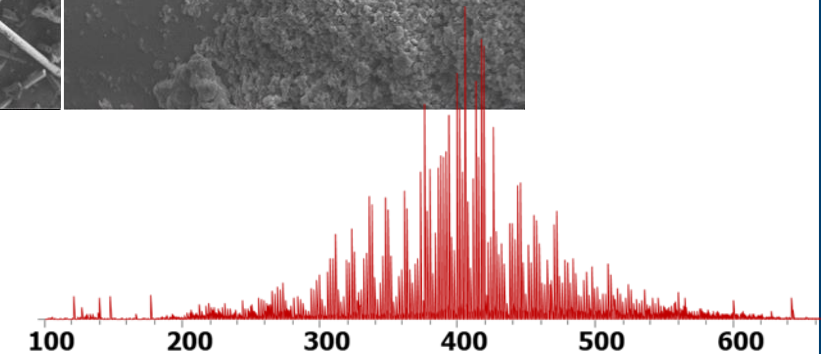
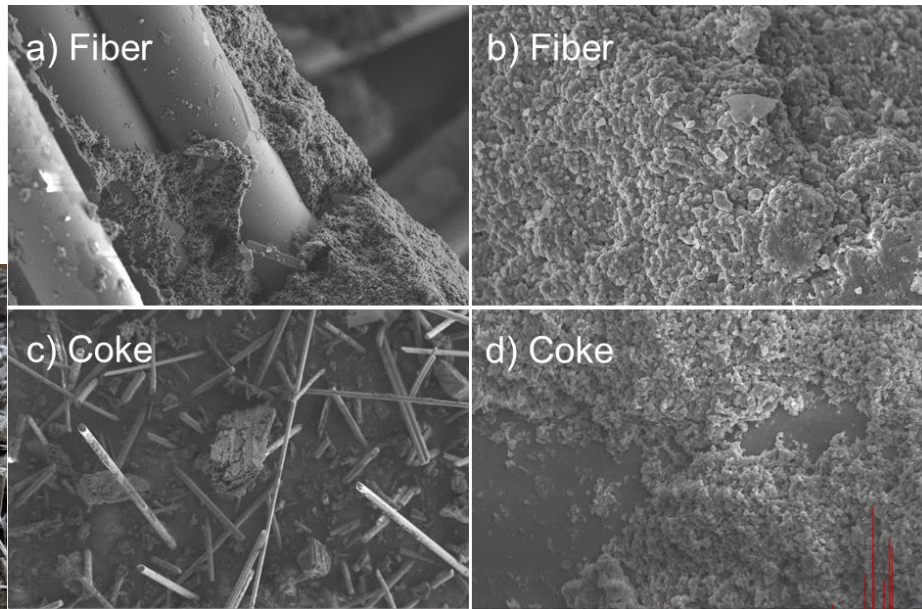


## Fostering Energy Transition and Recycling as Research Topics

PhD Thesis of Lukas Friederici on composite recycling guided by thermal analysis mass spectrometry



Sampling next to pilot plant for  
pyrolytic composite recycling

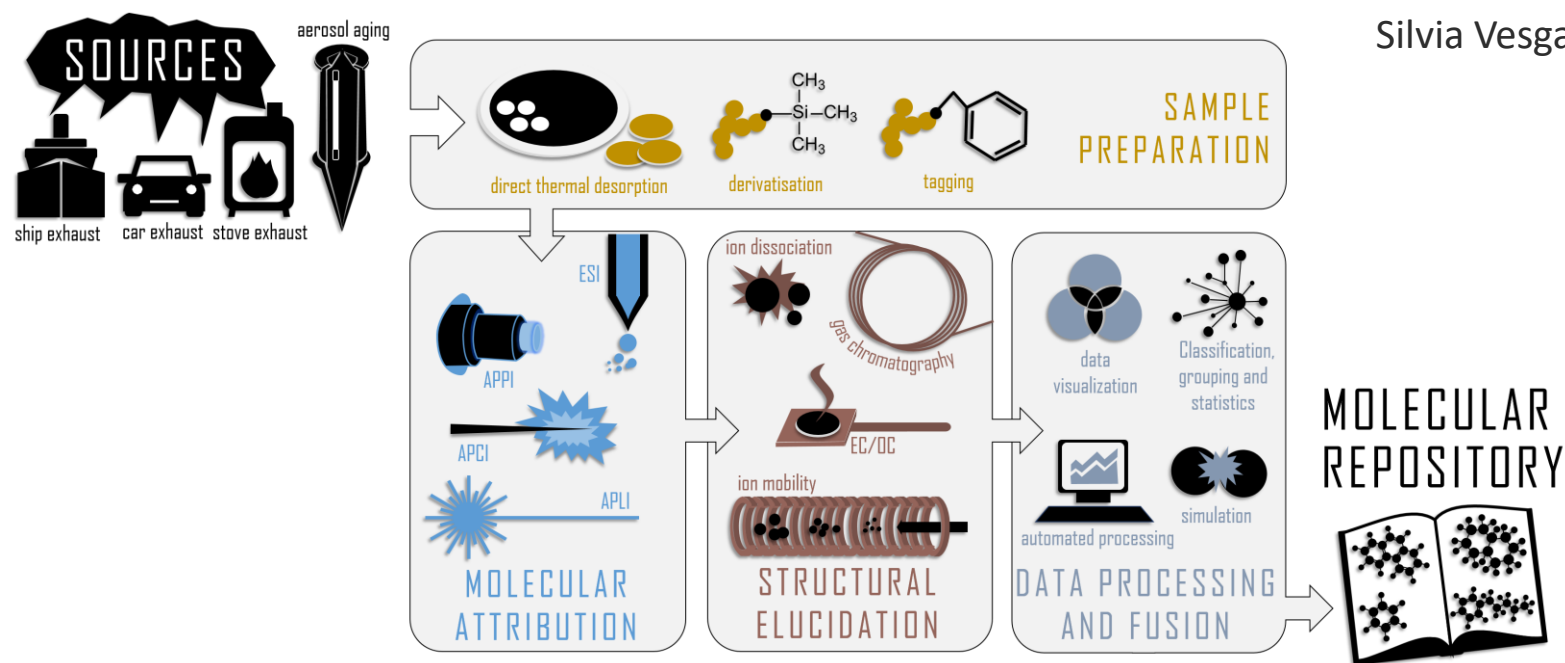




New Projects Resulted from the Network with New Faces:

## DFG-ANR 3-year research project TIMSAC

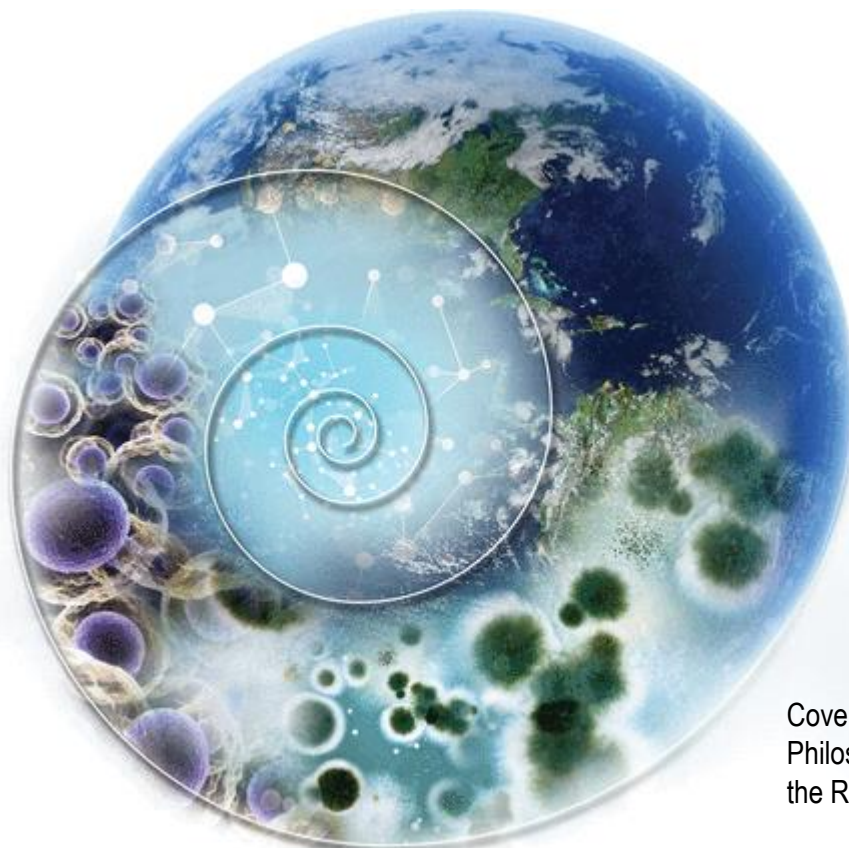
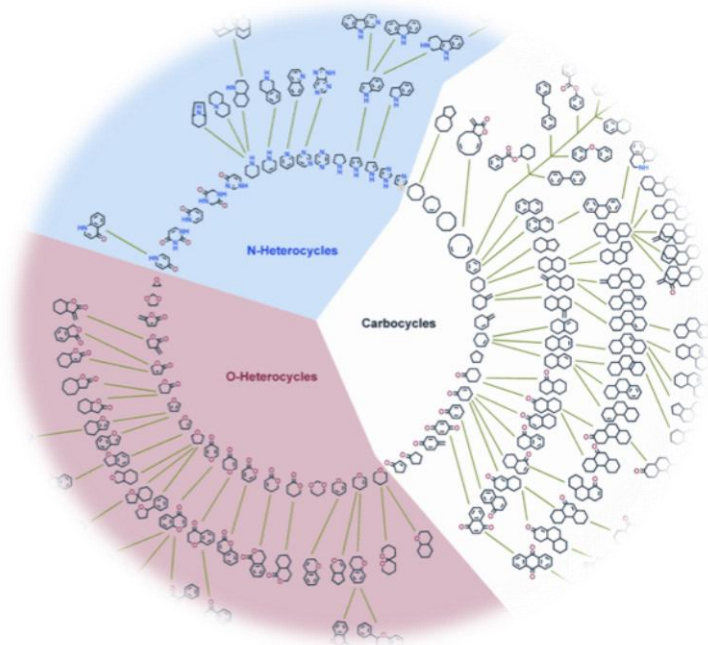
**Thermal analysis and ion mobility coupled to high-resolution mass spectrometry for organic aerosol characterization**



Maxime Sueur & Silvia Vesga-Martinez



## Complexity of organic mixtures – The chemical space



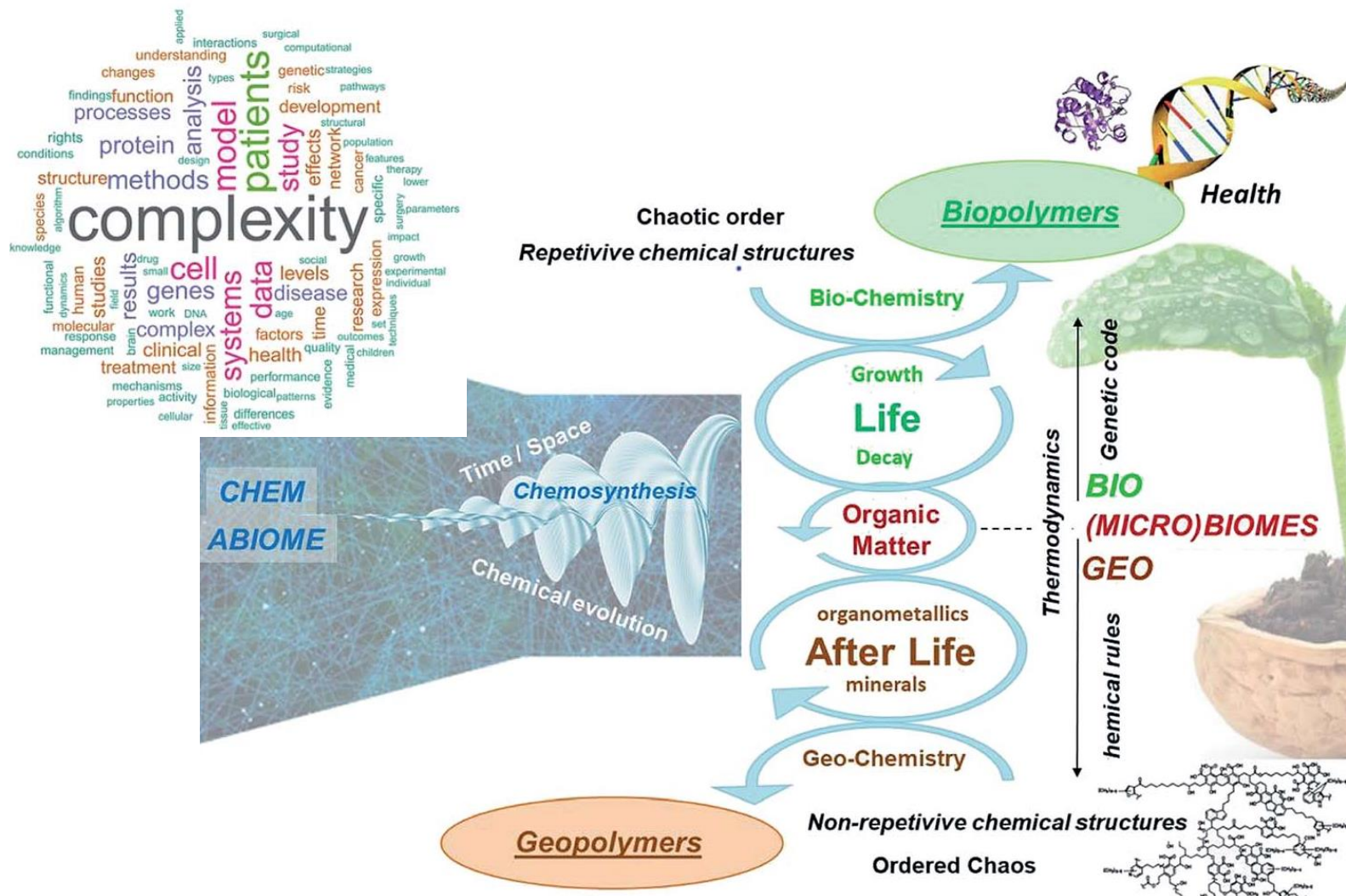
Cover of Vol. 375, 2109,  
Philosophical Transactions of  
the Royal Society A, Dec. 2017

Estimated number of organic molecules with MW < 500:  **$10^{62}$**

Bohacek, R. S., McMartin, C., and Guida, W. C. (1996), Med. Res. Rev. 16, 3–50

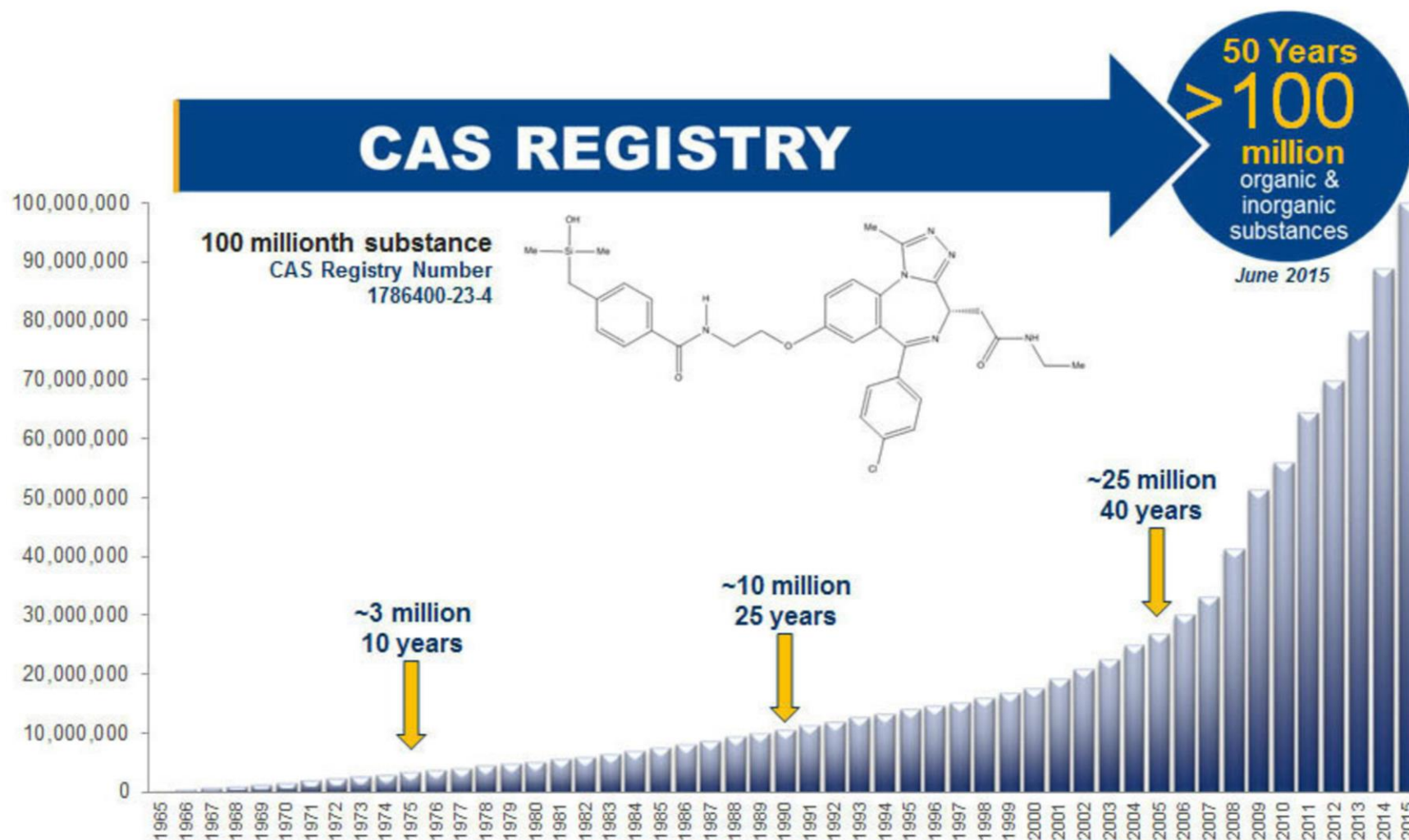


## Complex Mixtures – Everywhere in Science

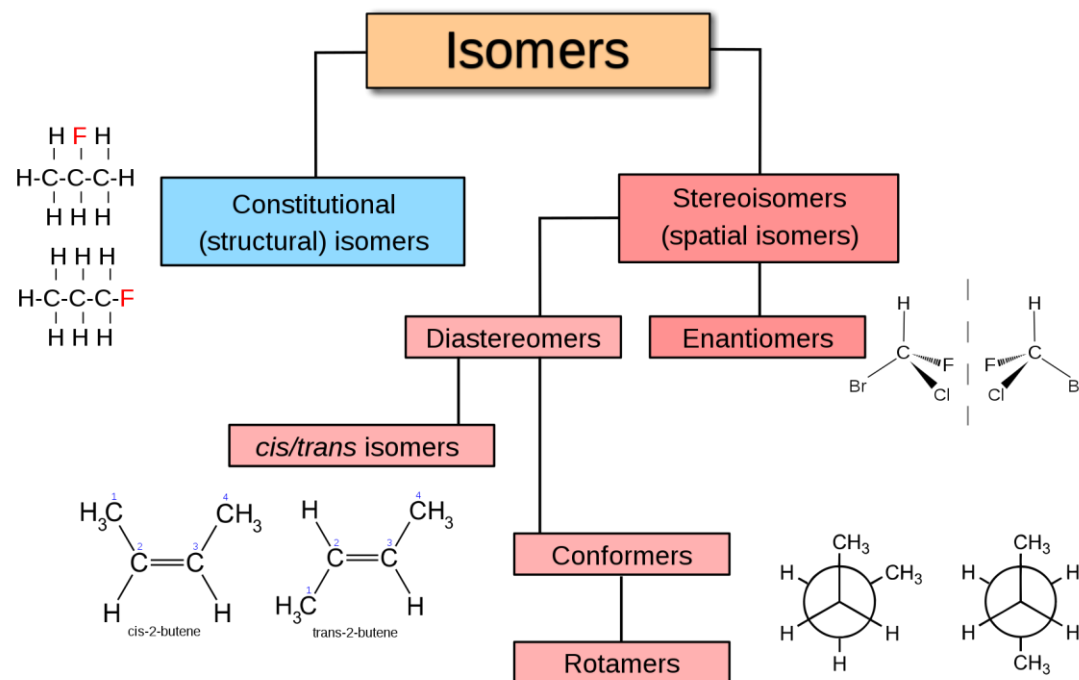


Schmitt-Kopplin et al., *Faraday Discussions*, **2019**, 218, 9-28



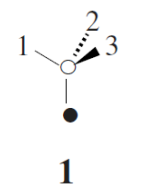


**Other useful general databases:** NIST Webbook (US), ChemSpider (RCS), Pubchem (NIH), HMDB, ...

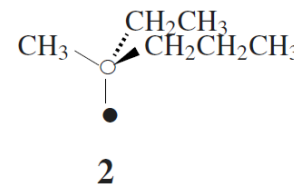


## Isomers in complex mixtures to be tackled by:

- fragmentation, MS/MS
- pre-ionization separation, e.g., LC, GC
- post-ionization separation, e.g., IMS



$\text{C}_{3v}$ -skeleton  
( $\text{C}_{3v}/\text{C}_s$ )

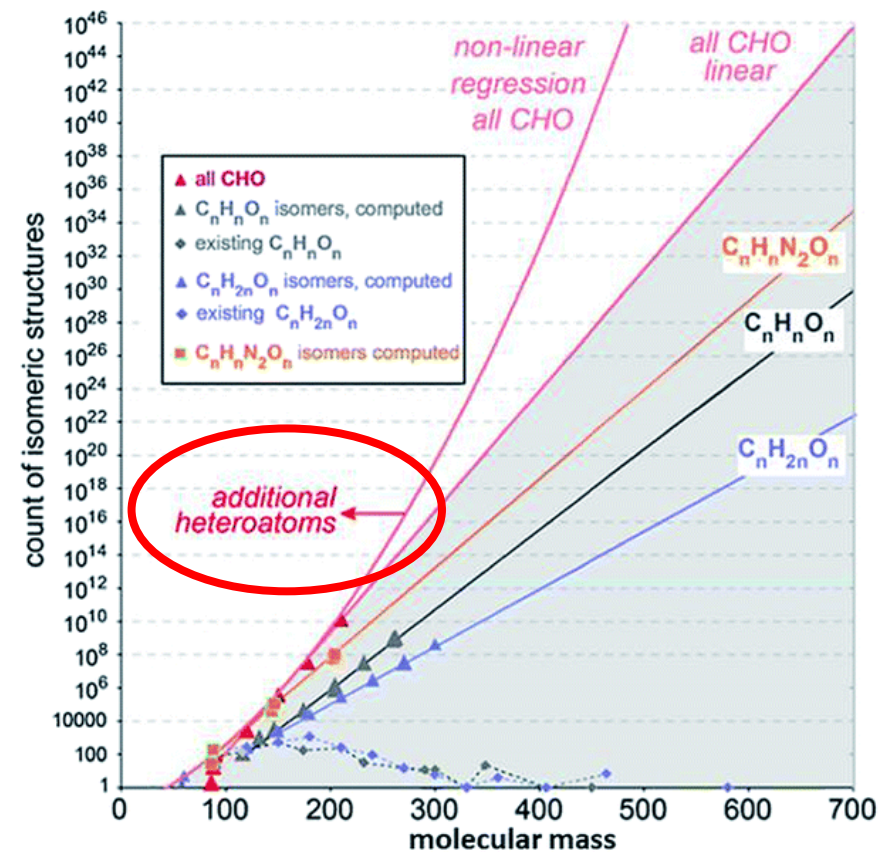
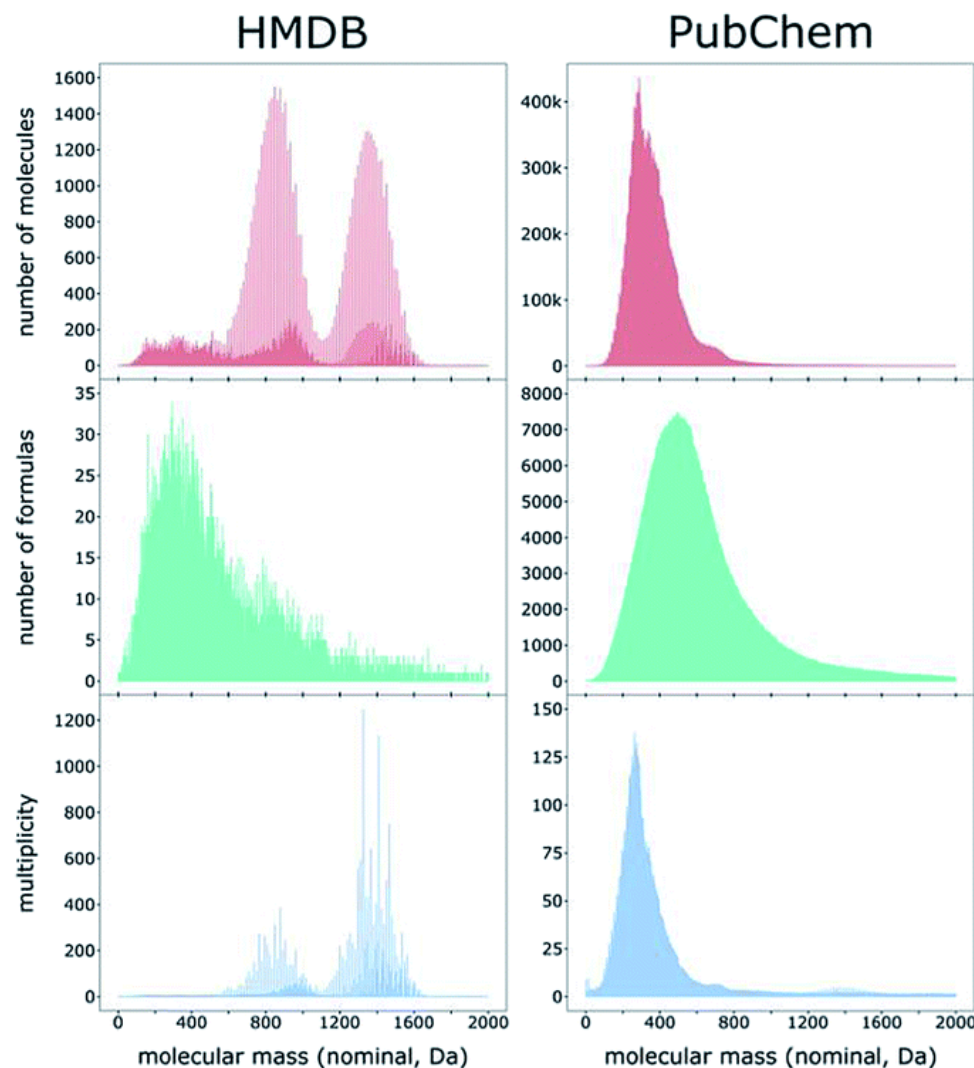


alkyl ligand  
planted 3D-tree

Table 1: Numbers of Centroidal Alkanes as Stereoisomers

$k$	$\hat{B}_k$ (Total)	$\hat{A}_k$ (Achiral)	$\hat{C}_k$ (Chiral)
1	1	1	0
2	0	0	0
3	1	1	0
4	3	3	0
5	1	1	0
6	2	2	0
7	9	7	2
8	8	7	1
9	38	21	17
10	46	22	24
11	203	61	142
12	283	72	211
13	1209	186	1113
14	2004	220	1784
15	9347	507	8780
16	15758	717	15041
17	72505	1755	70750
18	129281	2209	127072
19	589612	5454	584158
20	1098656	7149	1091507
21	4954686	17070	4937616
22	9576645	22476	9554169
23	42671509	53628	42617881
24	84998202	72656	84925546
25	374749447	169175	374580272
26	765965475	229676	765735799
27	3344714436	535267	3344179169
28	6992429665	743026	6991686639
29	30264120901	1698322	30262422579
30	64538102227	2361476	64535740751
31	277096805630	5400908	277091404722
32	601441729659	7642893	601434086766
33	2563418291362	17211368	2563401079994
34	5652900424627	24394779	5652876029848
35	23931052067297	54947147	2393097120150
36	53534003586744	79009726	53534824577018
37	225226025743122	175702378	22522580040744
38	510446528635659	252964410	510446275671249
39	2135109239262173	562645937	2135108676616236
40	4896889667780240	819922295	489688847857945
41	2037287658025143	180408396	20372874770166747
42	47238614063478058	263182074	47238611431657314
43	195544793394384827	5791497722	195544787602887105
44	457999560484205773	8536377160	457999551947828613
45	1886989279103128211	18611821161	1886989260491307050
46	4461040396950967790	27458156708	4461040369492811082
47	18298681742426380229	59870273288	18298681682556106941
48	43636081544149901047	89120741244	43636081455029159803
49	178246302614039769705	192762694240	178246302421277075465
50	428497965173462968567	287167536146	428497964886295432421
51	1734575977870305954708	621145058010	173457597249160896698
52	4222957579213874326998	93263649624	4222957578281237677374
53	17118606500538110493165	2003060193783	17118606498535050299382
54	41757573020631974557921	300964412440	41757573017622330145481
55	168676827177458246245600	6464001746606	16867682177099424498994
56	414192406700454464211608	9780036558542	414192406690674607653066
57	1667507044106396700614662	20873421744449	1667507044085523278870213
58	4120269449060079231919190	31600908341722	4120269449028478323577468
59	16535111535321800418856805	67445191538640	16535111535254355227318165
60	4109822466126905529132145	102743321459580	4109822466024162207672565
61	164431691004690928193890810	218049093481679	164431691004472878290416331
62	410979018222119235952994626	332350277618212	410979018221786885675376414
63	1639544206288762558253253718	705330165952872	1639544206288057228087300846
64	411952978825653749220428115	1081087049647181	4119529788255452662170780934
65	1638881993489362771459167067	2282686396696017	16388819934891345085062471050
66	4138522578224983679926208327	3500478814410329	413852257822466336320451797998
67	164207682036849477130738064324	7391016289967130	164207682036842086114484097194
68	416635607643584641443966956505	11391634034768010	41663560764357324989932188495
69	1648923729893987703415359750119	23941657967808209	1648923729893963761757391941910
70	420270863802068286397777378556	36917161577580590	4202708638020645946818199779966
71	1659258987561223600901193290288	77586381466034947	16592589875613158505431727257741
72	42473372537692324926023951462175	120189306991204216	42473372537692204736714260167959
73	167295424464204241785279087528744	251528935349306793	167295424464203990256343738221951
74	430006772381311411463428100669081	389800782399677001	430006772381311021662645700992080
75	1689911835064564750230170718349855	81574114033806827	168991183506456393448903380281628
76	4360786726830840048241690321351540	1269537259911351156	436078672683083877870376413000384
77	1710060975107414421714560807768366	264648989629959150	171006097510741570684664508176881
78	44294354185222186820769124153521749	4120240983149128308	44294354185222182700528141004393441
79	17335468778163767515568474781956550	8588824555686539622	17335468778163758926743919095416928
80	450600427378156845559982800702584531	13423903573385498107	45060042737815683213607922717086424
81	1759767313496477955950136185436237113	27882748457230290862	1759767313496477928067387728205946251

Fujita, *MATCH Commun. Math. Comput. Chem.* **57** (2007) 299-340



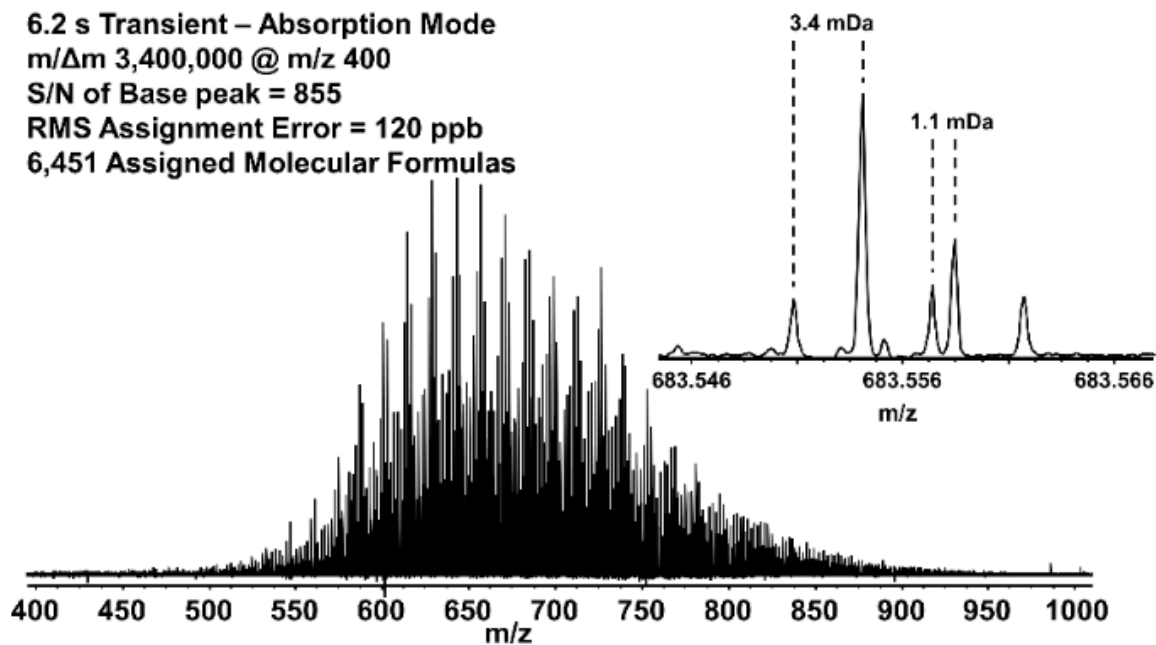


What is complexity really? How do we define a complex mixture?

a single doublet of not resolved (mass spectrometrically or analytically in general) compounds

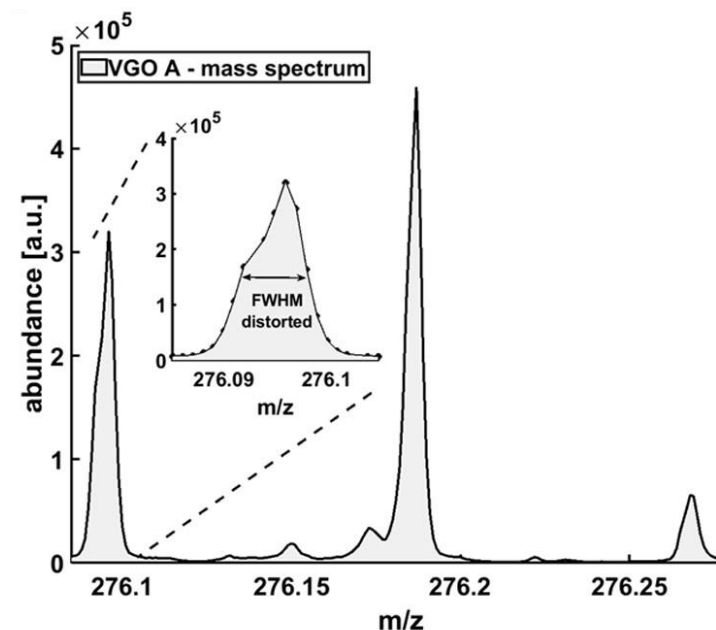
## Single Scan from LC Run

6.2 s Transient – Absorption Mode  
m/Δm 3,400,000 @ m/z 400  
S/N of Base peak = 855  
RMS Assignment Error = 120 ppb  
6,451 Assigned Molecular Formulas



unpublished data – 21T FT-ICR MS platform NHMFL

or



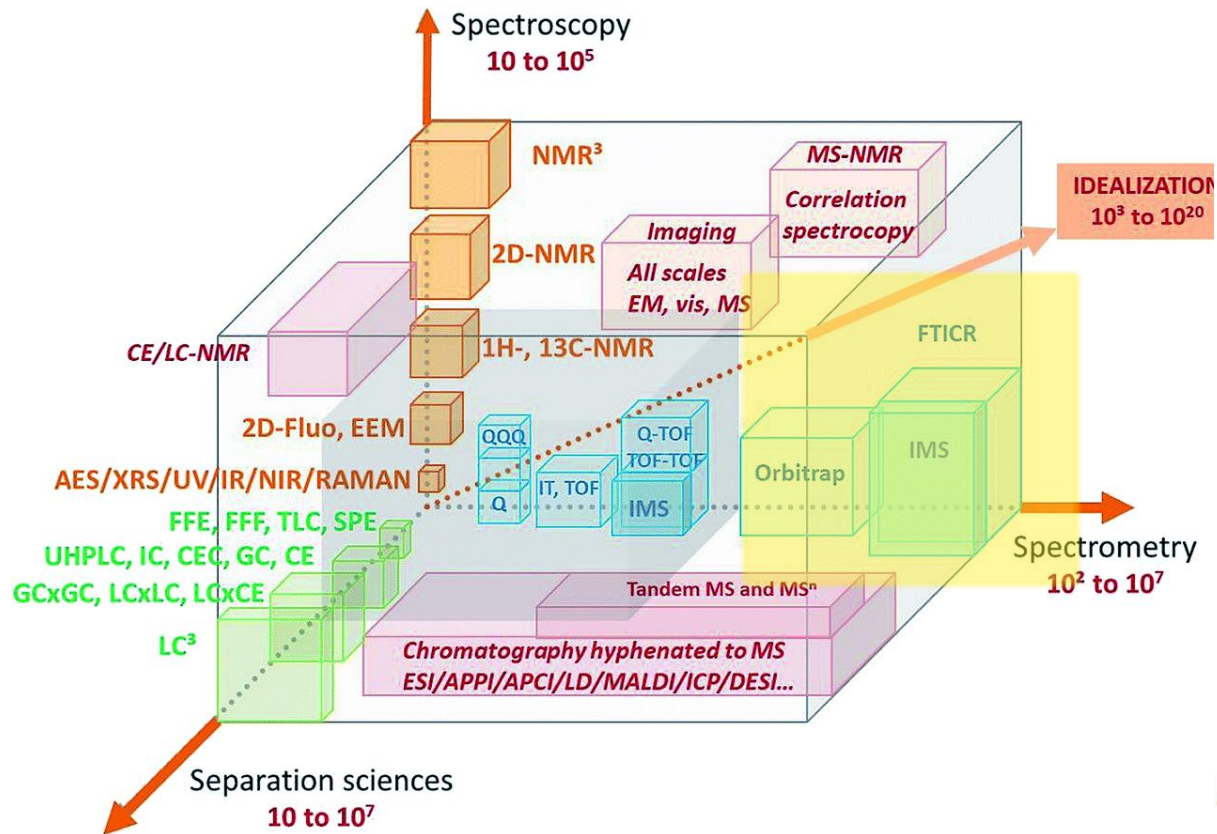
Rüger, et al., *Analytical Chemistry* 93 (14), S. 5872–5881.

Challenging complex mixture as somewhat philosophical question (courtesy to Prof. Jan Andersson, Univ. Munster)

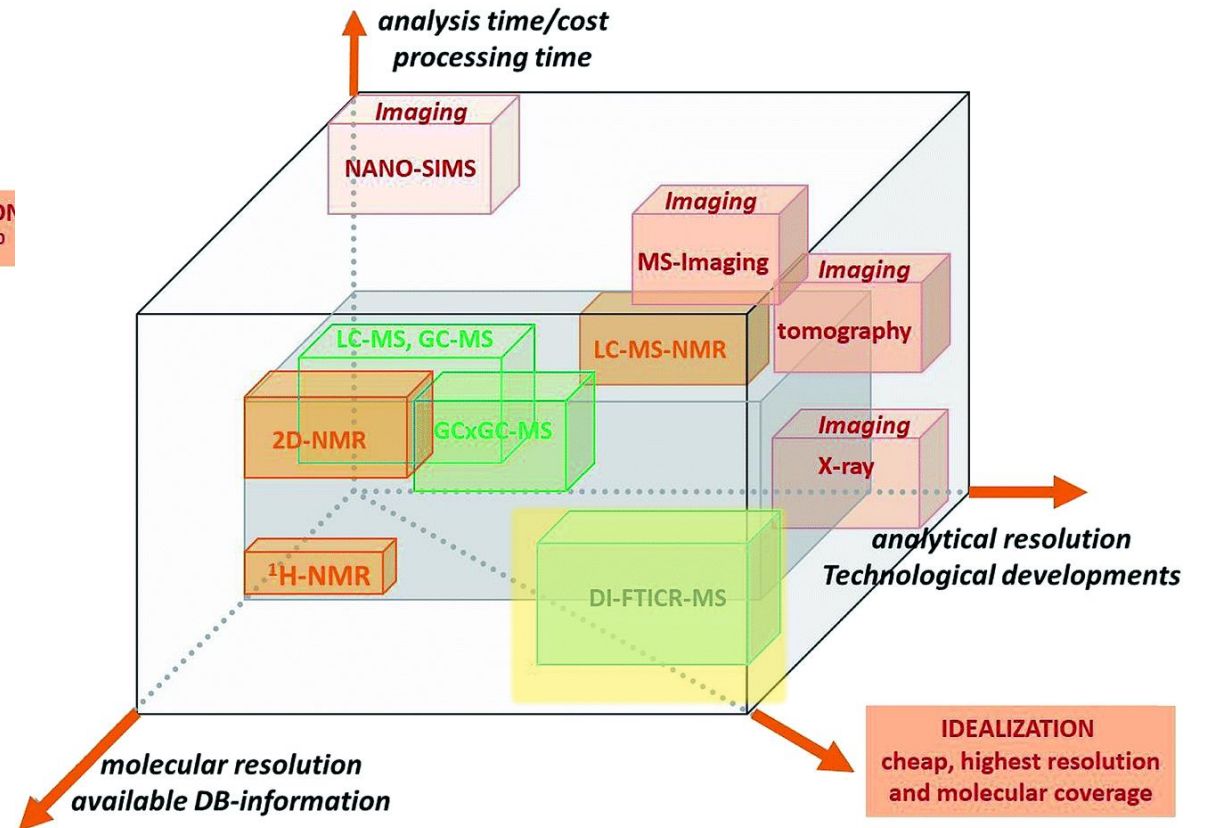
*“I have a friend who’s an artist and has sometimes taken a view which I don’t agree with very well. He’ll hold up a flower and say “look how beautiful it is,” and I’ll agree. Then he says “I as an artist can see how beautiful this is but you as a scientist take this all apart and it becomes a dull thing,” and I think that he’s kind of nutty. First of all, the beauty that he sees is available to other people and to me too, I believe... I can appreciate the beauty of a flower. At the same time, I see much more about the flower than he sees. I could imagine the cells in there, the complicated actions inside, which also have a beauty. I mean it’s not just beauty at this dimension, at one centimeter; there’s also beauty at smaller dimensions, the inner structure, also the processes. The fact that the colors in the flower evolved in order to attract insects to pollinate it is interesting; it means that insects can see the color. It adds a question: does this aesthetic sense also exist in the lower forms? Why is it aesthetic? All kinds of interesting questions which the science knowledge only adds to the excitement, the mystery and the awe of a flower. It only adds. I don’t understand how it subtracts.”*

*Richard Feynman, Nobel-laureate in physics (from BBC Interview for Horizon ‘The Pleasure of Finding Things Out’ (<https://www.bbc.co.uk/sn/tvradio/programmes/horizon/broadband/archive/feynman/>) animated by Fraser Davidson/<https://vimeo.com/55874553>).*

## Separation capabilities according to the three main domains:



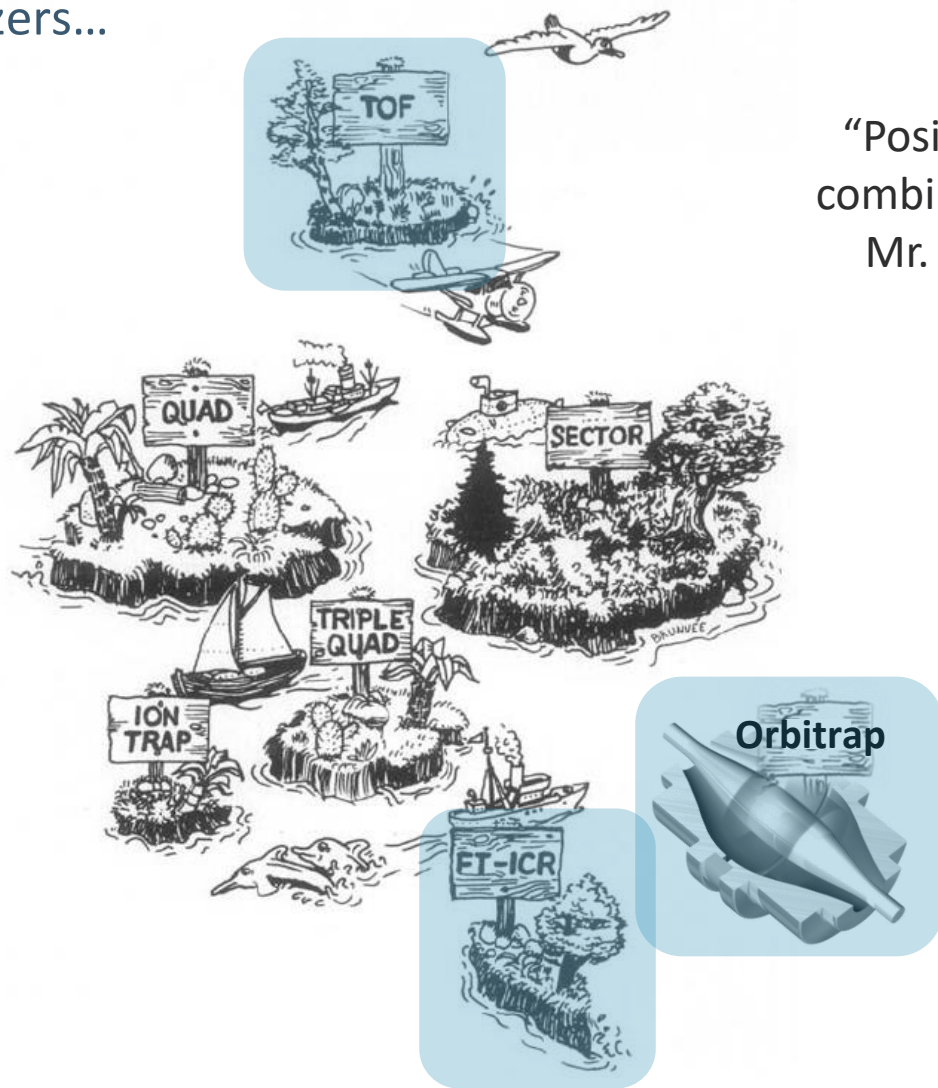
## Pragmatic combination of coverage, price/time, etc.:





# Which Mass Spectrometric Platform?

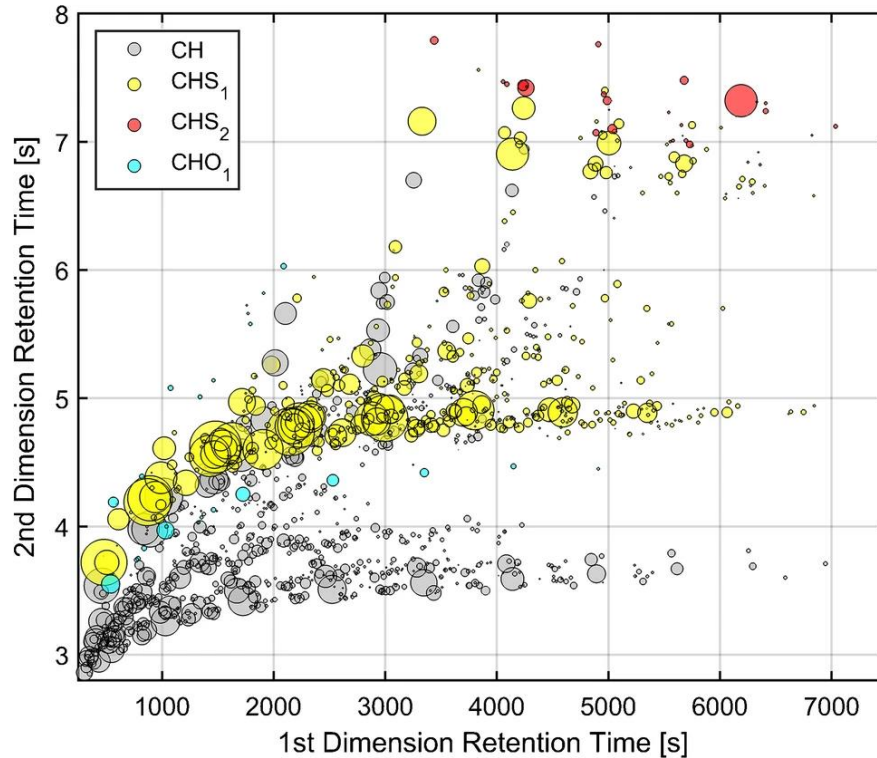
The realm of mass analyzers...



“Positive and negative features are always combined in one instrument as Dr. Jekyll and Mr. Hyde are combined in one person.”

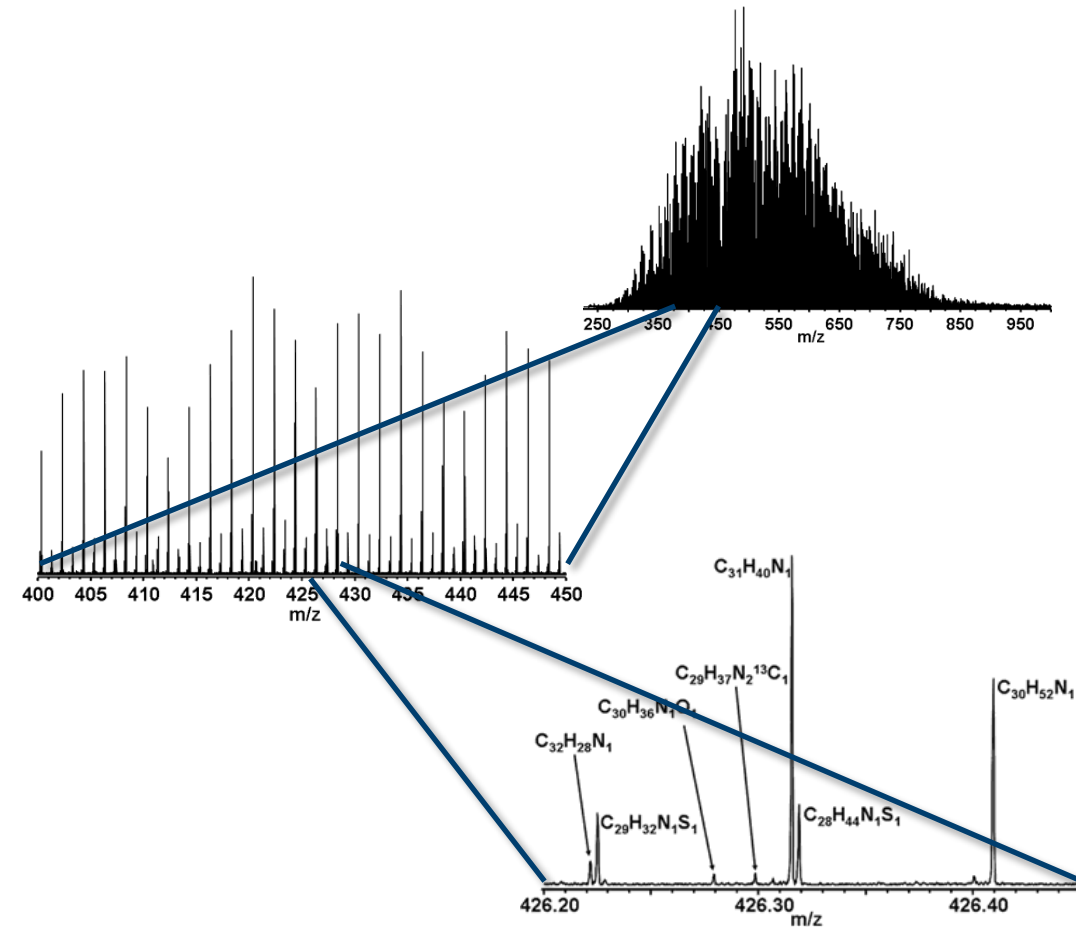
adapted based on:  
Curt Brunnée, The Ideal Mass  
Analyzer Fact or Fiction, *Int. J.  
Mass Spectrom and Ion  
Processes*, 76 (1987), 125-237

## Comprehensive GCxGC



Schwalb et al. *Analytical and Bioanalytical Chemistry*. DOI: 10.1007/s00216-022-04393-w.

**Peak capacity:** Maximum theoretical number of components that can be separated. → Reality check: peak coverage



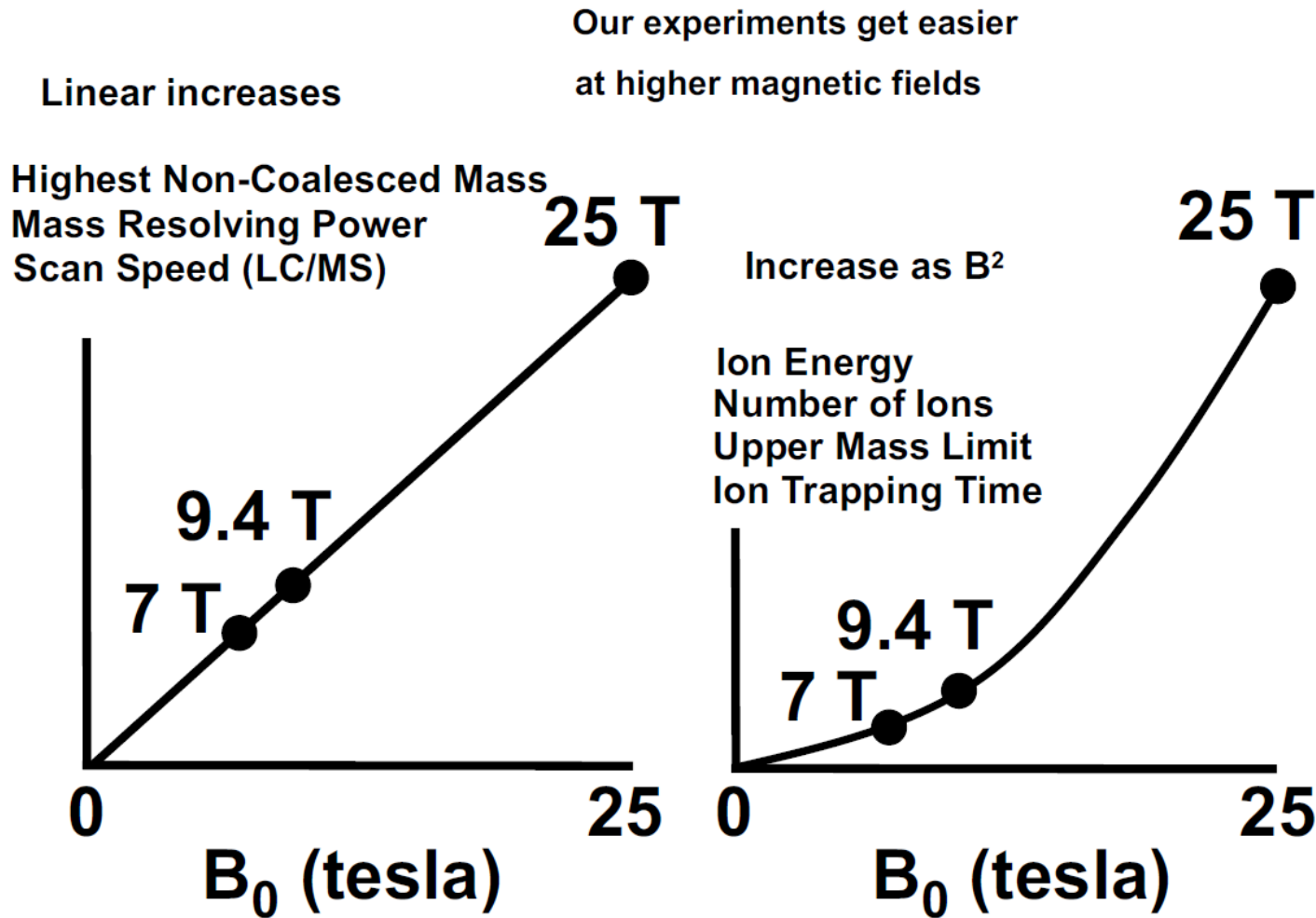
Even highly complex mixtures are not covering the entire  $m/z$ -domain giving a limited peak coverage and utilization of the FTMS peak capacity (naturally by the mass defect)

## “Can we cope with the complexity by FT-ICR MS?” – Wishful thinking

- analytical challenges: classical direct infusion versus hyphenated solution (some covered here)
- expanding the accessible chemical space via ionization techniques (see talk from Janne Jänis and Carlos Afonso)
- fragmentation techniques and novel unique 2D solutions (see talk from Pete O’Connor and Maria van Agthoven)
- open-code/access software solutions for data processing, visualization and comparison (see talk from Maxim Sueur)

→ **Limitation for this lecture:** Classical ‘omics sciences (biomedical/biochemistry/biopolymers via proteomics, metabolomics, lipidomics, etc.) excluded and not covered in this talk!



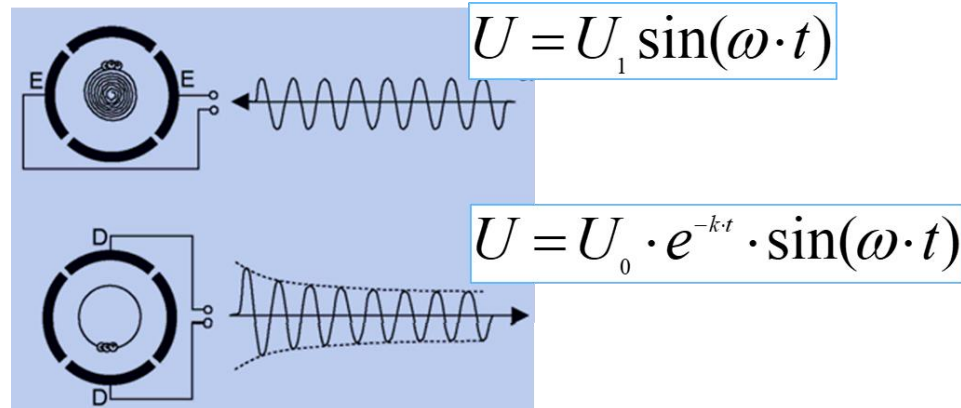
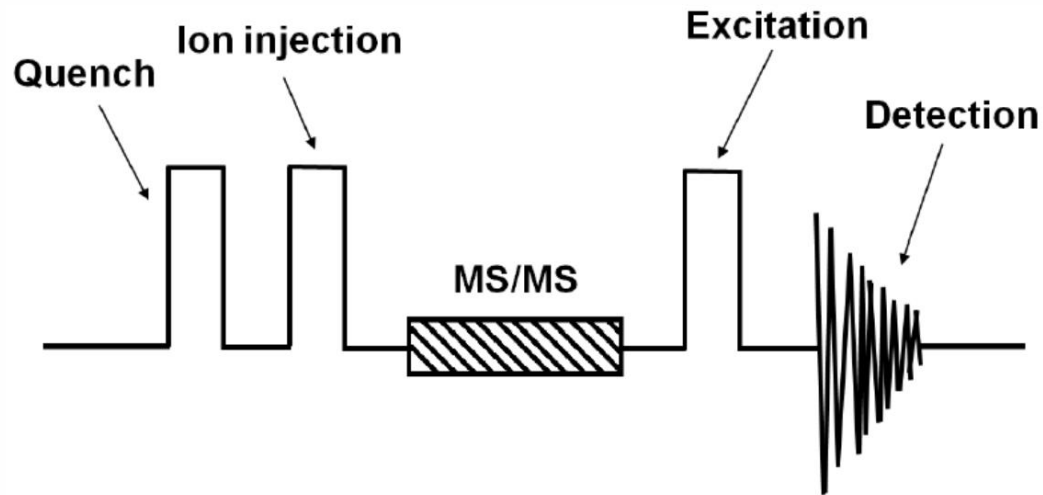


Marshall et al., *Mass Spectrometry Reviews* (1998) 17, 1-35



21T FT-ICR MS platform at the NIMFL  
(2<sup>nd</sup> comparable system at PNNL)

# From Transient to Mass Spectrum



Cyclotron frequency ( $\sim 1$  Mhz)

$$\omega_c = \frac{q}{m} \cdot B$$

simplest case, only the value of the magnetic field is considered

Trapping frequency ( $\sim 1$  khz)

$$\omega_T = \frac{2}{a} \cdot \sqrt{\alpha \cdot U_T \cdot \frac{q}{m}}$$

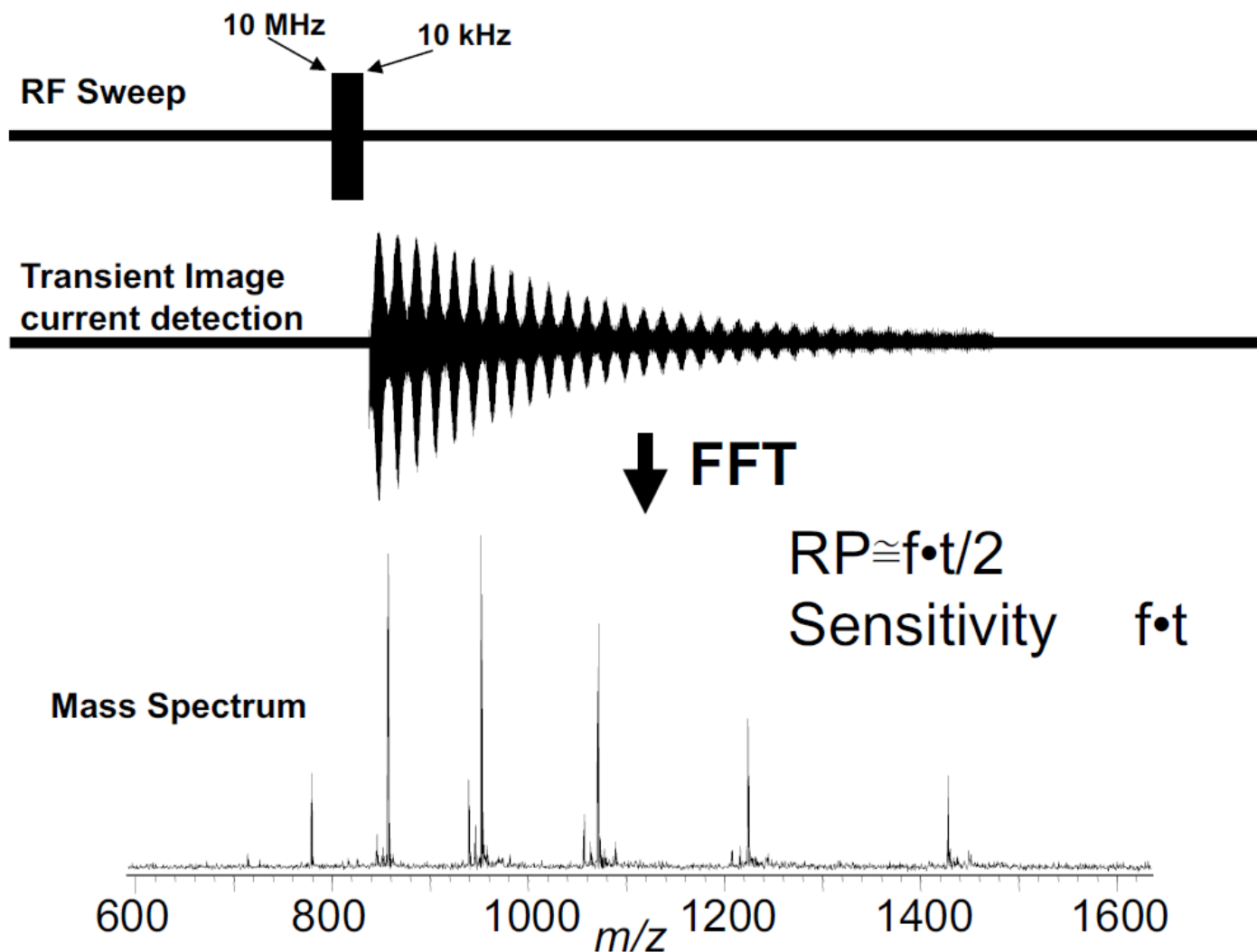
A more accurate equation takes into account the reduction of  $\omega_c$  relative to the application of the trapping electric field

Effective Frequency

$$\omega_{eff} \cong \omega_c - \omega_m \quad \omega_m \cong \frac{\omega_T^2}{2 \cdot \omega_c} = \frac{2 \cdot \alpha \cdot U_T}{a^2 \cdot B}$$

so called “magnetron frequency” ( $\sim 10$  Hz)

# From Transient to Mass Spectrum



## What to do with the time domain signal:

- Zero-filling
- Windowing (Apodization)
- Fourier-transformation
- Phasing
- Calibration

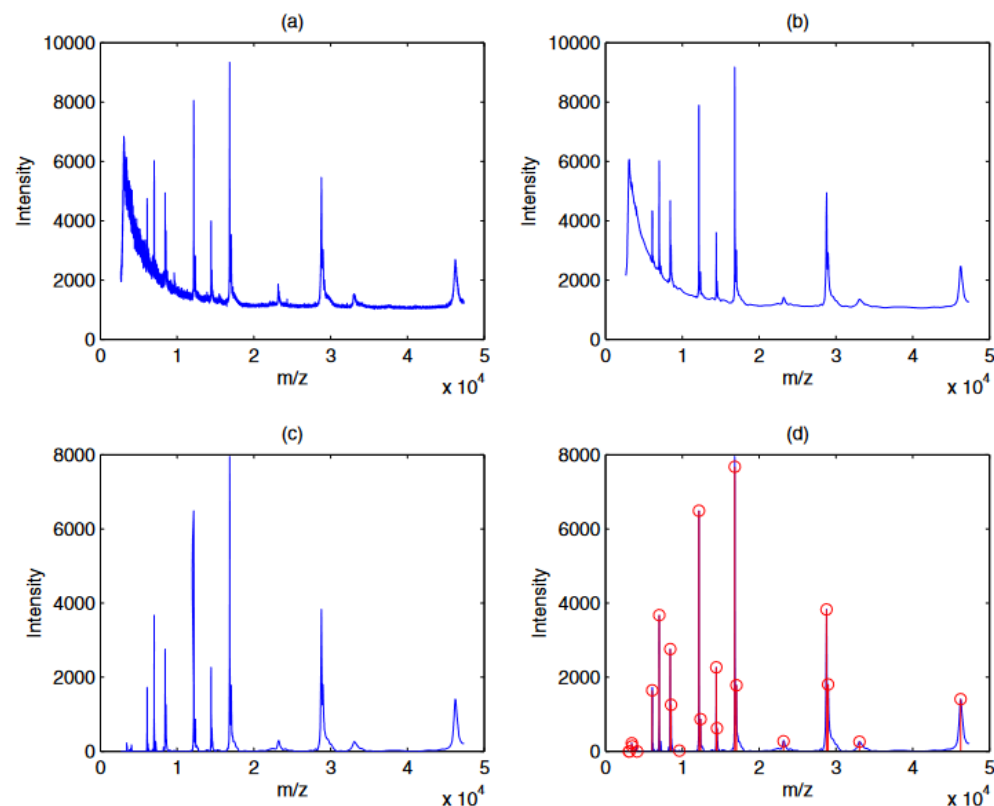
## Software Solutions:

- vendor-based (Bruker/Thermo)
- custom free solutions, *e.g.*, EU\_FT-ICR\_MS development, phasing (Pete O'Connor) or Predator
- external companies, *e.g.*, Spectroswiss

Slide courtesy mostly to Pete O'Connor



## Peak Picking – From profile data to line centroided data



**Figure 2**  
An example of the peak detection process. (a) A raw spectrum, (b) the spectrum after smoothing, (c) the spectrum after smoothing and baseline correction and (d) final peak detection results with peaks marked as circles.

Yang et al., *BMC Bioinformatics* 2009, 10:4



© American Society for Mass Spectrometry, 2016



J. Am. Soc. Mass Spectrom. (2017) 28:253–262  
DOI: 10.1007/s13361-016-1549-z

### RESEARCH ARTICLE

## Autopicker - a Robust and Reliable Peak Detection Algorithm for Mass Spectrometry

David P. A. Kilgour,<sup>1</sup> Sam Hughes,<sup>2</sup> Samantha L. Kilgour,<sup>3</sup> C. Logan Mackay,<sup>2</sup> Magnus Palmblad,<sup>4</sup> Bao Quoc Tran,<sup>5</sup> Young Ah Goo,<sup>5</sup> Robert K. Ernst,<sup>3</sup> David J. Clarke,<sup>2</sup> David R. Goodlett<sup>5</sup>

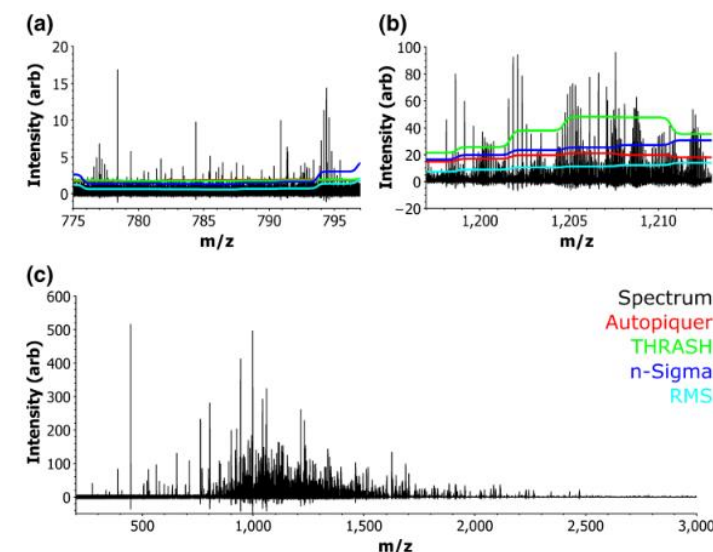
<sup>1</sup>Department of Chemistry and Forensics, Nottingham Trent University, Nottingham, NG11 8NS, UK

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<sup>5</sup>Department of Pharmaceutical Sciences, School of Pharmacy, University of Maryland, Baltimore, MD 21201, USA



**Figure 4.** Showing peak detection thresholds being set for an absorption mode FT-ICR MS spectrum of the electron capture dissociation product ions of denatured horse heart myoglobin. The complete spectrum is shown in panel (c), bottom row. Panels (a) and (b), top row, show the peak detection thresholds generated by different algorithms in two regions of interest – one with low peak density and the other with high peak density. For all methods, the window width is 3 Da. The spectrum is shown in black and the thresholds calculated by the Autopicker, THRASH, *n*-Sigma, and RMS methods are shown in red, green, blue, and cyan, respectively

## Peak Picking – “Where to cut the noise?”

analytical  
chemistry

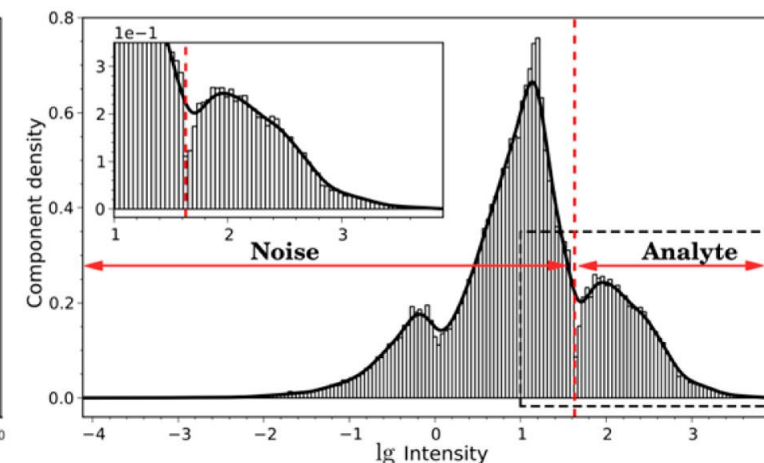
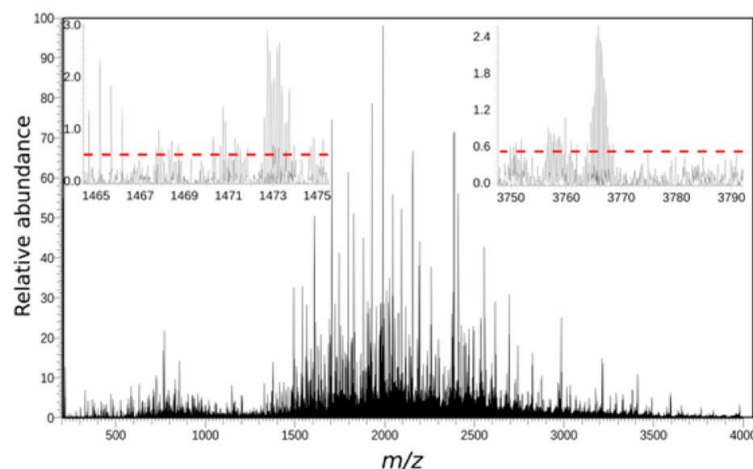
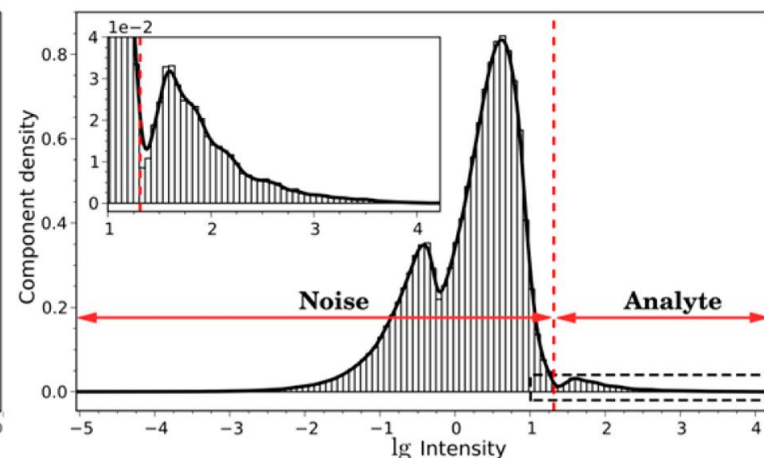
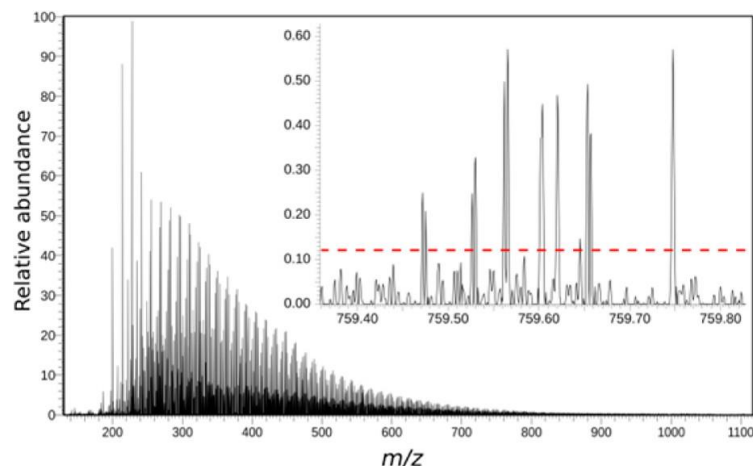
Article  
pubs.acs.org/ac

### Distinguishing Analyte from Noise Components in Mass Spectra of Complex Samples: Where to Cut the Noise?

Konstantin O. Zhurov, Anton N. Kozhinov, Luca Fornelli, and Yury O. Tsybin\*

Biomolecular Mass Spectrometry Laboratory, Ecole Polytechnique Fédérale de Lausanne, 1015 Lausanne, Switzerland

Analysis of full-profile mass spectra of South American crude oil fraction (top) and Humira IgG (bottom). Left panels: absorption-mode-type mass spectra acquired on Orbitrap Elite FTMS. Right panels: distributions of logarithmic intensity. Red lines indicate the noise threshold values.

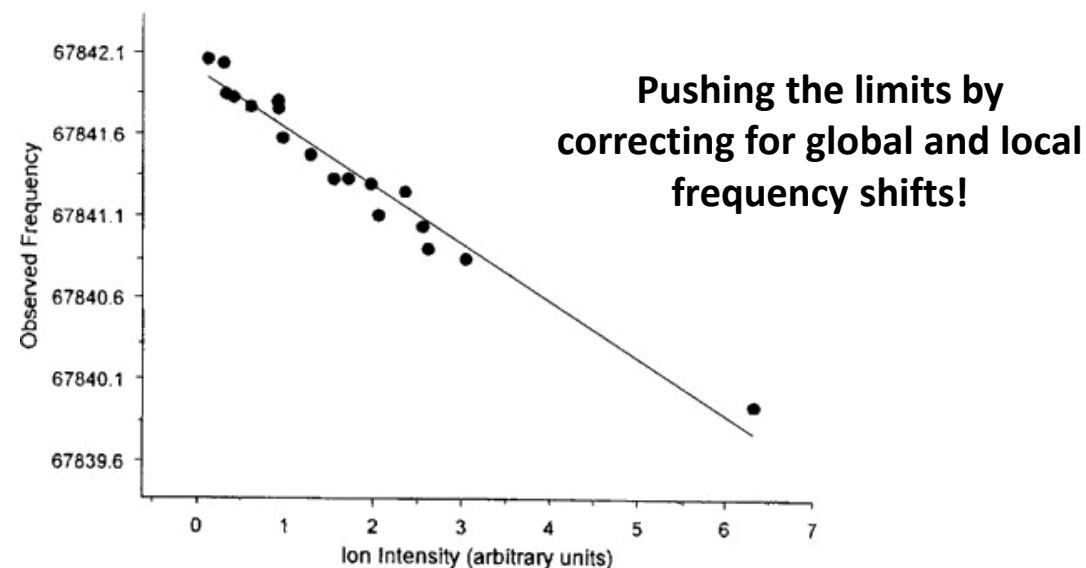




## Mass Spectral Calibration – the ion trap behaviour and physics of FTMS

**TABLE 1.** Proposed calibration procedures

$f = \frac{a}{m}$	basic law of ions in a B field
$f^2 = \frac{a}{m^2} + \frac{b}{m}$	(Beauchamp-Armstrong et al., 1969)
$f^2 = \frac{a}{m^2} + \frac{b}{m} + c$	(Ledford et al., 1980)
$f_{\text{sideband}} = \frac{a}{m}$	(Allemann et al., 1981)
$f = \frac{a}{m} + c$	(Francl et al., 1983)
$\left(\frac{M}{Z}\right) = \frac{a}{f_{\text{obsd}}} + \frac{b}{f_{\text{obsd}}^2}$	(Ledford et al., 1984b)
$f_{\text{estimated}} = f_{\text{measured}} + c(I_{\text{calibrant}} - I_{\text{analyte}})$	(Easterling et al., 1999)
$\frac{m}{z} = \frac{A}{f_{\text{estimated}}} + \frac{B}{f_{\text{estimated}}^2} + \frac{C}{f_{\text{estimated}}^3}$	(Bruce et al., 2000)
$M = \left(\frac{kB}{f_n + \Delta f}\right)n - n(M_c)$	(Masselon et al., 2002)
$\left(\frac{M}{Z}\right)_i = \frac{a}{f_{\text{obsd}}} + \frac{b}{f_{\text{obsd}}^2} + \frac{CI_i}{f_{\text{obsd}}^2}$	(Wang et al., 1988)
$\frac{m}{z} = \frac{A}{v} + \frac{B}{v^2} + \frac{C}{v^3} + \frac{BC}{Av^4}$	



Frequency shift as a function of ion population for the monoisotopic peak of bradykinin, showing the change in observed frequency as a function of total ion intensity as ions from a single laser shot are remeasured.

Zhang, L. K.; Rempel, D.; Pramanik, B. N.; Gross, M. L.; Accurate mass measurements by Fourier transform mass spectrometry *Mass Spectrom. Rev.* **2005**, *24*, 286-309.  
T. J. Francl, M. G. Sherman, R. L. Hunter, M. H. Locke, W. D. Bowers, R. T. McIver, *Int. J. Mass Spec.* 1983, *54*, 189-199.  
M. L. Easterling, T. H. Mize, I. J. Amster, *Anal. Chem.* 1999, *71*, 624-632.  
C. Masselon, A. V. Tolmachev, G. A. Anderson, R. Harkewicz, R. D. Smith, *J. Am. Soc. Mass Spectrom.* 2002, *13*, 99-106.

## Mass Spectral Calibration – “Walking” Calibration Equation

analytical chemistry

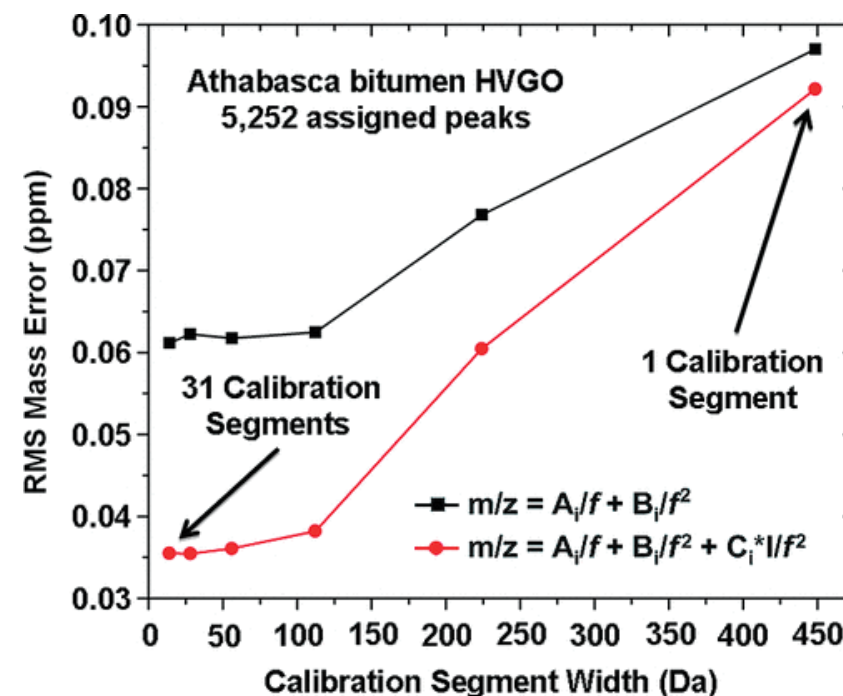
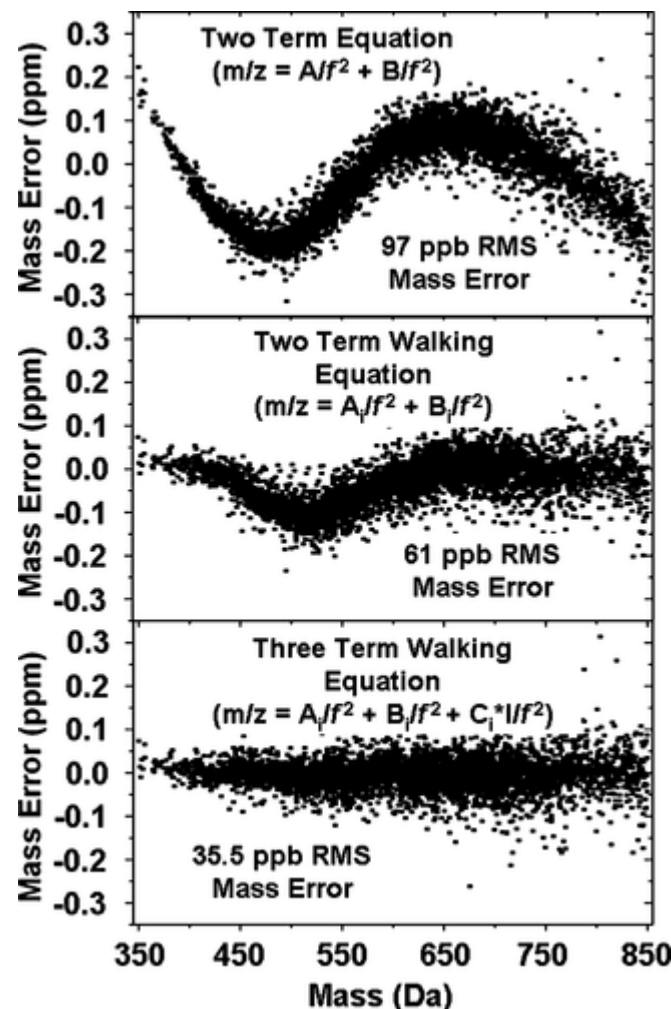
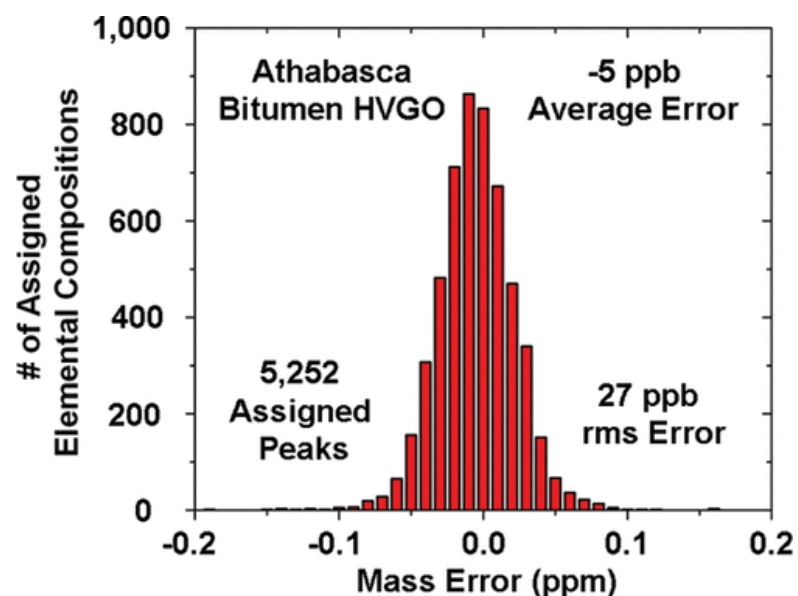
ARTICLE  
pubs.acs.org/ac

Parts-Per-Billion Fourier Transform Ion Cyclotron Resonance Mass Measurement Accuracy with a “Walking” Calibration Equation

Joshua J. Savory,<sup>†</sup> Nathan K. Kaiser,<sup>†</sup> Amy M. McKenna,<sup>†</sup> Feng Xian,<sup>‡</sup> Greg T. Blakney,<sup>†</sup> Ryan P. Rodgers,<sup>†,‡</sup> Christopher L. Hendrickson,<sup>\*,†,‡</sup> and Alan G. Marshall<sup>\*,†,‡</sup>

<sup>†</sup>National High Magnetic Field Laboratory, Florida State University, 1800 East Paul Dirac Drive, Tallahassee, Florida 32310-4005, United States

<sup>‡</sup>Department of Chemistry, Florida State University, 95 Chieftain Way, Tallahassee, Florida 32306, United States



“Addition of an abundance-dependent term to the walking calibration equation and <sup>13</sup>C isotopomers to the calibration series was also advantageous.”



## Mass Spectral Calibration – Iterative Recalibration

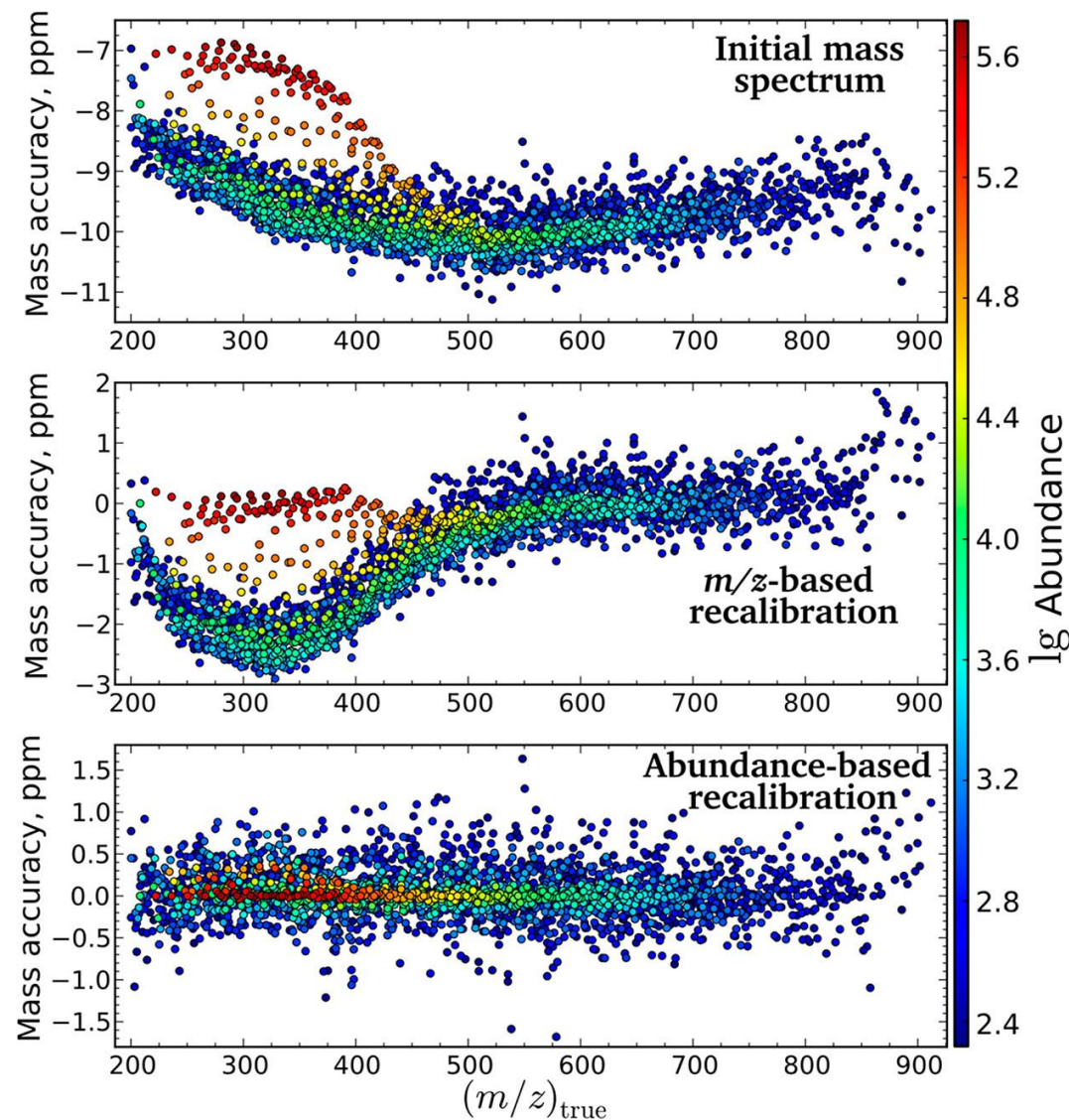
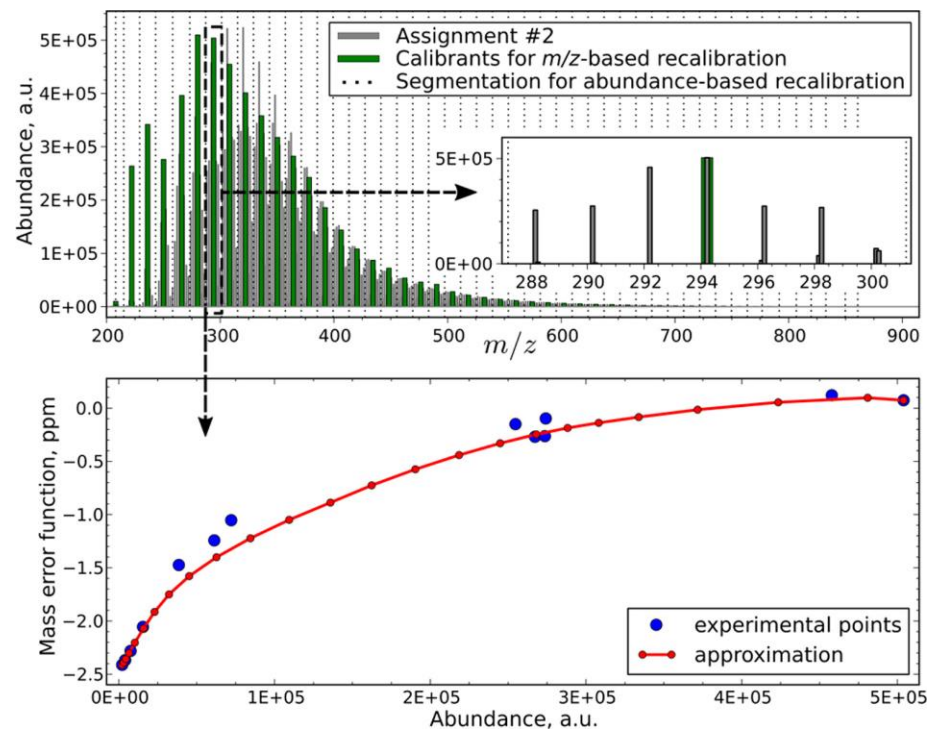
analytical  
chemistry

Article  
pubs.acs.org/ac

### Iterative Method for Mass Spectra Recalibration via Empirical Estimation of the Mass Calibration Function for Fourier Transform Mass Spectrometry-Based Petroleomics

Anton N. Kozhinov, Konstantin O. Zhurov, and Yuri O. Tsybin\*

Biomolecular Mass Spectrometry Laboratory, Ecole Polytechnique Fédérale de Lausanne, 1015 Lausanne, Switzerland



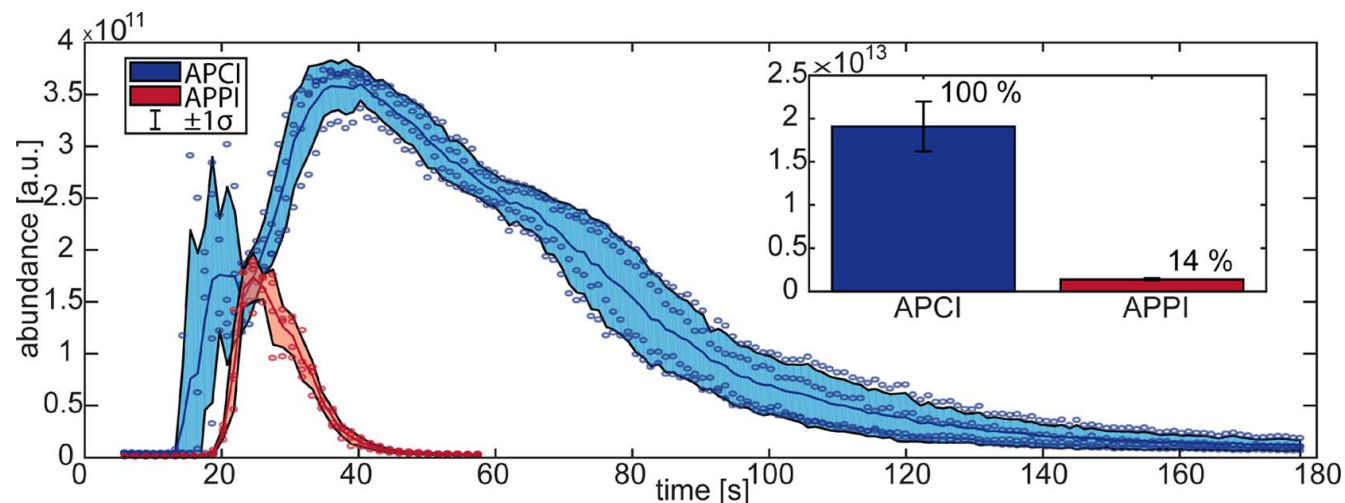
## Hyphenated Solutions – Time-Dependent Data

pubs.acs.org/jasms

Research Article

### Direct Inlet Probe Atmospheric Pressure Photo and Chemical Ionization Coupled to Ultrahigh Resolution Mass Spectrometry for the Description of Lignocellulosic Biomass

Clément Castilla, Christopher P. Rüger, Stéphane Marcotte, Hélène Lavanant,\* and Carlos Afonso



Time-resolved total ion count (TIC) for the DIP experiment conducted with APCI (blue) and APPI (red) on ground beech pellets. The shaded area corresponds to  $\pm 1\sigma$ , and the scattered circles are derived from individual mass spectra. The averaged summed TIC with the respective standard deviation is given in the inset.

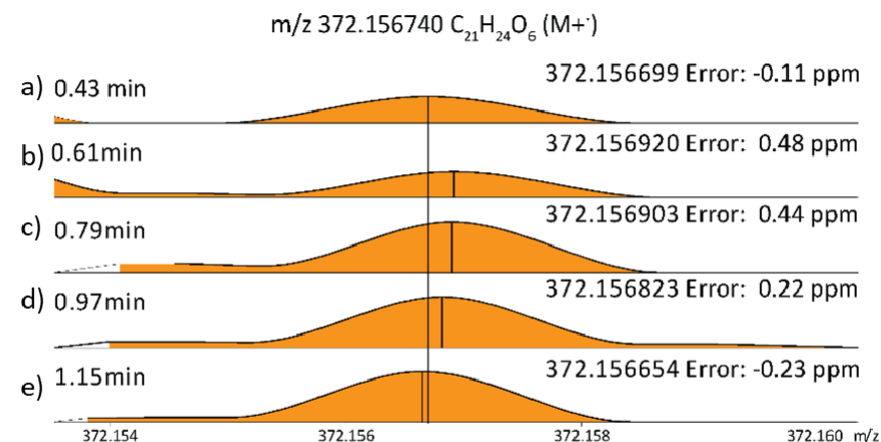


Figure S4. Mass shifts observed during pyrolysis process of beech pellet for  $m/z$  372.15640 ion. With mass shift observed at a) 0.43 min, b) 0.61 min, c) 0.79 min, d) 0.97 min, e) 1.15 min

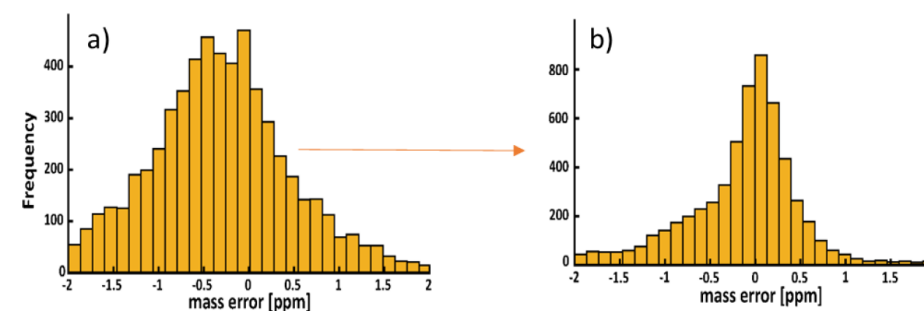


Figure S5. Improvement of attribution error using scan-to-scan recalibration, with a) the mass error before calibration and b) the mass error after scan to scan calibration



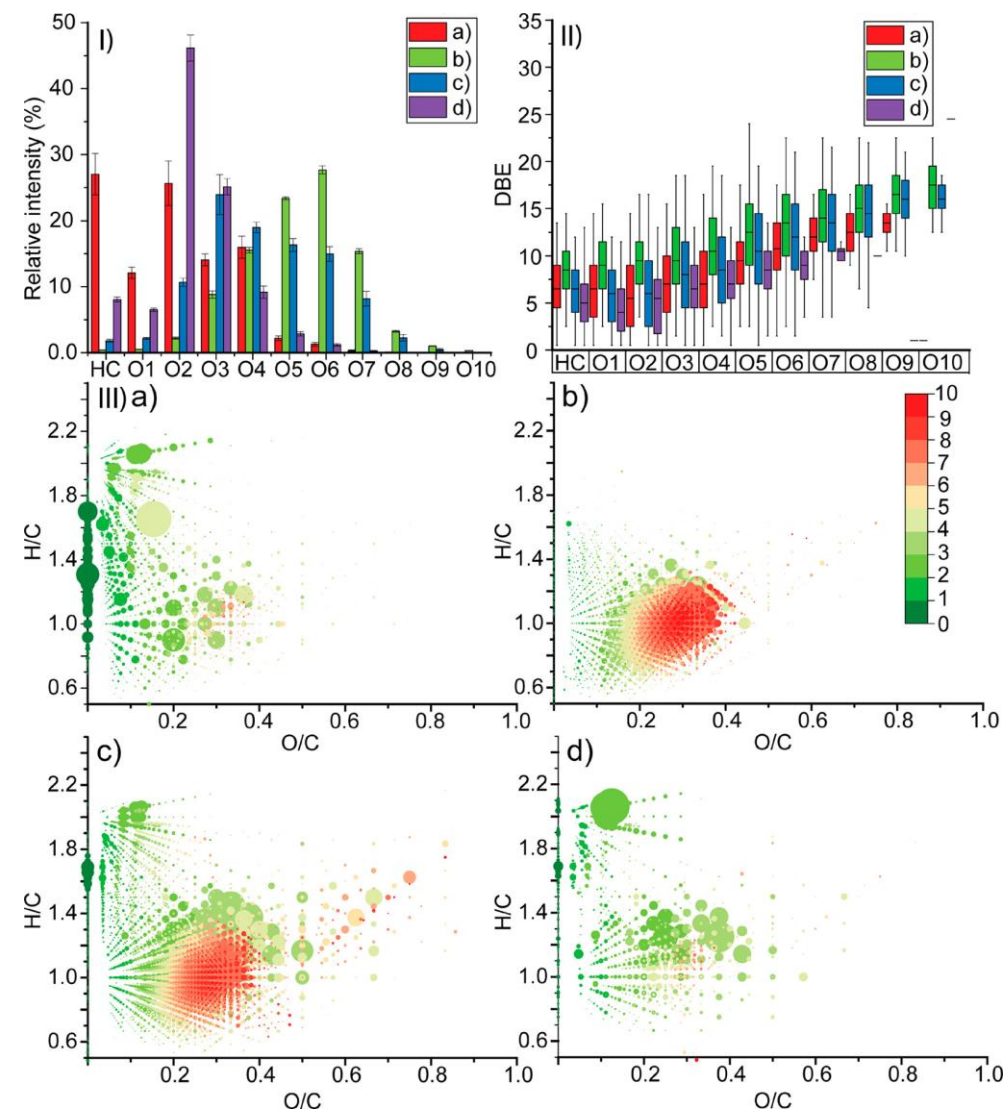
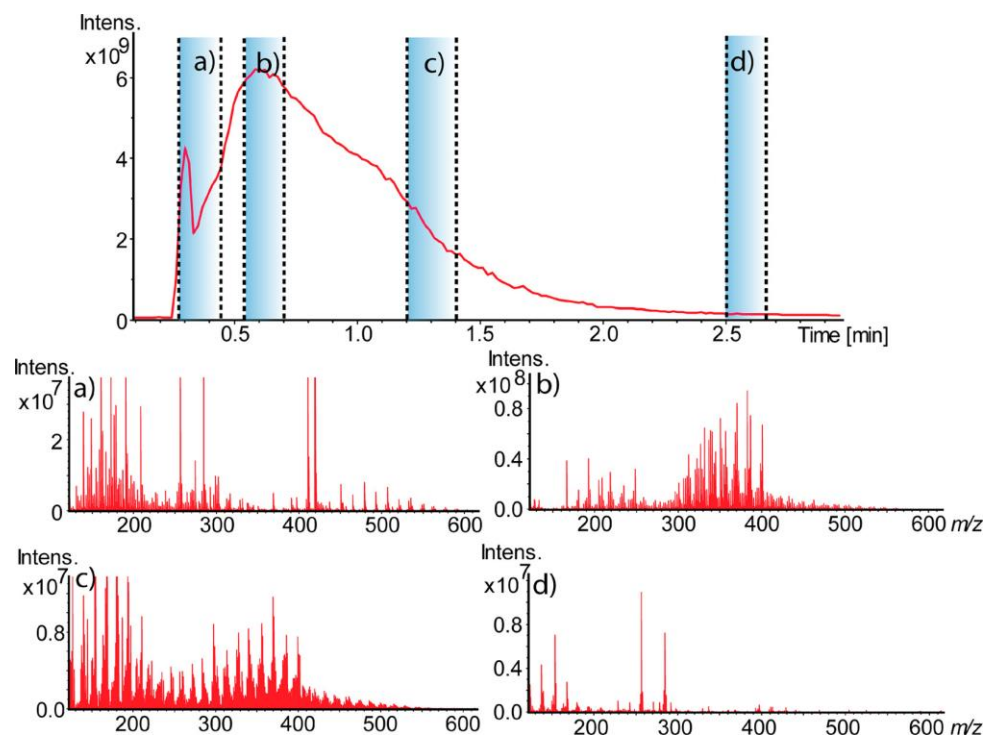
## Hyphenated Solutions – Time-Dependent Data

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### Direct Inlet Probe Atmospheric Pressure Photo and Chemical Ionization Coupled to Ultrahigh Resolution Mass Spectrometry for the Description of Lignocellulosic Biomass

Clément Castilla, Christopher P. Rüger, Stéphane Marcotte, Hélène Lavanant,\* and Carlos Afonso





## Software Solutions – Example: Kairos MS

analytical chemistry

pubs.acs.org/ac

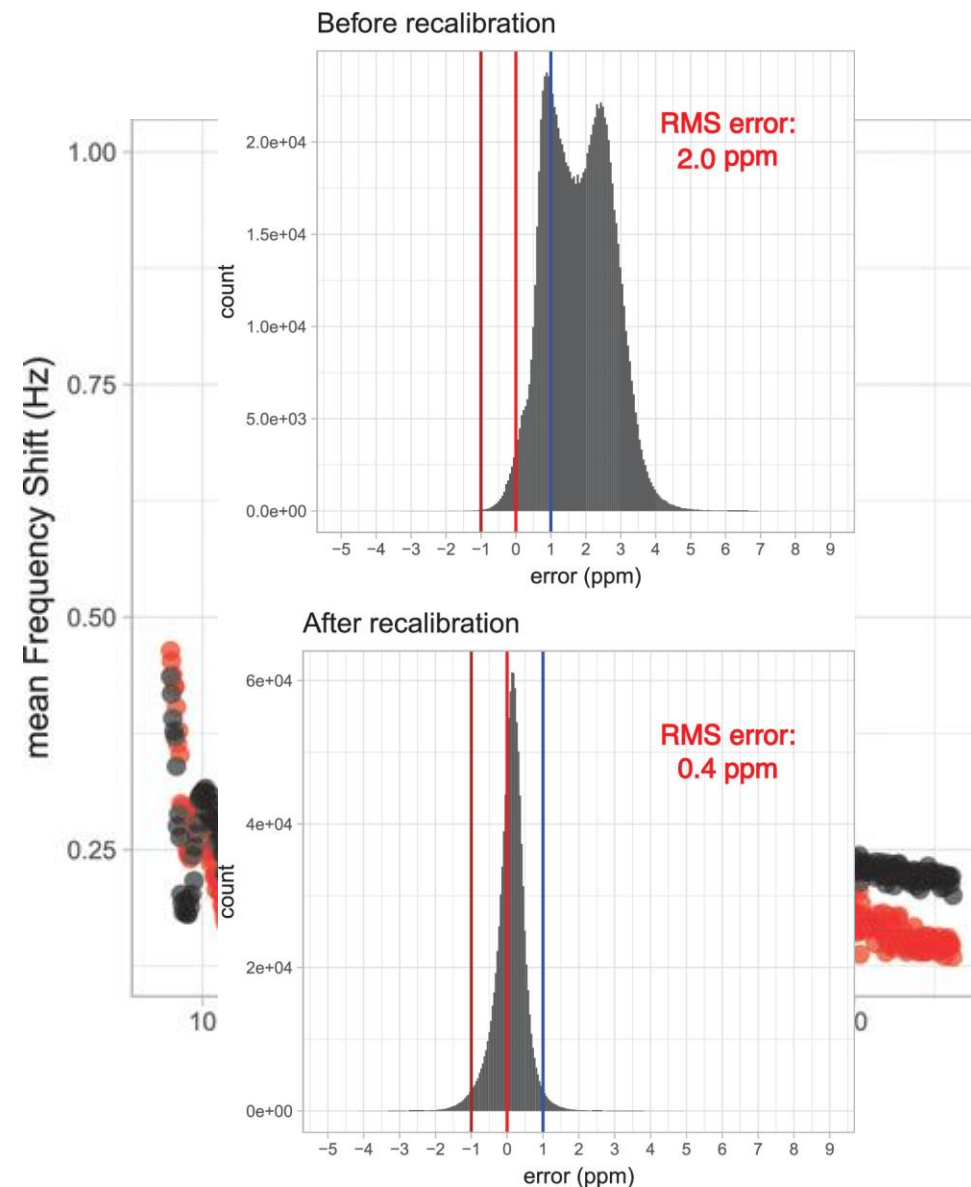
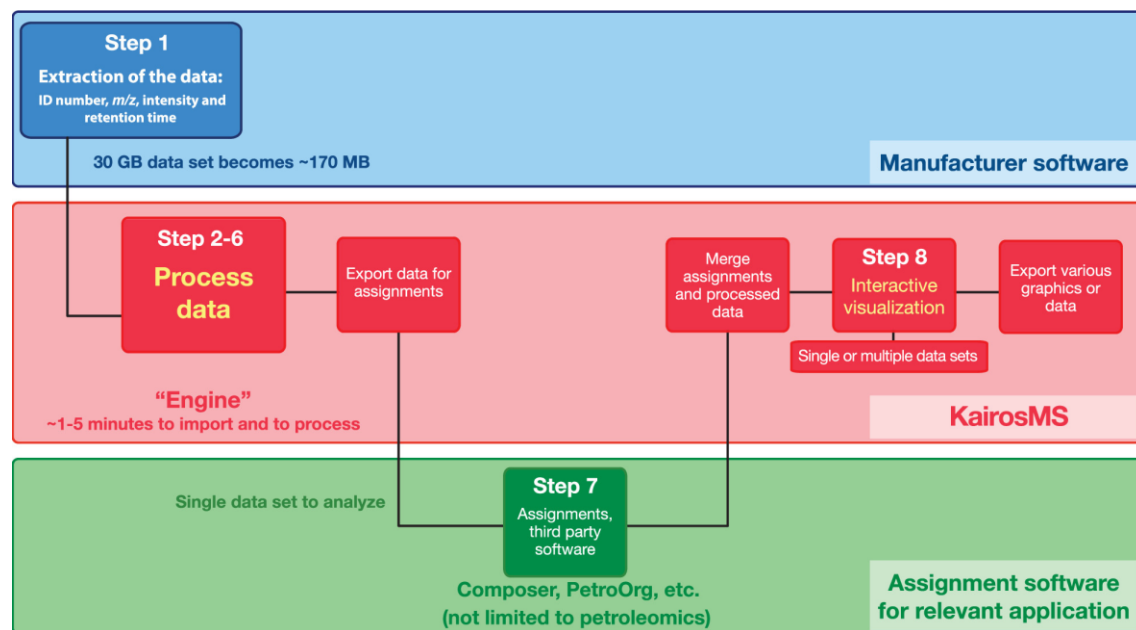
Article

### KairosMS: A New Solution for the Processing of Hyphenated Ultrahigh Resolution Mass Spectrometry Data

Remy Gavard, Hugh E. Jones, Diana Catalina Palacio Lozano, Mary J. Thomas, David Rossell, Simon E. F. Spencer, and Mark P. Barrow\*

Cite This: *Anal. Chem.* 2020, 92, 3775–3786

Read Online



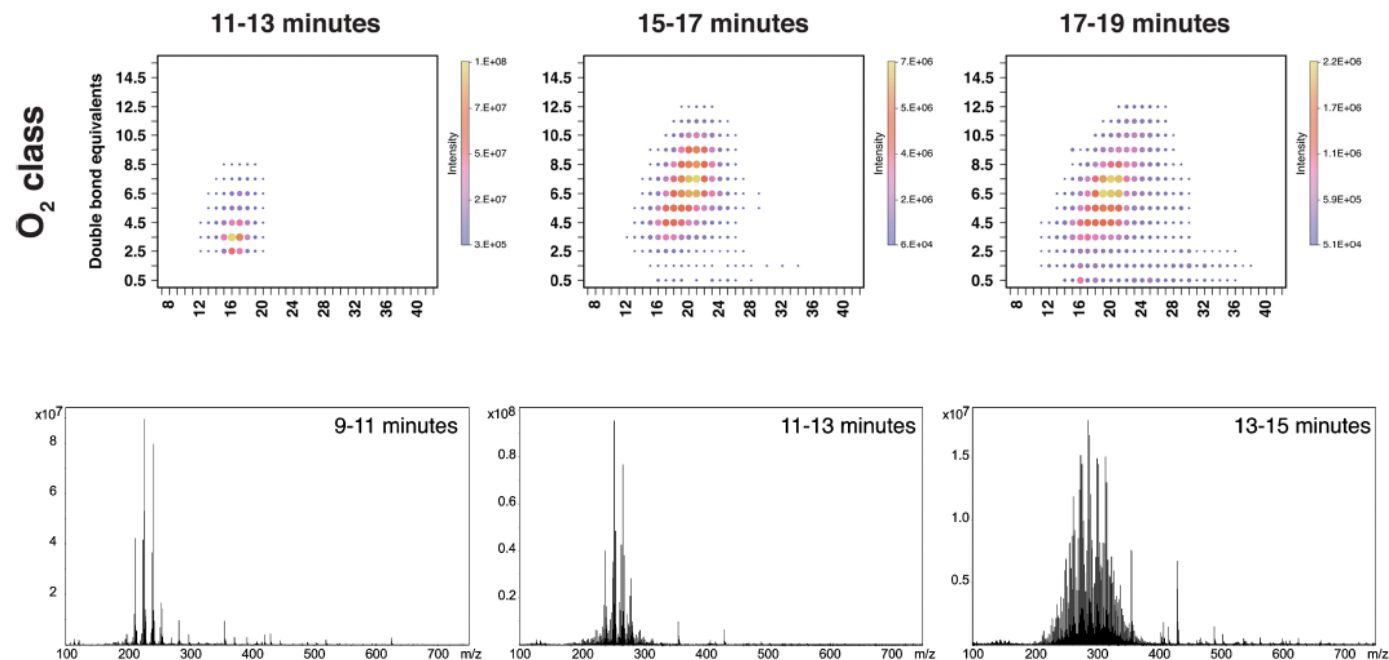
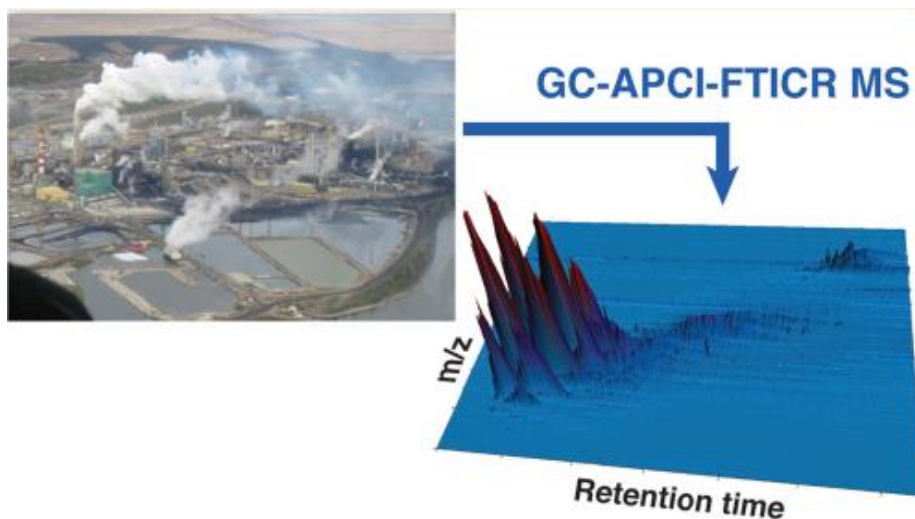
## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation

### An Added Dimension: GC Atmospheric Pressure Chemical Ionization FTICR MS and the Athabasca Oil Sands

Mark P. Barrow,<sup>\*,†</sup> Kerry M. Peru,<sup>‡</sup> and John V. Headley<sup>‡</sup>

<sup>†</sup>Department of Chemistry, University of Warwick, Coventry, CV4 7AL United Kingdom

<sup>‡</sup>Water Science and Technology Division, Environment Canada, 11 Innovation Boulevard, Saskatoon, Saskatchewan S7N 3H5 Canada



## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation

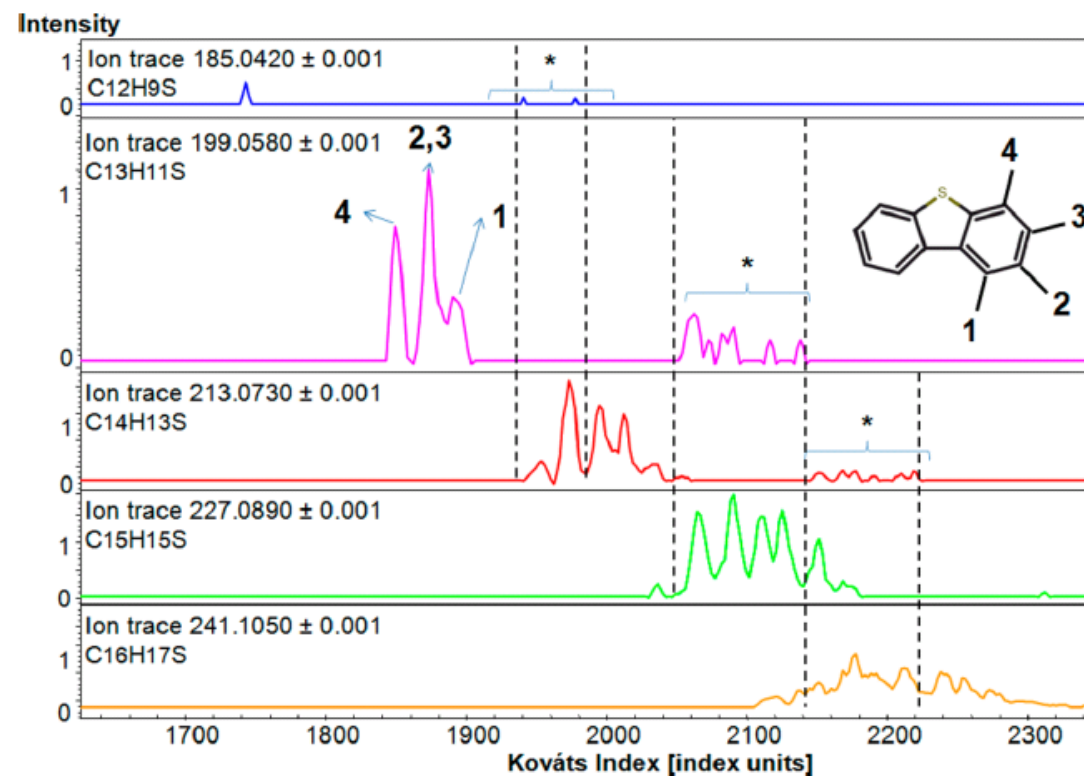
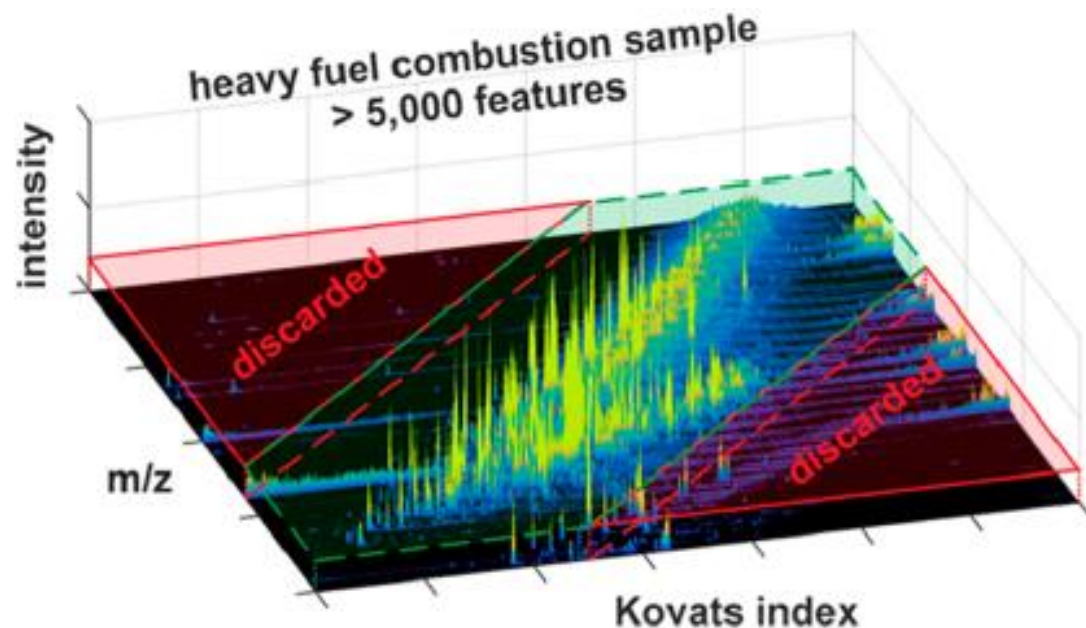
### Gas Chromatography Coupled to Atmospheric Pressure Chemical Ionization FT-ICR Mass Spectrometry for Improvement of Data Reliability

Theo Schwemer,<sup>†,‡</sup> Christopher P. Rüger,<sup>†</sup> Martin Sklorz,<sup>\*,†,§</sup> and Ralf Zimmermann<sup>†,‡,§</sup>

<sup>†</sup>Joint Mass Spectrometry Centre/Chair of Analytical Chemistry, University of Rostock, 18051 Rostock, Germany

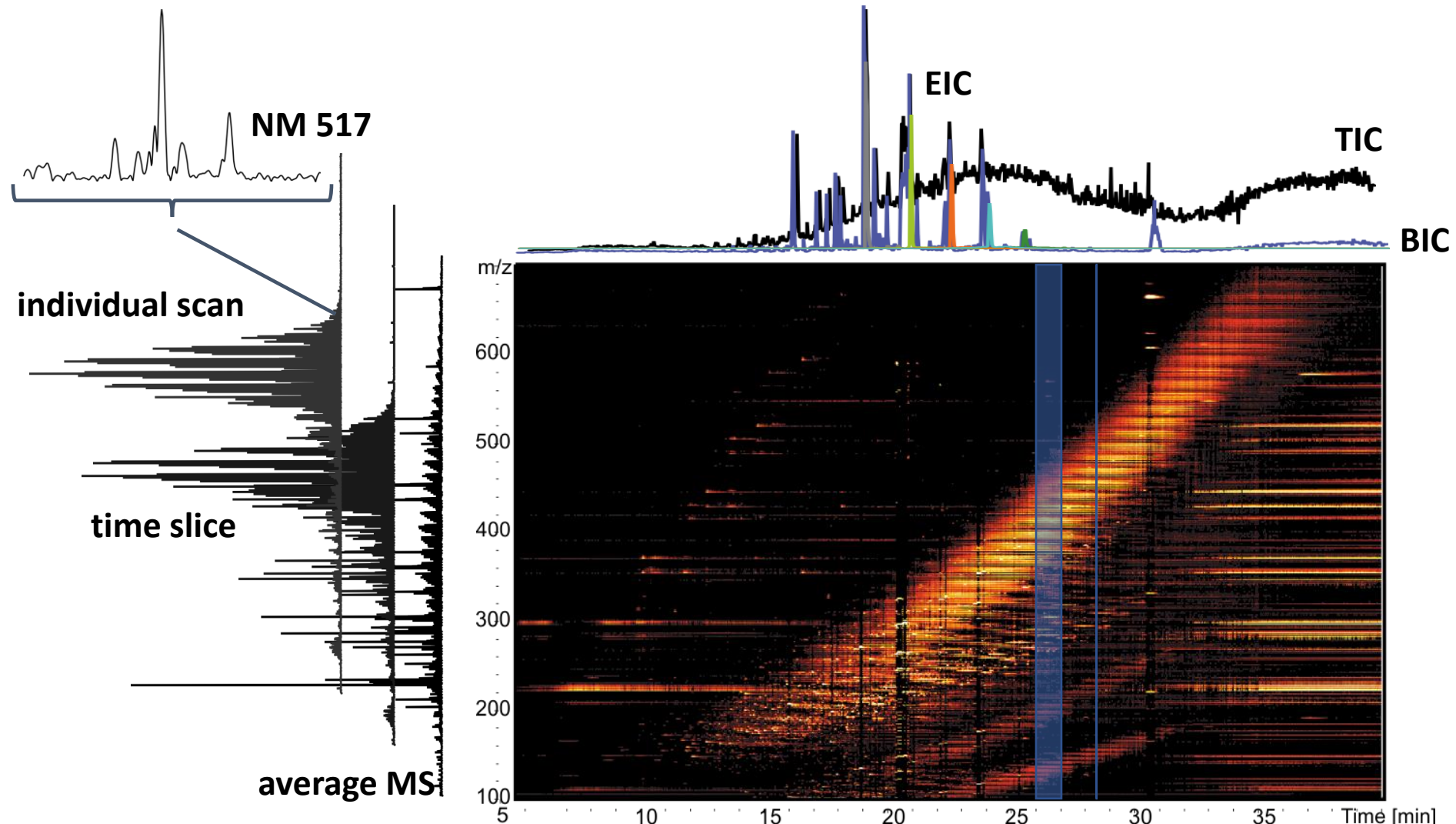
<sup>‡</sup>HICE – Helmholtz Virtual Institute of Complex Molecular Systems in Environmental Health – Aerosols and Health, 85764 Neuherberg, Germany, [www.hice-vi.eu](http://www.hice-vi.eu)

<sup>§</sup>Joint Mass Spectrometry Centre/Cooperation Group Comprehensive Molecular Analytics, Helmholtz Zentrum München, 85764 Neuherberg, Germany





## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation



## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation

Optimized Automatic Noise Level Calculations for FT-ICR MS Data

Bull. Korean Chem. Soc. **2009**, Vol. 30, No. 11 2665

### Optimized Automatic Noise Level Calculations for Broadband FT-ICR Mass Spectra of Petroleum Give More Reliable and Faster Peak Picking Results

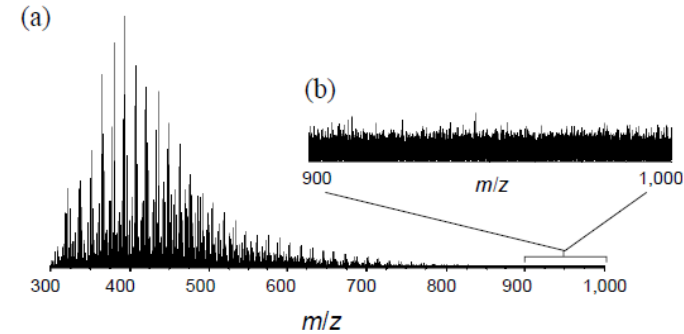
Manhoi Hu, Han Bin Oh,<sup>†</sup> and Sunghwan Kim<sup>\*,‡</sup>

BNF Technology Inc., Daejeon 305-500, Korea

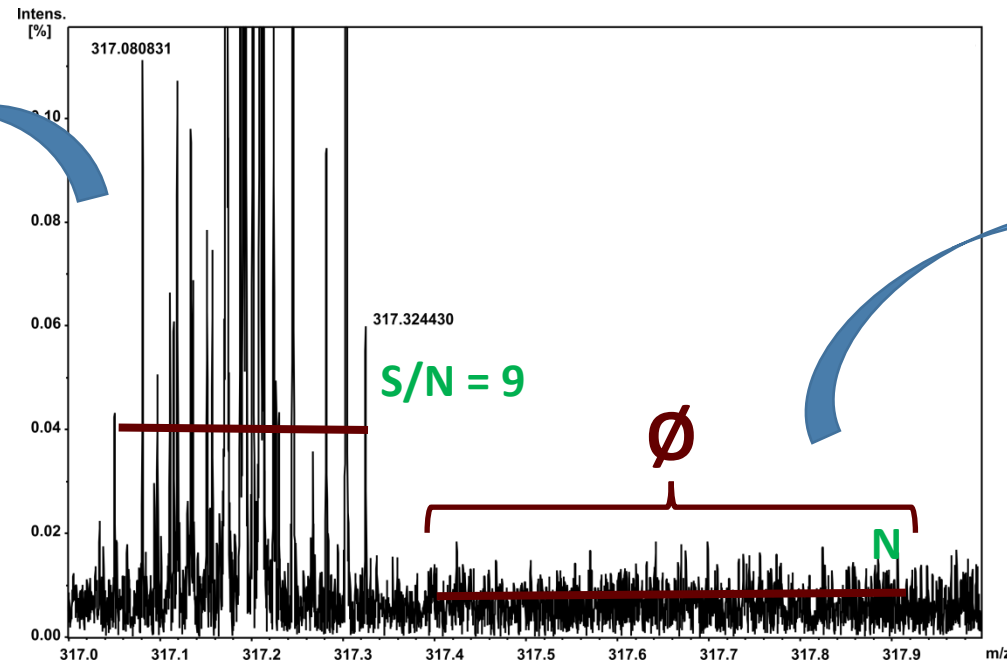
<sup>†</sup>Department of Chemistry, Sogang University, Seoul 121-742, Korea (200811036)

<sup>‡</sup>Kyungpook National University, Department of Chemistry, Daegu 702-701, Korea. \*E-mail: sunghwank@knu.ac.kr

Received July 4, 2009, Accepted September 17, 2009



chemically  
accessed “island”  
for Petroleum

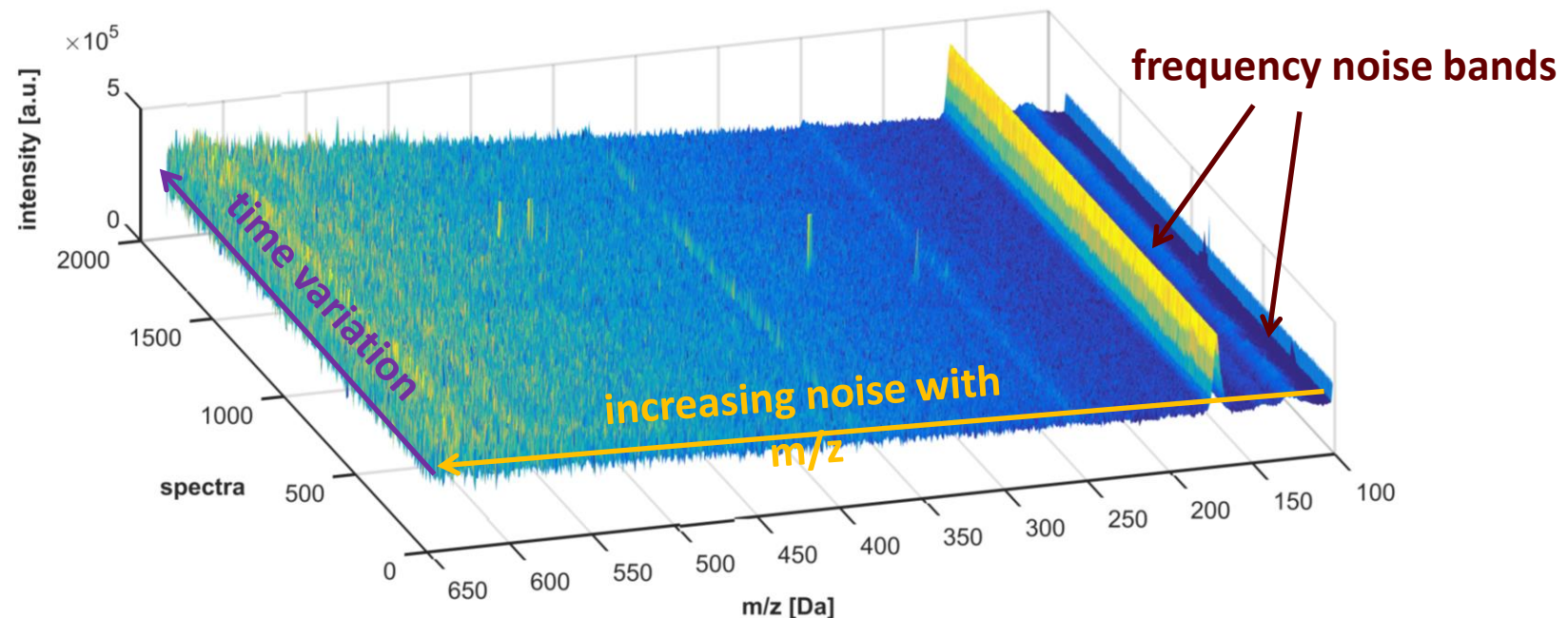


chemically not accessed  
space – noise

## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation

### Noise estimation for GC-APCI run:

(550 nominal  $m/z$ -values \* 2000 spectra = 1.1e6 noise values)



Noise estimation based on the signal height at the  $m/z$  interval chemically not accessible for each nominal mass



## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation

Analytica Chimica Acta 1173 (2021) 338674



Dynamic binning peak detection and assessment of various lipidomics liquid chromatography-mass spectrometry pre-processing platforms

Xiaodong Feng<sup>a</sup>, Wenxuan Zhang<sup>b,c</sup>, Folkert Kuipers<sup>a,b</sup>, Ido Kema<sup>a</sup>, Andrei Barcaru<sup>a,1</sup>, Péter Horvátovich<sup>c,\*,1</sup>

<sup>a</sup> Department of Laboratory Medicine, University Medical Center Groningen, Hanzeplein 1, 9713, GZ Groningen, the Netherlands

<sup>b</sup> Department of Pediatrics, University Medical Center Groningen, Hanzeplein 1, 9713, GZ Groningen, the Netherlands

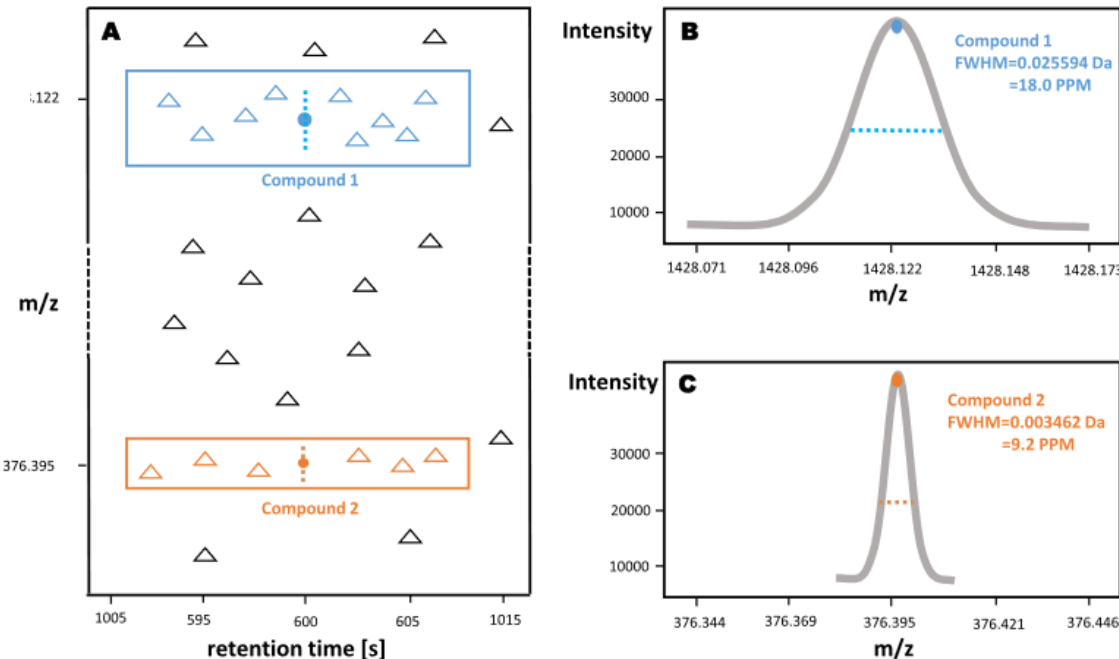
<sup>c</sup> Department of Analytical Biochemistry, University of Groningen, Antonius Deusinglaan 1, 9713, AV Groningen, the Netherlands

This uncertainty is dependent on the  $m/z$ , so a fixed mass tolerance value for peak detection may result in peak merging and/or failure to detect peaks at a specific  $m/z$  range. An alternative is to use a dynamic mass tolerance according to the uncertainty of acquired ions. Thus, the peak detection mass tolerance ( $PDMT$ ) should be set as a function of the MF (usually defined in ppm) and MD. The  $PDMT$  in Da ( $PDMT_{Da}$ ) can be calculated by

$$PDMT_{Da} = MD_{Da} + MF \cdot \frac{m}{z} \cdot 10^{-6} \quad (1)$$

FTICR:

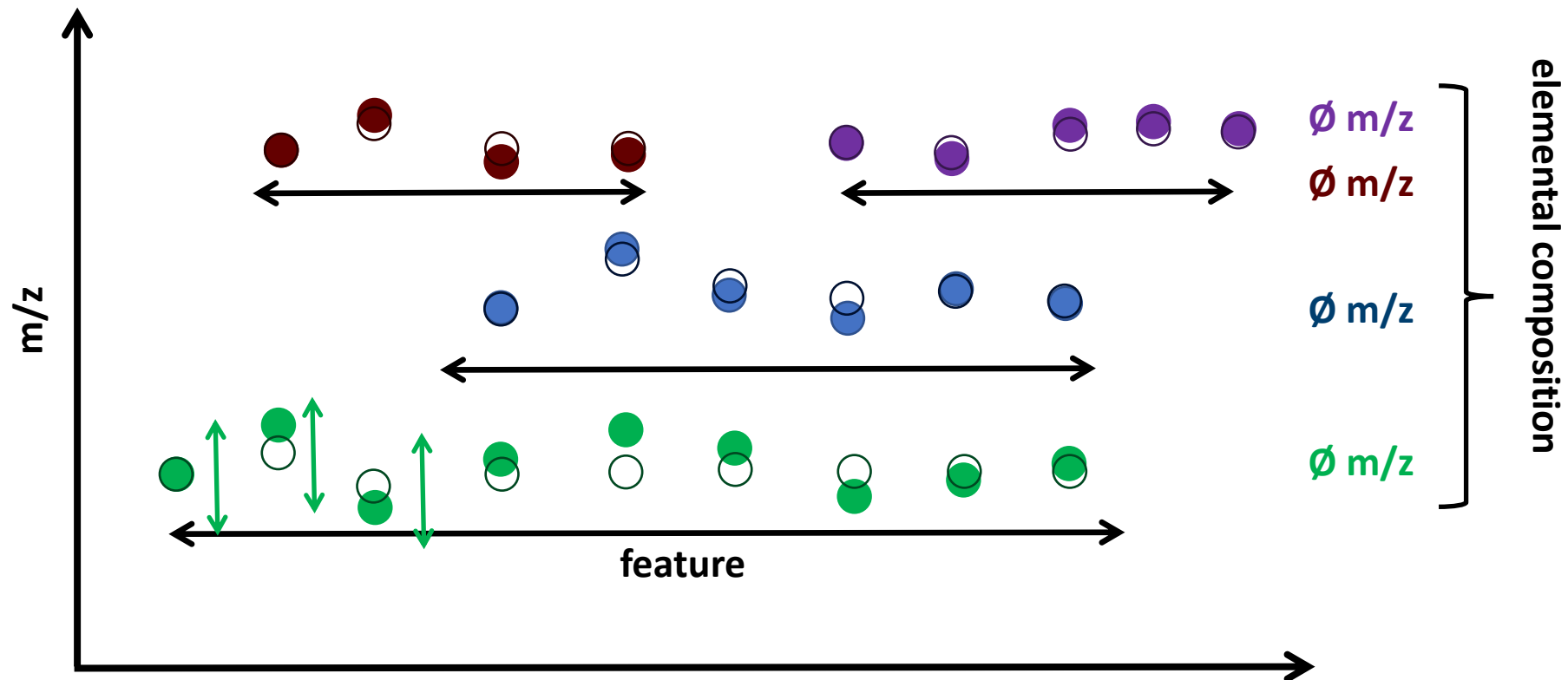
$$PDMT_{Da} = \left( \frac{A_{fti}}{2.35482} \right) \cdot \left( \frac{m}{z} \right)^2 + MF \cdot \frac{m}{z} \cdot 10^{-6} \quad (1a)$$



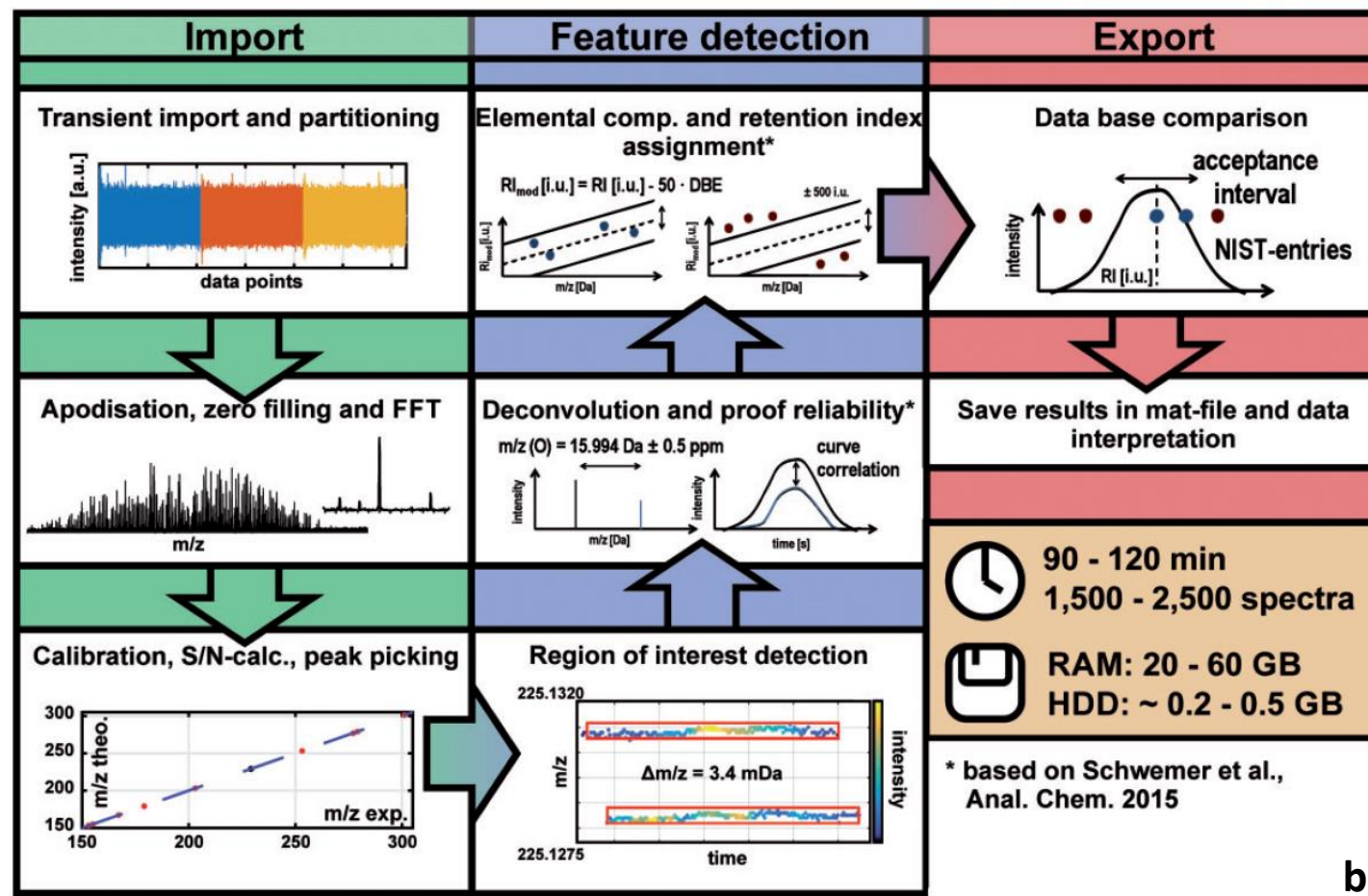
**Fig. 1.** Scheme showing the main aspects of the dynamic binning method demonstrating the change of peak width defined as  $FWHM$  (Full Width at Half Maximum) in function of  $m/z$ . (A) Scatter plot of detected peaks, with retention time in the x-axis and  $m/z$  in the y-axis. (B) Compound 1 is located at a high  $m/z$  1428.1220 Da, which has a large  $FWHM$ , around 0.025594 Da (18.0 ppm). (C) Compound 2 is situated at a low  $m/z$  of 376.3950 Da, which has a small  $FWHM$ , around 0.003462 Da (9.2 ppm).

## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation

- feature detection based on time trace and moving mass window
- automatic discarding of non-chromatographic, too low intense and too short features and deconvolution



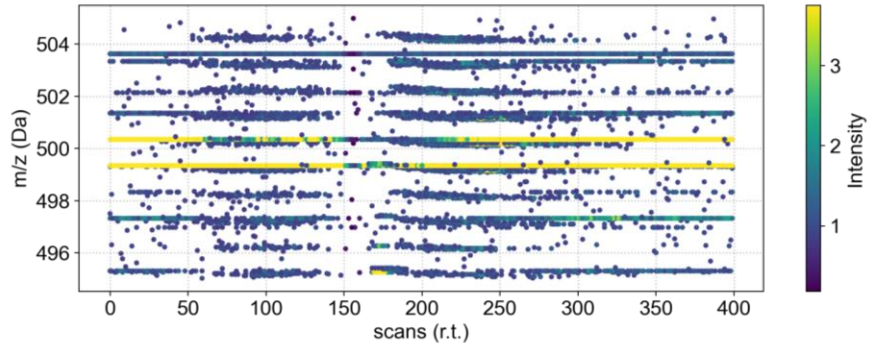
## “Where are the good data?” – Hyphenated Solutions – Example Gas Chromatographic Hyphenation



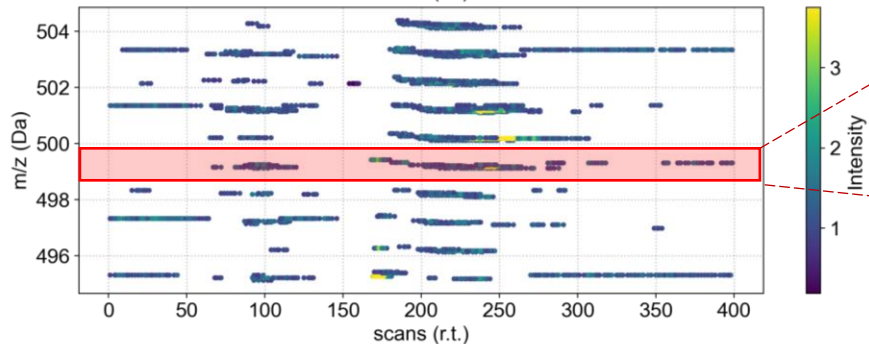
but poor scaling for larger data sets



“Where are the good data?”

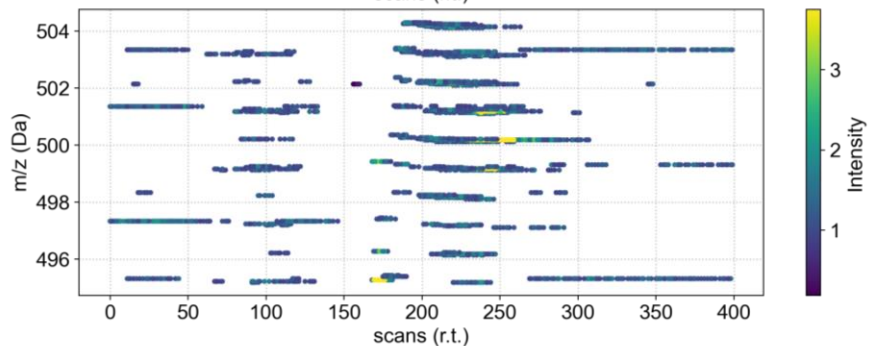


**RAW DATA**



**K-means**

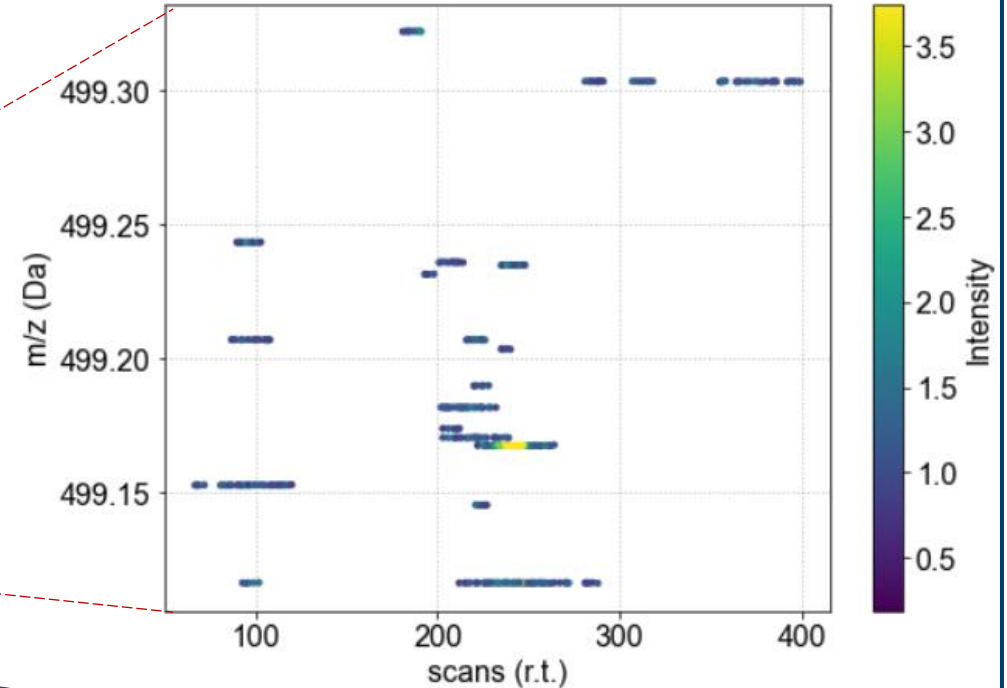
Time: 10.9 s  
ROIs detected: 26896



**Iterative method:**

Time: 3min 7.2s  
ROIs detected: 23350

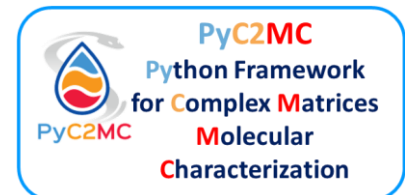
Zoom: 499 – 499.4 Da



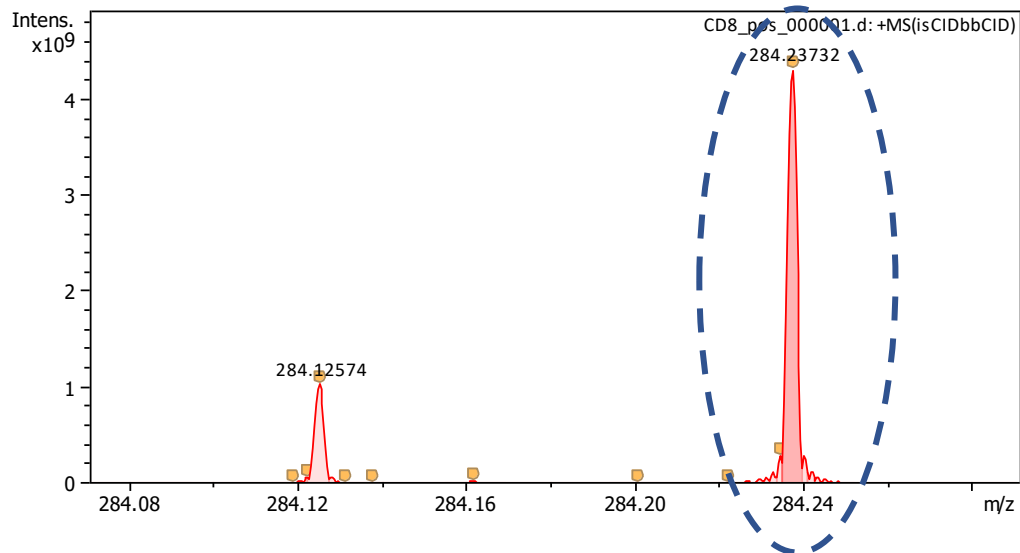
Local grouping approach by nominal mass

**Enhanced Iterative method:**

Time: 27.5 s  
ROIs detected: 23642



## Now we have a list of mass spectrometric features – What to do?



- even with ppm mass accuracy several mathematical possibilities – pure ab-initio calculation difficult / not possible

→ selection rules needed!

some heteroelements

SmartFormula Manually

Lower formula: C<sub>12</sub>  
Upper formula: N<sub>10</sub>O<sub>10</sub>S<sub>10</sub>Na<sub>2</sub>K<sub>2</sub>P<sub>5</sub>  
C 12-n, K 0-2, N 0-10, Na 0-2, O 0-10, P 0-5, S 0-10  
Note: for m < 2000 the elements C, H, N, and O are considered implicitly.

Adducts, pos: M+H  
Adducts, neg: M-H

Measured m/z: 284.23732 Tolerance: 10 mDa Charge: 1

Meas. m/z	#	Ion Formula	Score	m/z	err [mDa]	err [ppm]	mSigma
284.23732	1	C <sub>18</sub> H <sub>31</sub> NNa	25.33	284.23487	-2.45	-8.60	3.0
284.23732	2	C <sub>20</sub> H <sub>30</sub> N	100.00	284.23728	-0.04	-0.14	9.7
284.23732	3	C <sub>16</sub> H <sub>32</sub> NNa <sub>2</sub>	1.38	284.23247	-4.85	-17.07	15.7
284.23732	4	C <sub>15</sub> H <sub>30</sub> N <sub>3</sub> O <sub>2</sub>	4.81	284.23325	-4.06	-14.29	18.6
284.23732	5	C <sub>17</sub> H <sub>34</sub> NS	10.61	284.24065	3.33	11.72	19.8
284.23732	6	C <sub>15</sub> H <sub>35</sub> NNaS	64.66	284.23824	0.93	3.26	24.3
284.23732	7	C <sub>14</sub> H <sub>30</sub> N <sub>5</sub> O	0.03	284.24449	7.17	25.23	24.9
284.23732	8	C <sub>13</sub> H <sub>31</sub> N <sub>3</sub> NaO <sub>2</sub>	0.11	284.23085	-6.47	-22.76	29.4
284.23732	9	C <sub>12</sub> H <sub>34</sub> N <sub>3</sub> O <sub>2</sub> S	57.67	284.23662	-0.69	-2.43	34.2
284.23732	10	C <sub>12</sub> H <sub>35</sub> N <sub>3</sub> PS	0.00	284.22838	-8.93	-31.43	34.7
284.23732	11	C <sub>12</sub> H <sub>36</sub> N <sub>3</sub> P <sub>2</sub>	58.98	284.23790	0.58	2.05	35.5
284.23732	12	C <sub>12</sub> H <sub>31</sub> N <sub>5</sub> NaO	1.31	284.24208	4.77	16.76	35.6
284.23732	13	C <sub>14</sub> H <sub>38</sub> NS <sub>2</sub>	0.07	284.24402	6.70	23.58	39.3

☐ Automatically locate monoisotopic peak Maximum number of formulae 500

☒ Check rings plus double bonds Minimum -0.5 Maximum 40

Electron configuration even

☒ Filter H/C element ratio Minimum H/C: 0 Maximum H/C: 3

☒ Estimate carbon number ☒ Generate immediately

Copy to SmartFormula Parameters Show Pattern

CHNO only

SmartFormula Manually

Lower formula: C<sub>12</sub>  
Upper formula: C 12-n  
Note: for m < 2000 the elements C, H, N, and O are considered implicitly.

Adducts, pos: M+H  
Adducts, neg: M-H

Measured m/z: 284.23732 Tolerance: 10 mDa Charge: 1

Meas. m/z	#	Ion Formula	Score	m/z	err [mDa]	err [ppm]	mSigma	rdl
284.23732	1	C <sub>20</sub> H <sub>30</sub> N	100.00	284.23728	-0.04	-0.14	9.7	7.0
284.23732	2	C <sub>15</sub> H <sub>30</sub> N <sub>3</sub> O <sub>2</sub>	4.78	284.23325	-4.06	-14.29	18.8	3.0
284.23732	3	C <sub>14</sub> H <sub>30</sub> N <sub>5</sub> O	0.03	284.24449	7.17	25.23	25.0	3.0

☐ Automatically locate monoisotopic peak Maximum number of formulae 500

☒ Check rings plus double bonds Minimum -0.5 Maximum 40

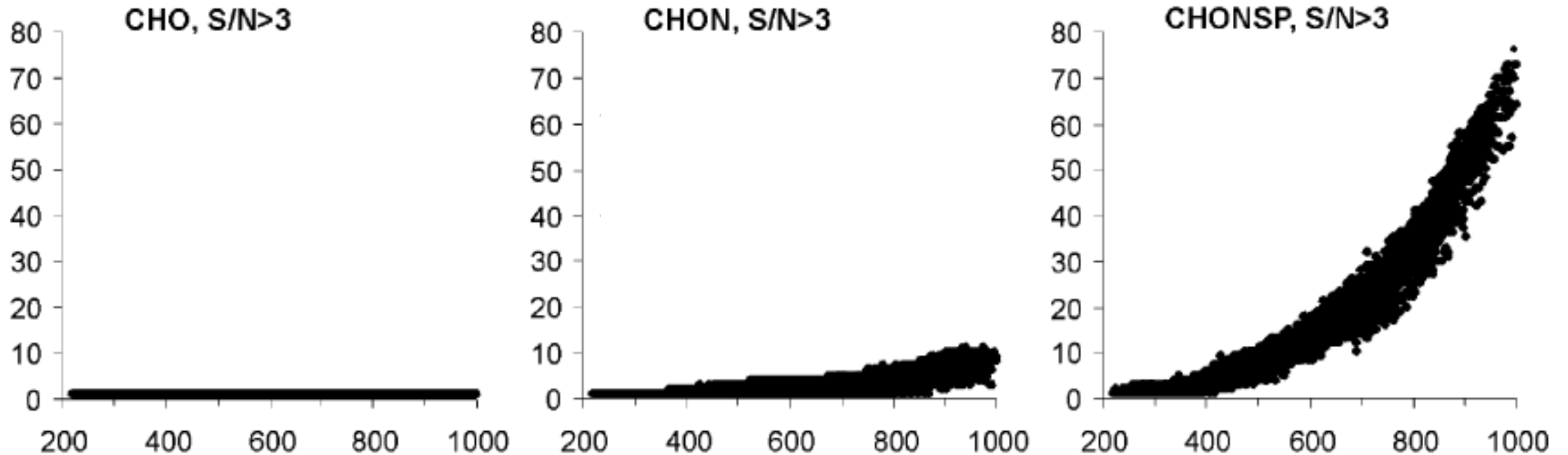
Electron configuration even

☒ Filter H/C element ratio Minimum H/C: 0 Maximum H/C: 3

☒ Estimate carbon number ☒ Generate immediately

Copy to SmartFormula Parameters Show Pattern

# From Mass to Chemistry



$\Delta m/m < 1 \text{ ppm}$

Suwannee River Fulvic Acid Standard, ESI negativ  $\approx 8,500$  Peaks

Koch et al., *Anal.Chem.*, 1758-1763 (2007)



## Attribution of elemental compositions:

“Seven golden rules”

### 1 . Restriction to element numbers

**Table 1: Restrictions for number of elements during formula generation for small molecules based on examination of the DNP and Wiley mass spectral databases. For each element, the higher count was taken for denominating the element restriction rule #1**

Mass Range [Da]	Library	C max	H max	N max	O max	P max	S max	F max	Cl max	Br max	Si max
< 500	DNP	29	72	10	18	4	7	15	8	5	
	Wiley	39	72	20	20	9	10	16	10	4	8
< 1000	DNP	66	126	25	27	6	8	16	11	8	
	Wiley	78	126	20	27	9	14	34	12	8	14
< 2000	DNP	115	236	32	63	6	8	16	11	8	
	Wiley	156	180	20	40	9	14	48	12	10	15
< 3000	DNP	162	208	48	78	6	9	16	11	8	

## Attribution of elemental compositions:

“Seven golden rules”

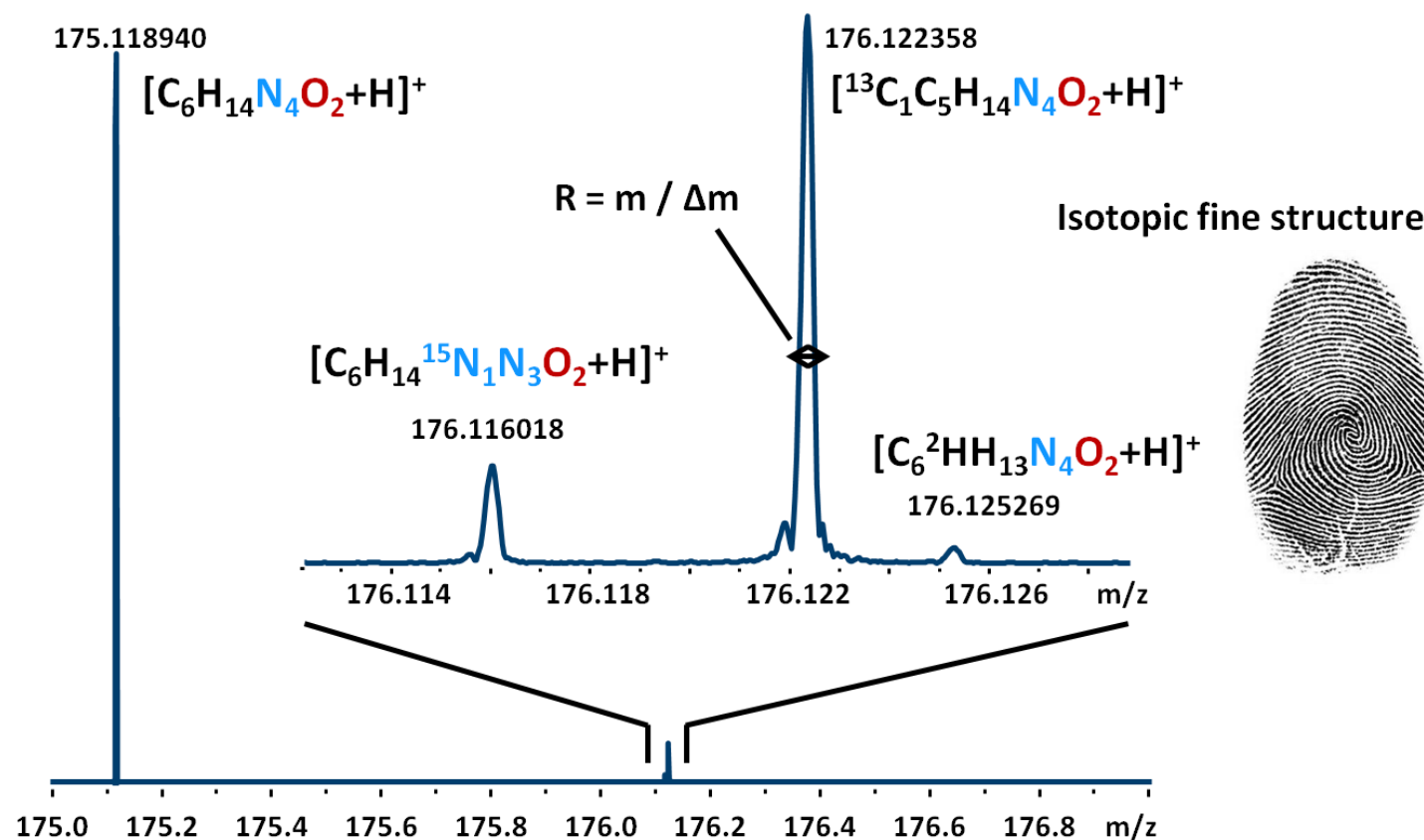
### 2. LEWIS and SENIOR check

- i) The sum of valences or the total number of atoms having odd valences is even;
- ii) The sum of valences is greater than or equal to twice the maximum valence;
- iii) The sum of valences is greater than or equal to twice the number of atoms minus 1.

## Attribution of elemental compositions:

“Seven golden rules”

### 3. Isotopic pattern filter

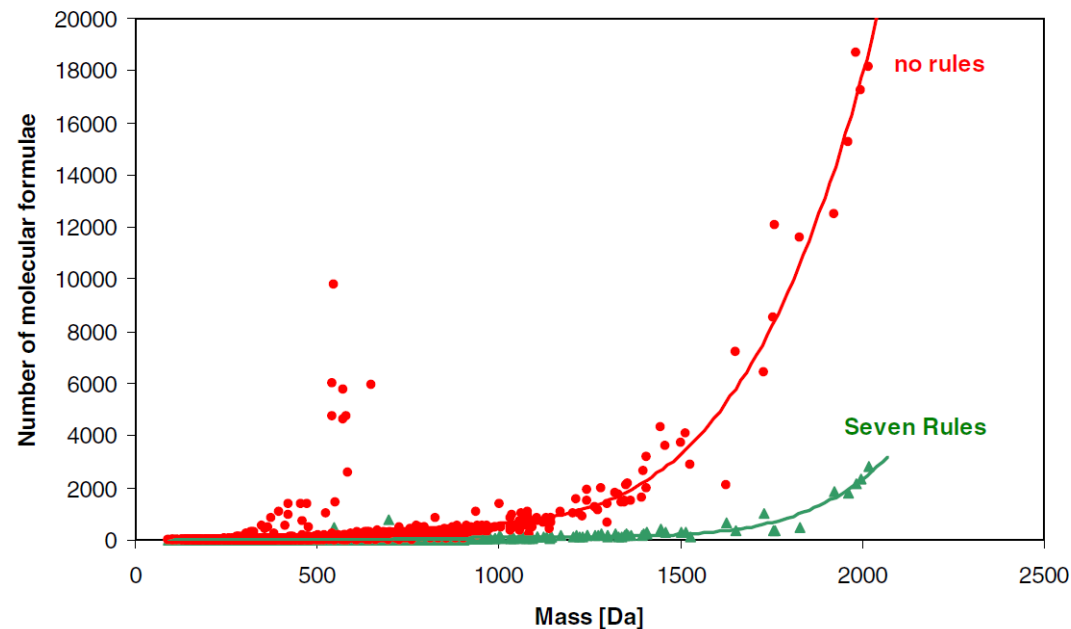




## Attribution of elemental compositions:

### “Seven golden rules”

- 1) apply heuristic restrictions for number of elements during formula generation
- 2) perform LEWIS and SENIOR check
- 3) perform isotopic pattern filter
- 4) perform H/C ratio check (hydrogen/carbon ratio)
- 5) perform NOPS ratio check (N, O, P, S/C ratios)
- 6) perform heuristic HNOPS probability check (H, N, O, P, S/C high probability ratios)
- 7) perform -TMS check (for GC-MS if a silylation step is involved)




**Figure 5**  
 Mass dependence of calculated, chemically possible formulas derived from 1,200 randomly selected DNP molecules, imposed with simulated 3 ppm mass accuracy  $\pm$  5% isotope ratio measurement errors. Red graph: number of calculated formulas with common molecular generators. Green graph: number of formulas constrained by the seven rules. Outliers around 600 Dalton were found to be halogen containing compounds.

## Available Software Solutions

- vendor Solutions, e.g., Bruker Data Analysis, ThermoFisher Xcalibur, Agilent Mass Hunter, etc.
- but most often strong limitations in-handling complex data, isotopologues, validation concepts,...

## Custom Software Solution:

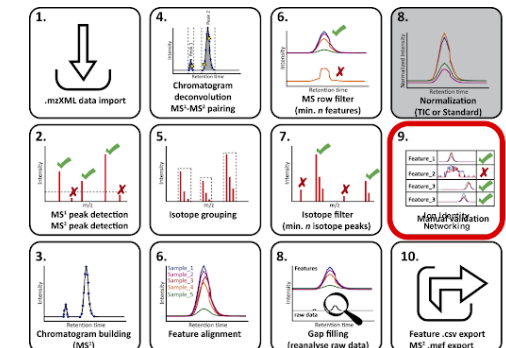
- Commercial: PetroOrg  Sierra Analytics Composer
- Free Software: CERES (Univ. Rostock), OpenMS, mMass, MZmine / MZmine2
- Specialized Solutions:
  - OCEAN (Univ. Oldenburg for DOM)
  - ...



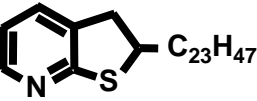


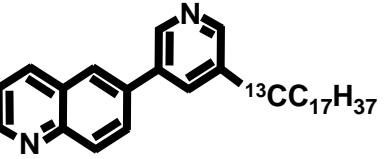
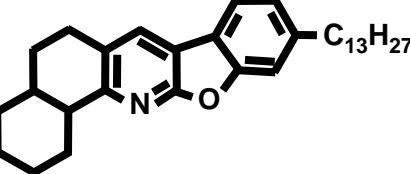
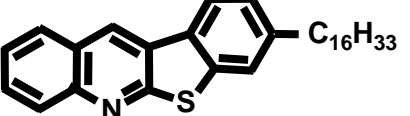
Sierra Analytics Composer



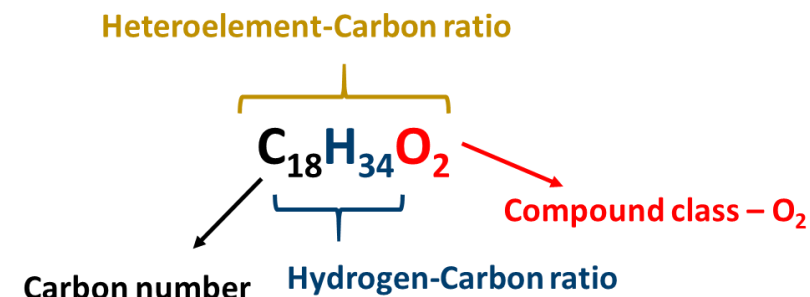
Peak-by-Peak



## Characteristic mass splits and critical aspects in elemental composition attribution:

		Doublet	$m_2 - m_1$	$\frac{m_2}{m_2 - m_1}$
	459.3899 $C_{30}H_{53}NS$		<b>3.4 mDa</b>	<b>135,000</b>
	459.3865 $C_{33}H_{49}N$			
	459.3695 $^{13}CC_{31}H_{46}N_2$			
	459.3501 $C_{32}H_{45}NO$			
	459.2960 $C_{31}H_{41}NS$			

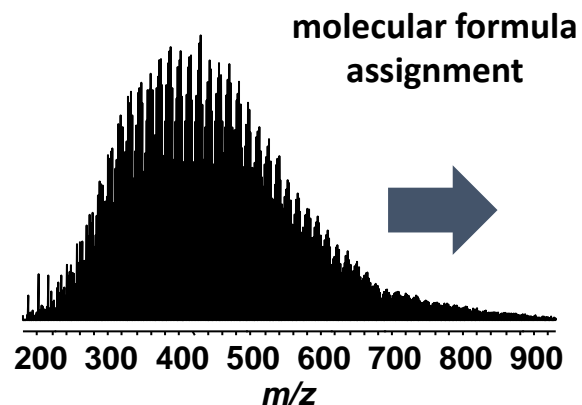
→ Let us imagine we attributed the correct elemental compositions. How to handle several thousands of formula efficiently?



Rodgers, R., Presentation at Mag Lab Summer School 2009

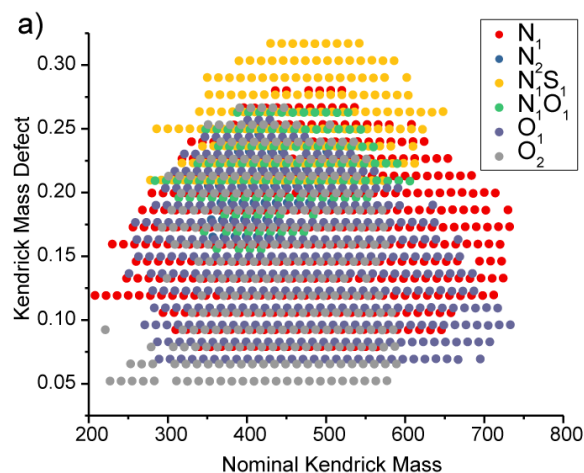
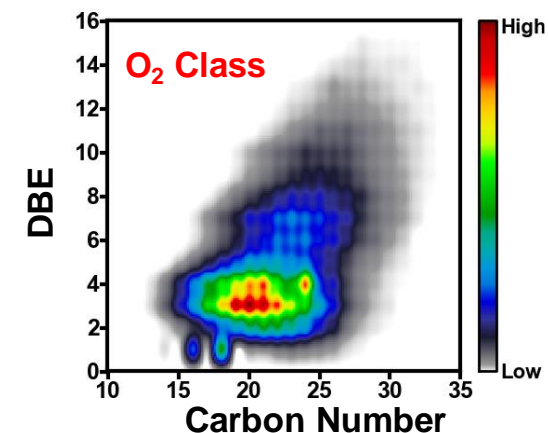


## Complex Mixture

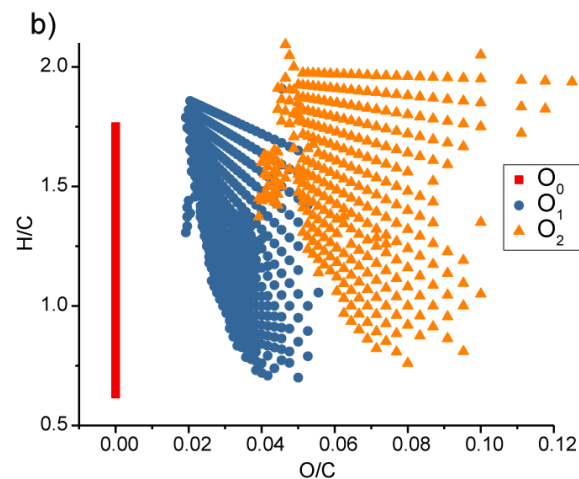


Molecular Formula	Measured Mass	Calculated Mass	ppm error
C <sub>29</sub> H <sub>42</sub> N <sub>1</sub>	404.3323	404.33227	0.16
C <sub>30</sub> H <sub>44</sub> N <sub>1</sub>	418.348	418.34792	0.18
C <sub>31</sub> H <sub>46</sub> N <sub>1</sub>	432.3638	432.36357	0.48
C <sub>32</sub> H <sub>50</sub> N <sub>1</sub>	446.3793	446.37922	0.15
C <sub>27</sub> H <sub>47</sub> O <sub>2</sub>	403.3582	403.35815	0.09
C <sub>28</sub> H <sub>49</sub> O <sub>2</sub>	417.3738	417.37380	-0.13
C <sub>29</sub> H <sub>51</sub> O <sub>2</sub>	431.3895	431.38945	0.18
C <sub>30</sub> H <sub>53</sub> O <sub>2</sub>	445.4052	445.40510	0.12
C <sub>31</sub> H <sub>55</sub> O <sub>2</sub>	459.4210	459.42075	0.49
C <sub>32</sub> H <sub>57</sub> O <sub>2</sub>	473.4364	473.43640	-0.07

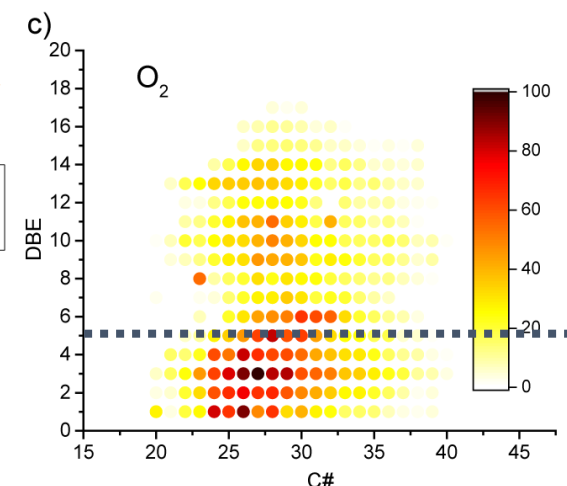
data grouping and visualization



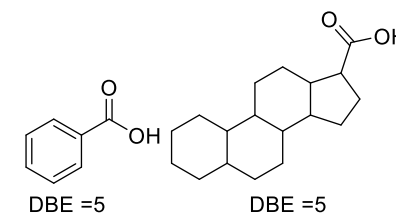
Kendrick Diagram



van Krevelen Diagram



DBE vs C#



$$DBE = 1 + \#C - \#H/2 + \#N/2$$

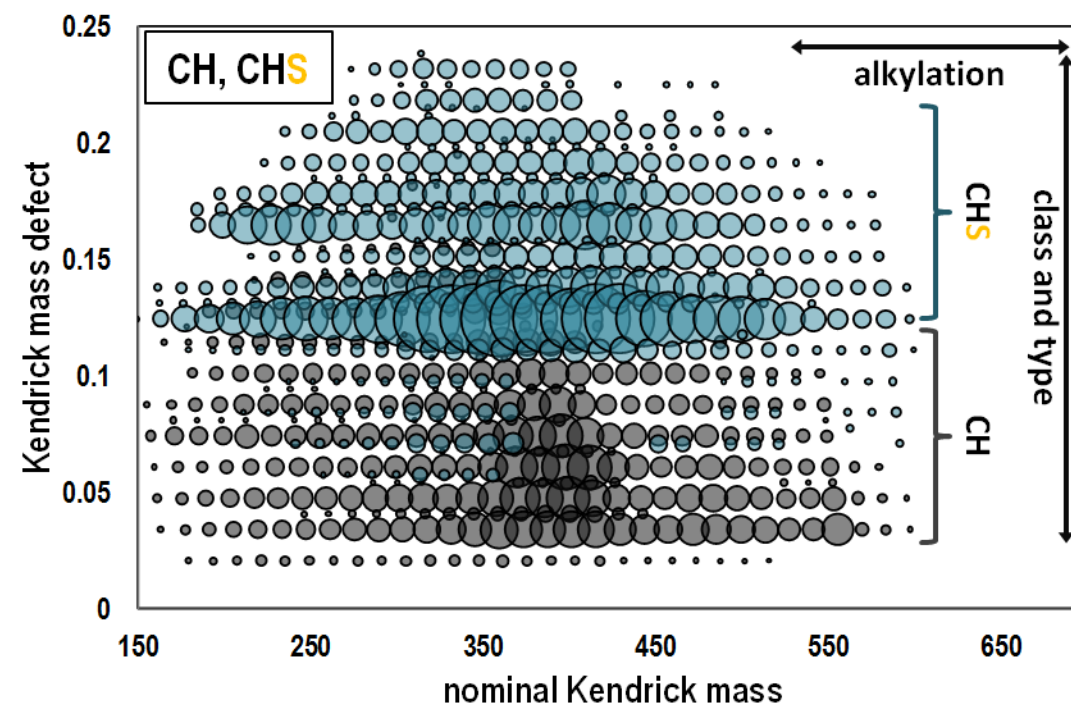
## Visualization and grouping of elemental compositions – Kendrick Mass Defect Diagram

$$KM = \text{mass}_{\text{detect}} \left( \frac{14.00000}{14.01565} \right)$$

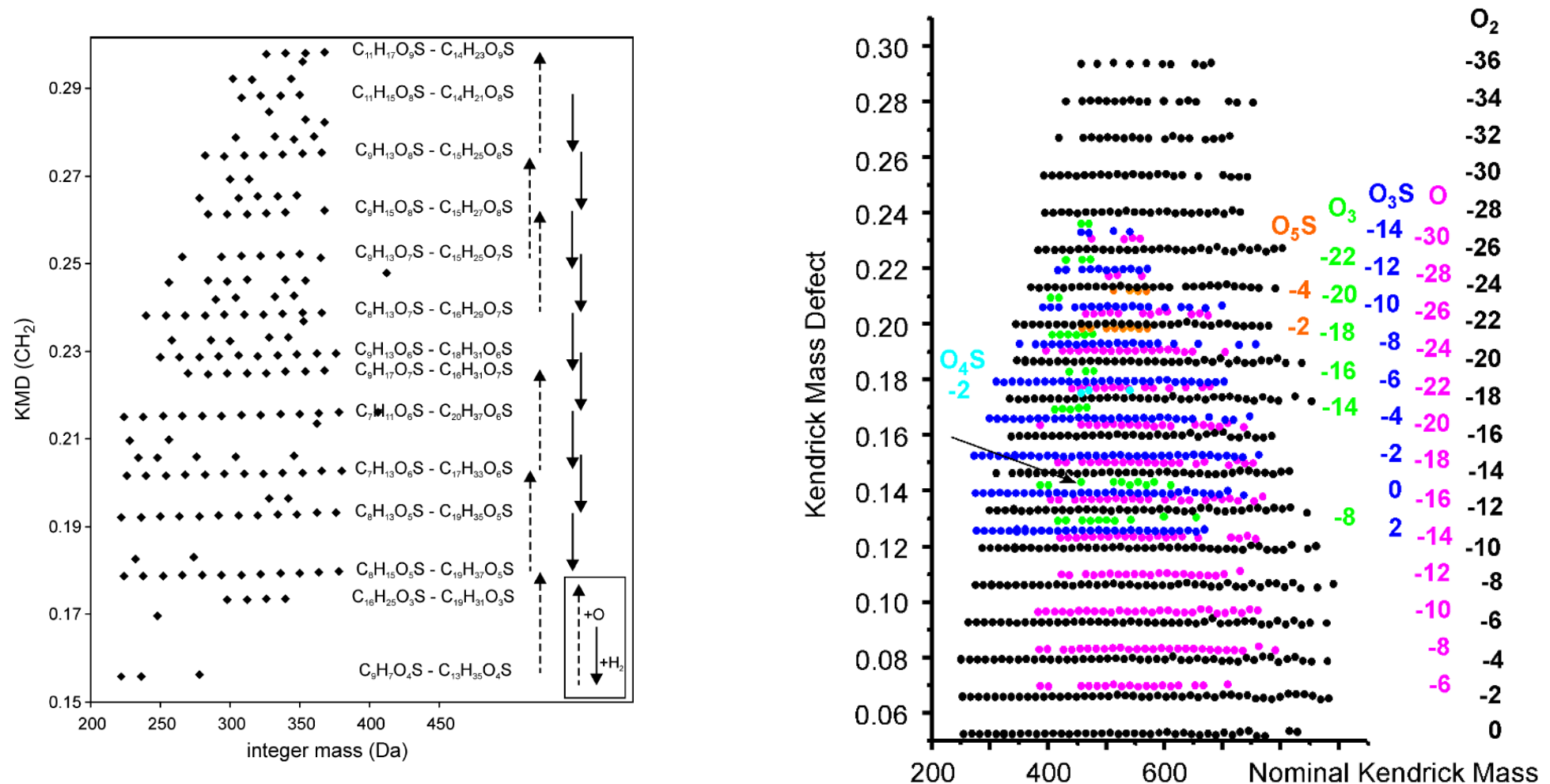
$$KMD = \text{mass}_{\text{integer}} - KM$$

$$\begin{aligned} KMD(C_6H_6) &= 78.04695 \times (14.0000/14.01565) \\ &- 78 = 0.9598 \end{aligned}$$

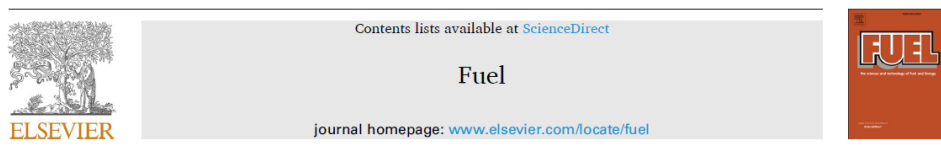
$$\begin{aligned} KMD(C_7H_8) &= 92.06300 \times (14.0000/14.01565) \\ &- 92 = 0.9598 \end{aligned}$$



## Visualization and grouping of elemental compositions – Kendrick Mass Defect Diagram







Full Length Article

Selective characterization of petroporphyrins in shipping fuels and their corresponding emissions using electron-transfer matrix-assisted laser desorption/ionization Fourier transform ion cyclotron resonance mass spectrometry

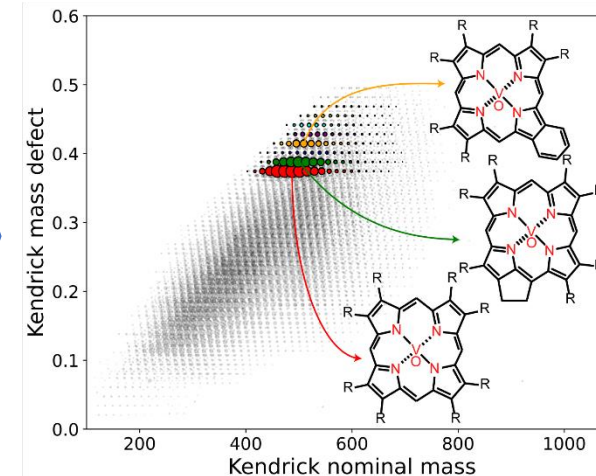
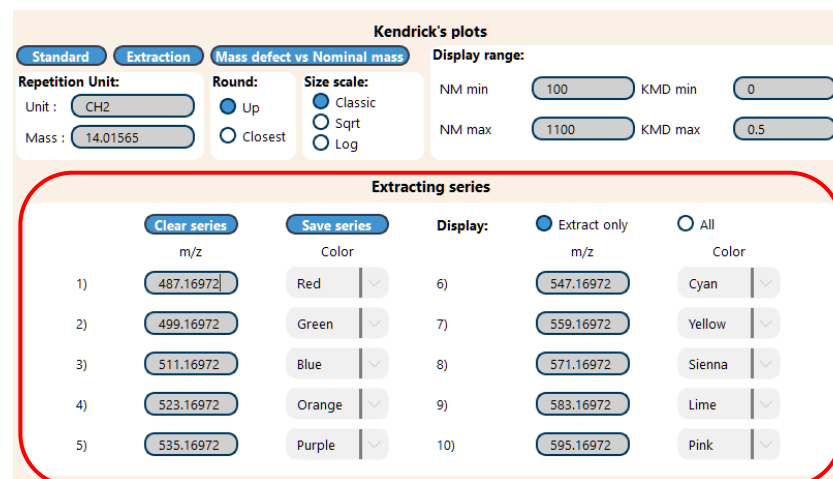
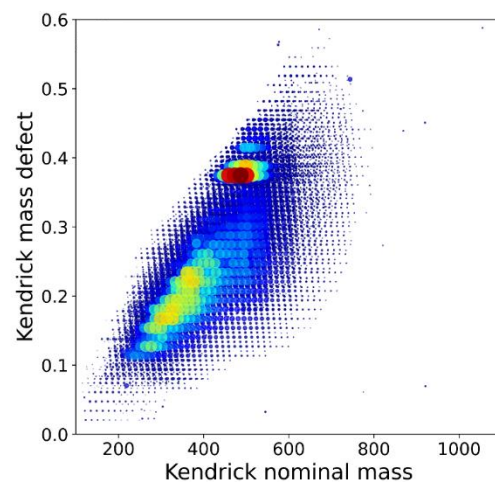
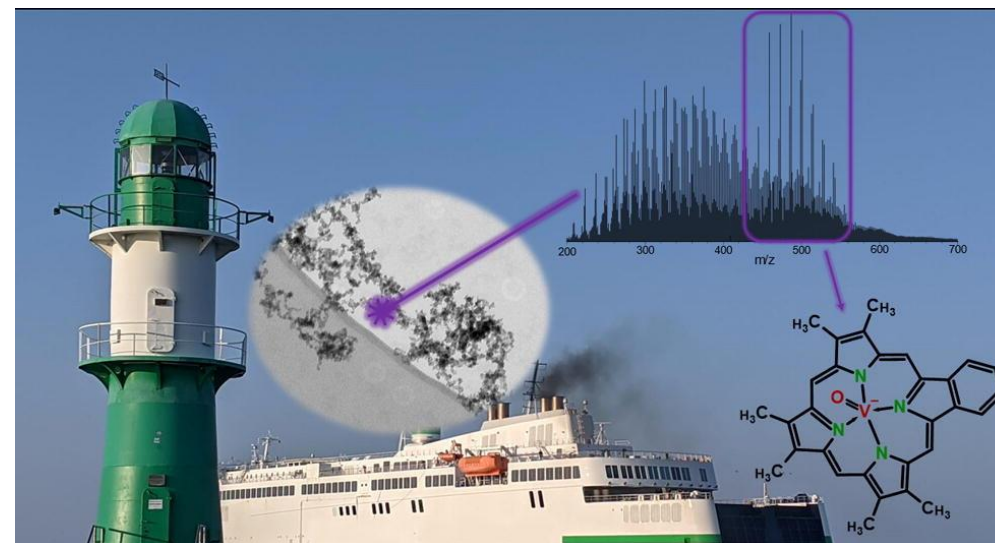
Maxime Sueur<sup>a,b</sup>, Christopher P. Rüger<sup>b,c,d,\*</sup>, Julien F. Maillard<sup>a,b</sup>, Hélène Lavanant<sup>a</sup>, Ralf Zimmermann<sup>c,d</sup>, Carlos Afonso<sup>a,b</sup>

<sup>a</sup> Normandie Université, COBRA, UMR 6014 et FR 3030, Université de Rouen, INSA de Rouen-Normandie, CNRS, IRCOF, Mont Saint Aignan Cedex, France

<sup>b</sup> International Joint Laboratory – IC2MC: Complex Matrices Molecular Characterization, TRTG, BP 27, 76700 Harfleur, France

<sup>c</sup> Joint Mass Spectrometry Centre/Chair of Analytical Chemistry, University of Rostock, 18059 Rostock, Germany

<sup>d</sup> Department Life, Light & Matter (LLM), University of Rostock, 18051 Rostock, Germany



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Article

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## Investigating the Trace Polar Species Present in Diesel Using High-Resolution Mass Spectrometry and Selective Ionization Techniques

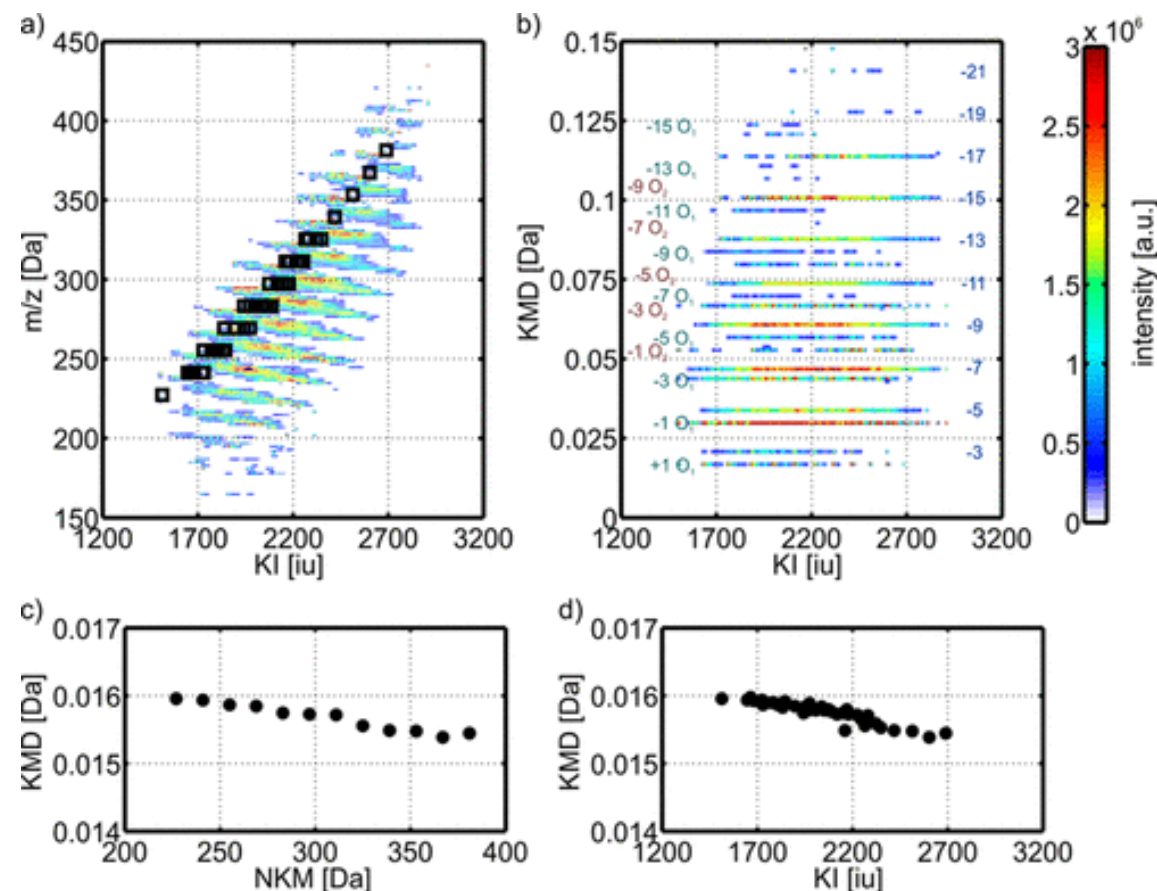
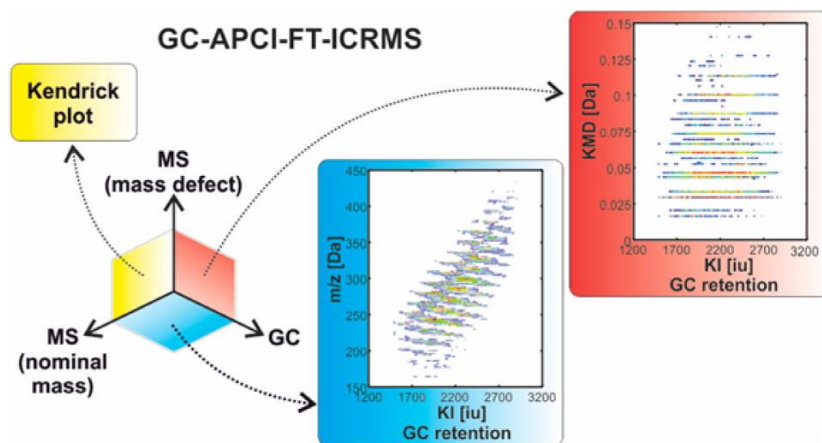
Elize Smit,<sup>\*,†</sup> Christopher P. Rüger,<sup>‡</sup> Martin Sklorz,<sup>‡,§</sup> Stefan De Goede,<sup>∇</sup> Ralf Zimmermann,<sup>‡,§</sup> and Egmont R. Rohwer<sup>†</sup>

<sup>†</sup>Department of Chemistry, University of Pretoria, Lynnwood Road, Pretoria, South Africa

<sup>‡</sup>Joint Mass Spectrometry Centre/Chair of Analytical Chemistry, University of Rostock, Rostock, Germany

<sup>§</sup>Joint Mass Spectrometry Centre/Cooperation Group Comprehensive Molecular Analytics, Helmholtz Zentrum München, Neuherberg, Germany

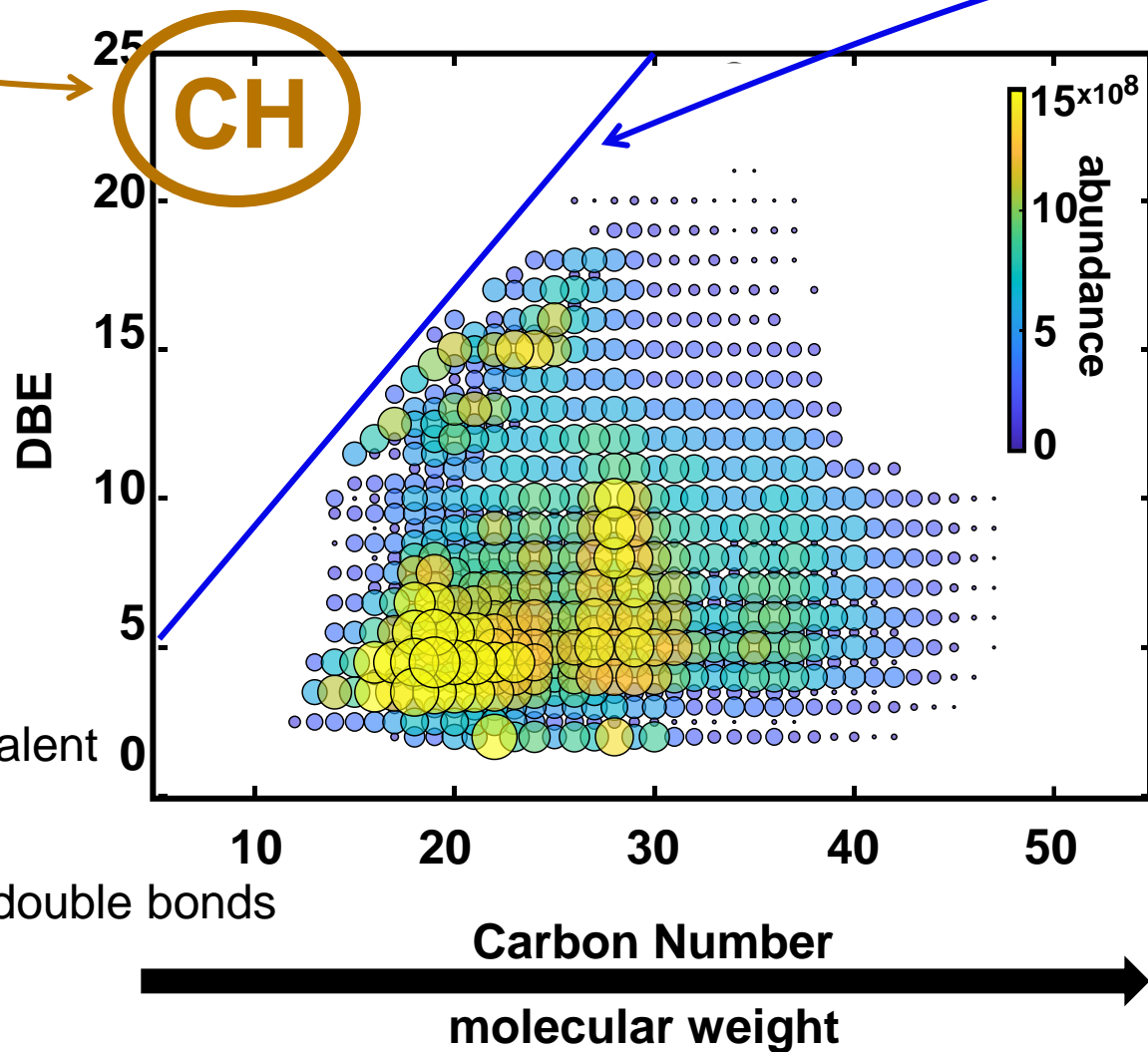
<sup>∇</sup>Sasol Southern Africa Energy, Energy Technology, Klasie Havenga Street, Sasolburg, South Africa



Different graphical representations of the chromatographic separation and mass spectral data obtained by GC-APCI-FT-ICRMS for sample CD5: (a) m/z vs KI values, with peaks of the +1O<sub>1</sub> subclass highlighted in black. (b) Plot of Kendrick mass defect vs KI values showing a more orthogonal distribution of peaks throughout the two-dimensional space. (c,d) Comparison of the +1O<sub>1</sub> subclass with (panel d) and without (panel c) chromatographic separation being taken into account.

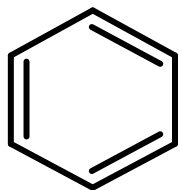
# Handling Attribution Data – DBE/#C

All compounds containing only CH are summarized in the CH-class, species containing C, H, and one O-atom are summarized in the O1-class



$$DBE = \#C - \frac{\#H}{2} + \frac{\#N}{2}$$

Ring and double bond equivalent



one ring + three double bonds  
DBE = 4



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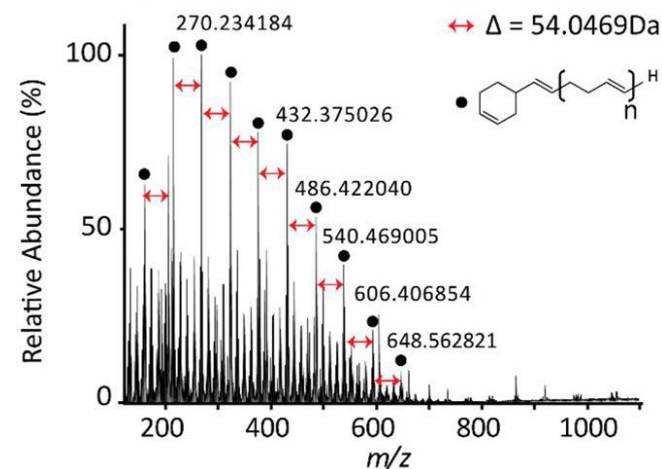
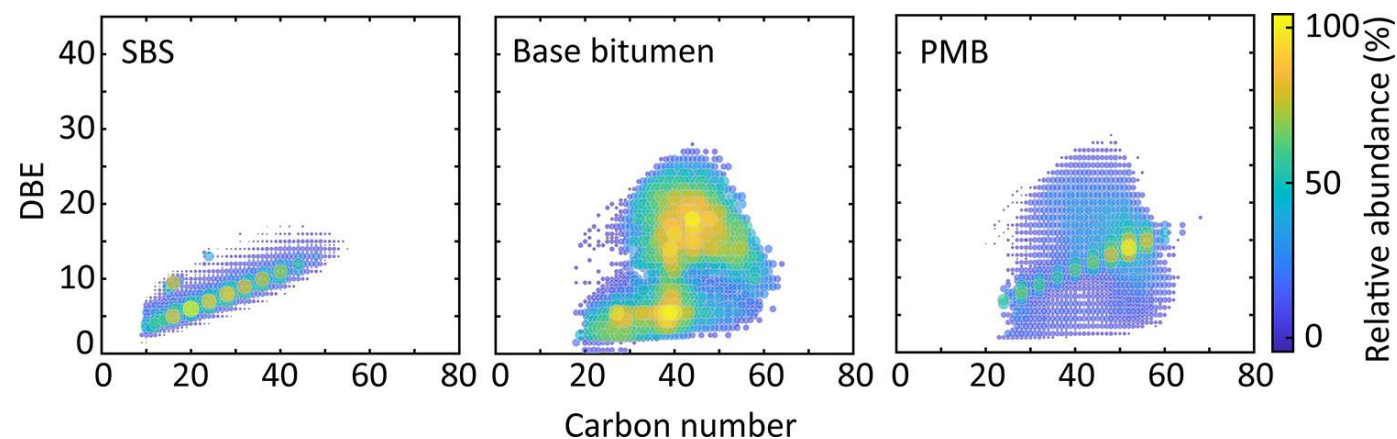
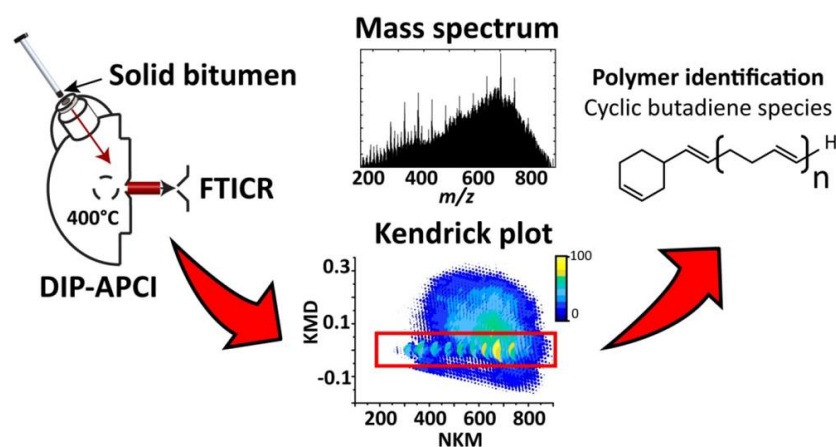
Article

## Direct Insertion Analysis of Polymer-Modified Bitumen by Atmospheric Pressure Chemical Ionization Ultrahigh-Resolution Mass Spectrometry

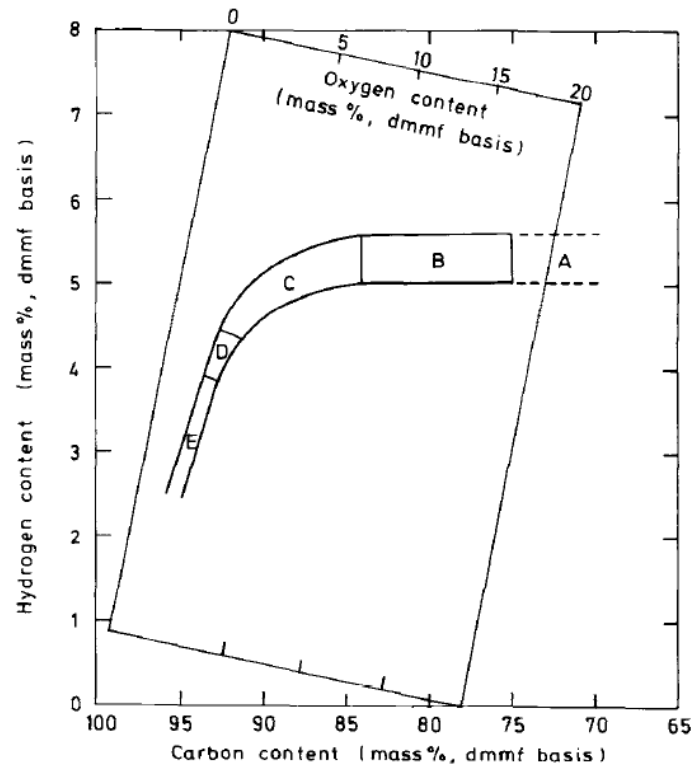
Oscar Lacroix-Andrivet, Clément Castilla, Christopher Rüger, Marie Hubert-Roux, Anna Luiza Mendes Siqueira, Pierre Giusti, and Carlos Afonso\*

Cite This: *Energy Fuels* 2021, 35, 2165–2173

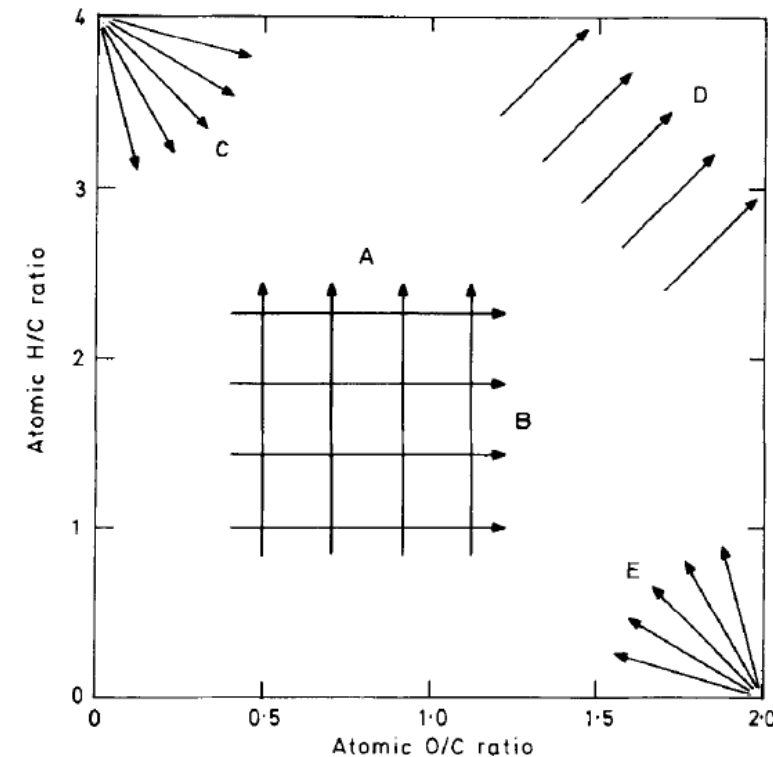
Read Online



## Visualization and grouping of elemental compositions – Van Krevelen Diagram

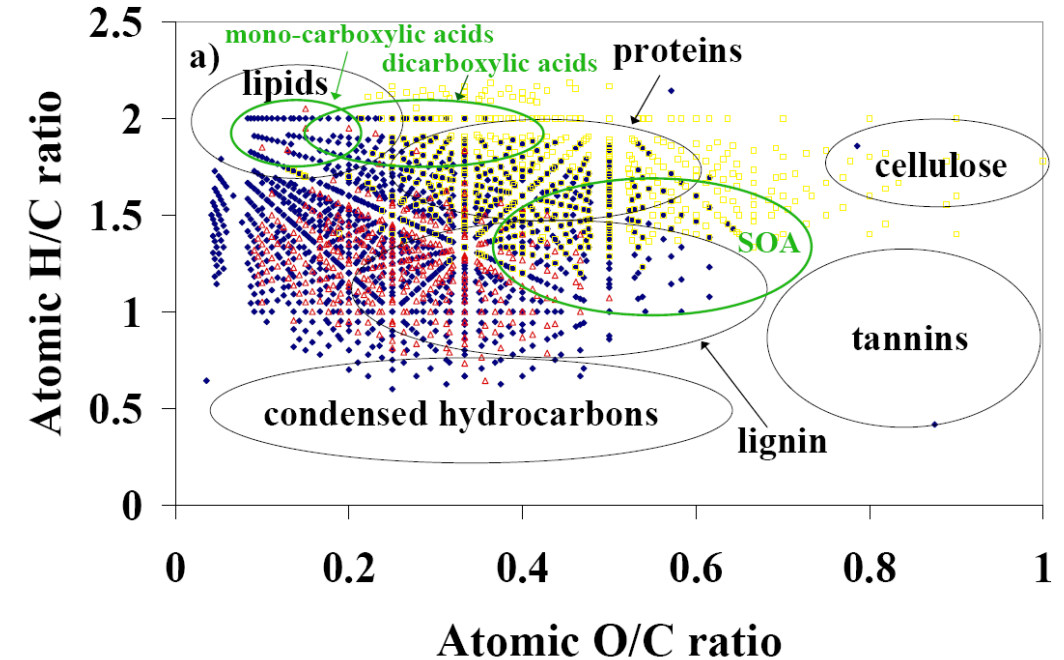
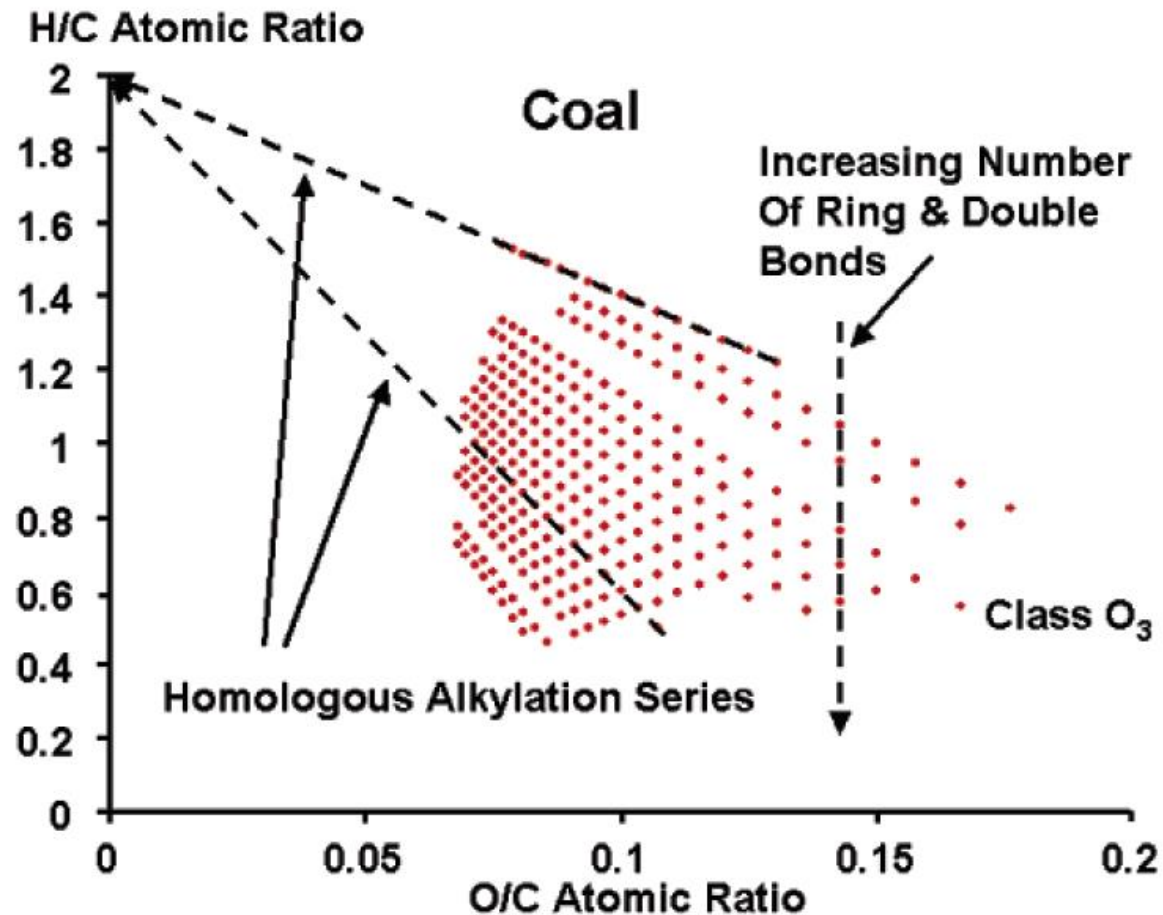


**Figure 1** Seyler's coal chart (adapted from Figure 5.23, Reference 1): A, brown coal and lignite; B, lignituous coal; C, bituminous coal; D, carbonaceous coal; E, anthracite. The broken lines for the brown coal band presumably indicate lack of experience in the classification of such coals in the U.K.



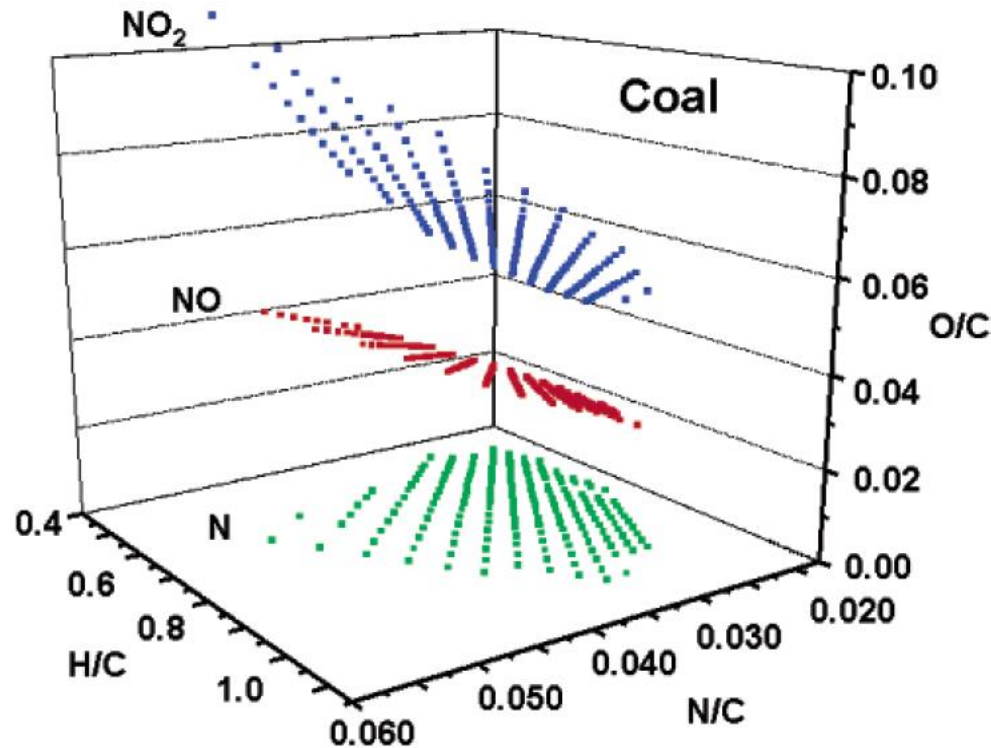
**Figure 2** Van Krevelen chart: A, hydrogenation trajectories; B, oxidation trajectories; C, demethanation trajectories; D, dehydration trajectories; E, decarboxylation trajectories

## Visualization and grouping of elemental compositions – Van Krevelen Diagram

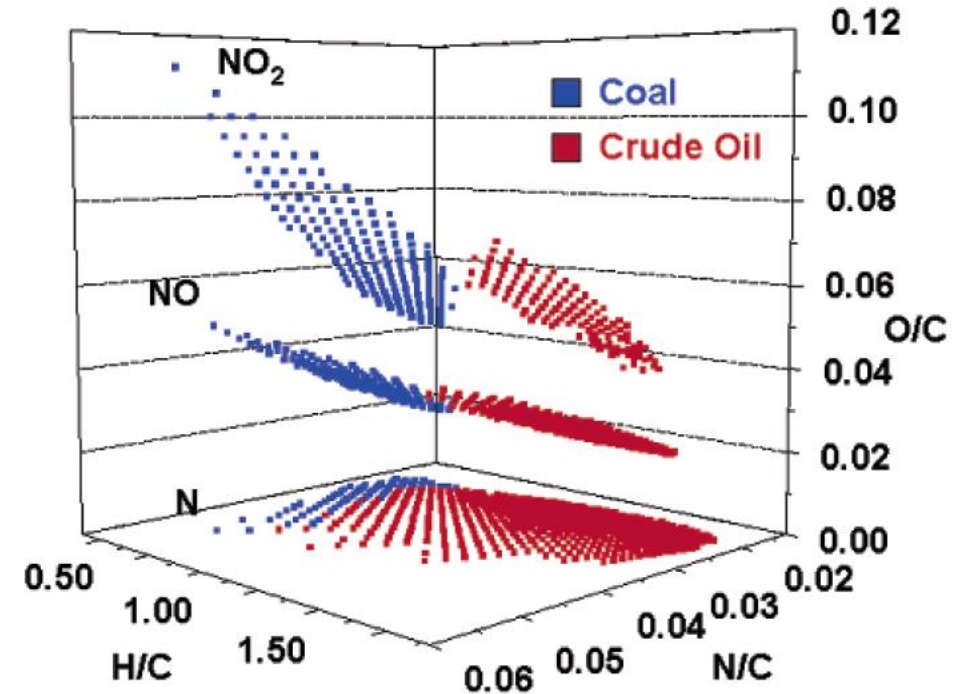


Wozniak et al., ACP (2008)





**Figure 6.** Three-dimensional van Krevelen diagram for members of the classes, N, NO, and NO<sub>2</sub>, from pyridine-extracted Pocahontas No. 3 coal. Each class differs by at least one heteroatom (by definition) and is thus shifted to a different plane. Different classes are thus completely separated in the three-dimensional display.



**Figure 7.** Three-dimensional van Krevelen diagram for the *same* classes (N, NO, NO<sub>2</sub>) for two *different* fossil fuels: coal (blue) and crude oil (red). Because the coal components are more aromatic than are the constituents of crude oil, the two fuels are readily distinguished graphically in the diagram.

Pocahontas #3 Coal



Illinois #6 Coal

## Green Chemistry

### PAPER



Cite this: *Green Chem.*, 2021, 23, 8949

### Unlocking the potential of biofuels *via* reaction pathways in van Krevelen diagrams†

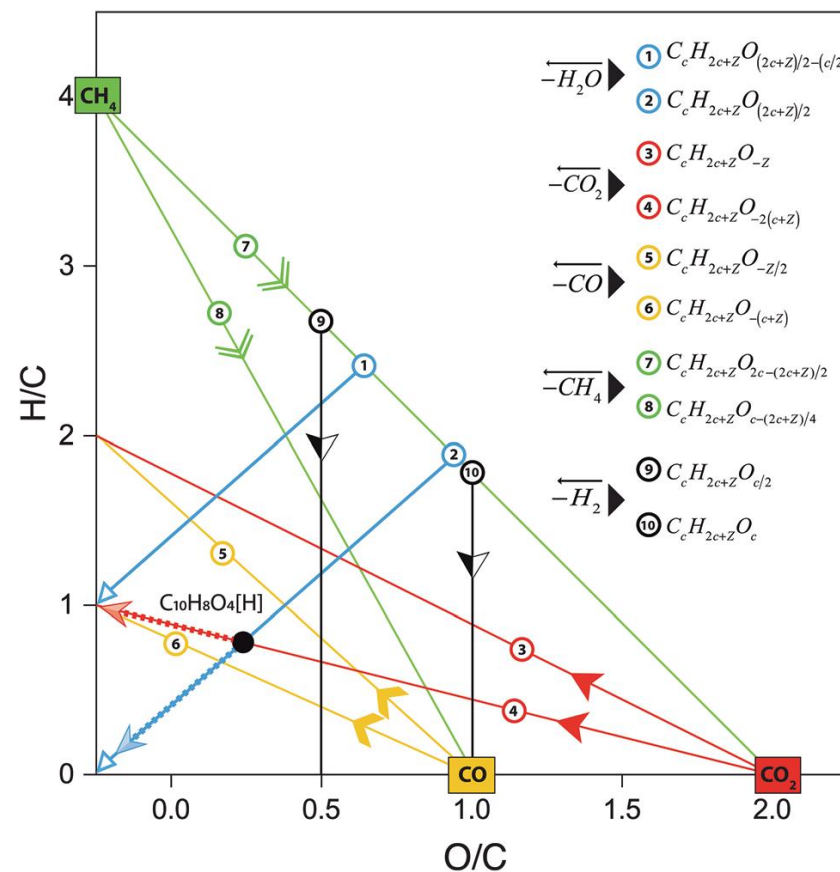
Diana Catalina Palacio Lozano, <sup>id</sup> <sup>a</sup> Hugh E. Jones, <sup>id</sup> <sup>a,b</sup> Tomas Ramirez Reina, <sup>id</sup> <sup>c</sup> Roberto Volpe <sup>id</sup> <sup>d</sup> and Mark P. Barrow <sup>id</sup> <sup>\*a</sup>

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**Table 2** General equations of the magnitude of the displacement along the axis H/C and O/C corresponding to losses of molecules of H<sub>2</sub>O, H<sub>2</sub>, CH<sub>4</sub>, CO, and CO<sub>2</sub> of a molecule with a general chemical formula C<sub>c</sub>H<sub>2n+Z</sub>O<sub>o</sub>

Principal reactions	Reaction vector (c,h,o)	Δ(O/C)	Δ(H/C)
Dehydration (−H <sub>2</sub> O)	(0,−2,−1)	$-\frac{1}{c}$	$-\frac{2}{c}$
Dehydrogenation (−H <sub>2</sub> )	(0,−2,0)	0	$-\frac{2}{c}$
Decarbonylation (−CO)	(−1,0,−1)	$\frac{1}{c-1} \left( \frac{o}{c} - 1 \right)$	$\frac{1}{c-1} \left( 2 + \frac{Z}{c} \right)$
Decarboxylation (−CO <sub>2</sub> )	(−1,0,−2)	$\frac{1}{c-1} \left( \frac{o}{c} - 2 \right)$	$\frac{1}{c-1} \left( 2 + \frac{Z}{c} \right)$
Demethanation (−CH <sub>4</sub> )	(−1,−4,0)	$\frac{1}{c-1} \left( \frac{o}{c} \right)$	$\frac{1}{c-1} \left[ \left( 2 + \frac{Z}{c} \right) - 4 \right]$

## Handling Attribution Data



Example of reaction processing lines in van Krevelen diagrams corresponding to losses of molecules of water, carbon dioxide, carbon monoxide, methane, and hydrogen. The characteristic molecular formulae of molecules in each line are shown on the right corner and the CO, CO<sub>2</sub>, and CH<sub>4</sub> poles are shown along the axis, 2c + Z correspond to the total hydrogen atoms within the molecule.

## Green Chemistry








### PAPER

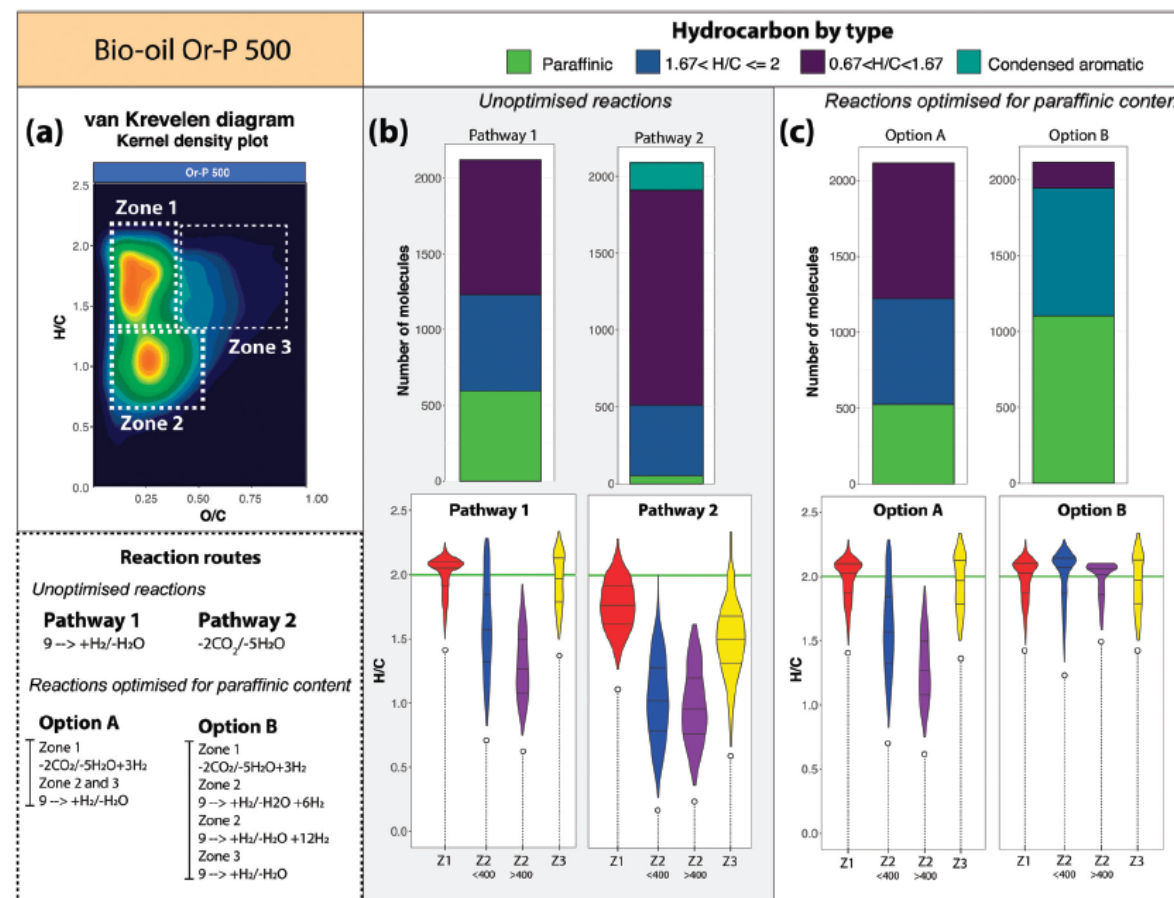
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### Unlocking the potential of biofuels via reaction pathways in van Krevelen diagrams†

Cite this: *Green Chem.*, 2021, 23, 8949

Diana Catalina Palacio Lozano, <sup>a</sup> Hugh E. Jones, <sup>a,b</sup> Tomas Ramirez Reina, <sup>c</sup> Roberto Volpe <sup>d</sup> and Mark P. Barrow <sup>\*a</sup>



**Fig. 7** Comparison of the starting material and the hydrocarbon distribution by type obtained after the simulation of different reaction pathways. (a) van Krevelen diagrams of the oxygenated species of the bio-oil obtained from orange pulp plotted as Kernel density plot. (b) and (c) – Distributions of the hydrocarbons obtained by the simulation of unoptimised reactions and reactions optimised for paraffinic content, respectively. Violin plots are used to illustrate the distribution of the hydrocarbons in the H/C-value by zone. The reactions were applied to four different fractions of the bio-oil Or-P 500. Zone 1 (Z1)  $1.5 \leq \text{H/C} < 2$ ,  $0 < \text{O/C} \leq 0.3$  (652 molecules), zone 2 (Z2)  $0.5 < \text{H/C} < 1.5$ ,  $0 < \text{O/C} < 0.67$  (1283 molecules), and zone 3 (Z3)  $1.5 < \text{H/C} < 2$ ,  $\text{O/C} > 0.3$  (402), the fraction of molecules in zone 2 was also divided by mass: <400 Da and >400 Da (697 and 586 molecules respectively). All the reactions were performed to reduce 9 oxygen atoms from the molecular formulae. About 2100 hydrocarbon molecules were obtained after the deoxidation reactions were applied.



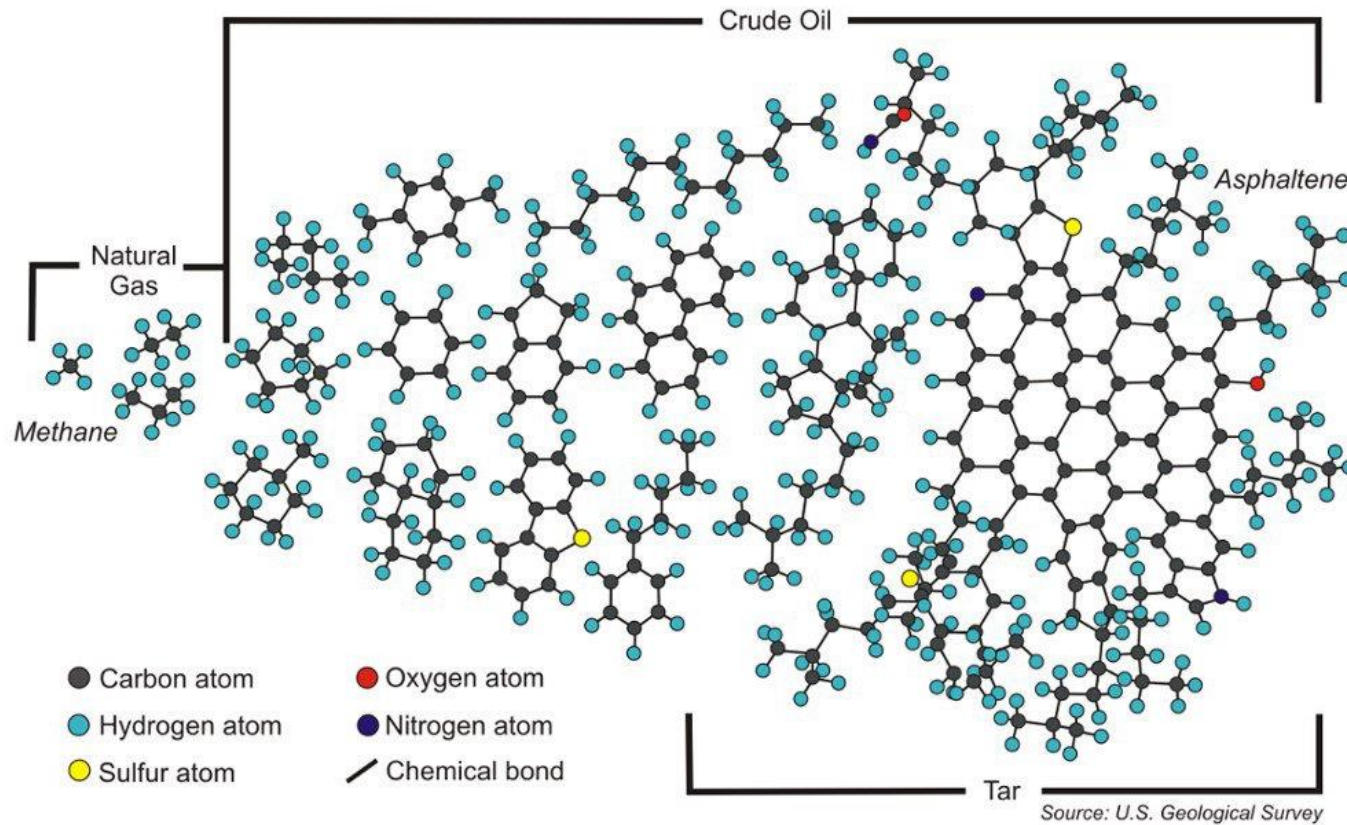
## Hyphenated Solutions – here Thermal Analysis Coupled to High-Resolution Mass Spectrometry

→ time-dependent processing of the temperature-resolved complex mass spectra needed applying the shown concept and steps for data mining

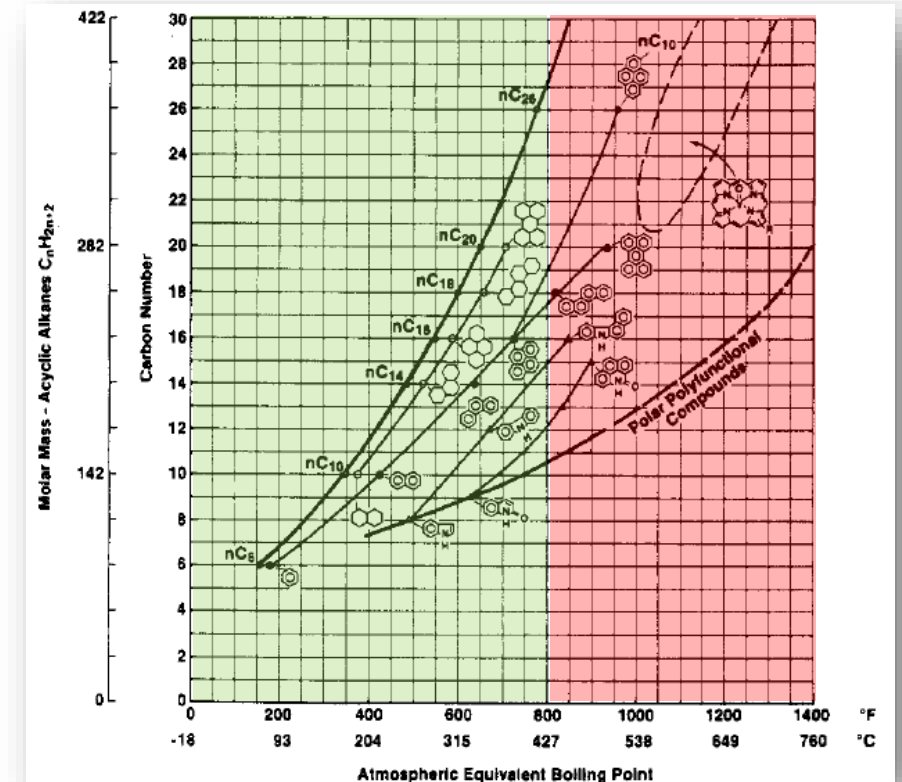
- **Aging of Bitumen** – Combining FT-ICR MS and GCxGC Data
- **Plastic Pyrolysis Chars** – Combination of Several Thermal Analysis Techniques
- **Carbon Fibers from Bitumen Feedstocks** – Cooperation with the 21T platform at NHMFL
- **Orbitrap Thermal Analysis and Online Emission Tracking** – non-ICR but deploying the same concepts

# Methodology - Thermal Analysis

# Complex Mixtures: Heavy Fractions and Residues



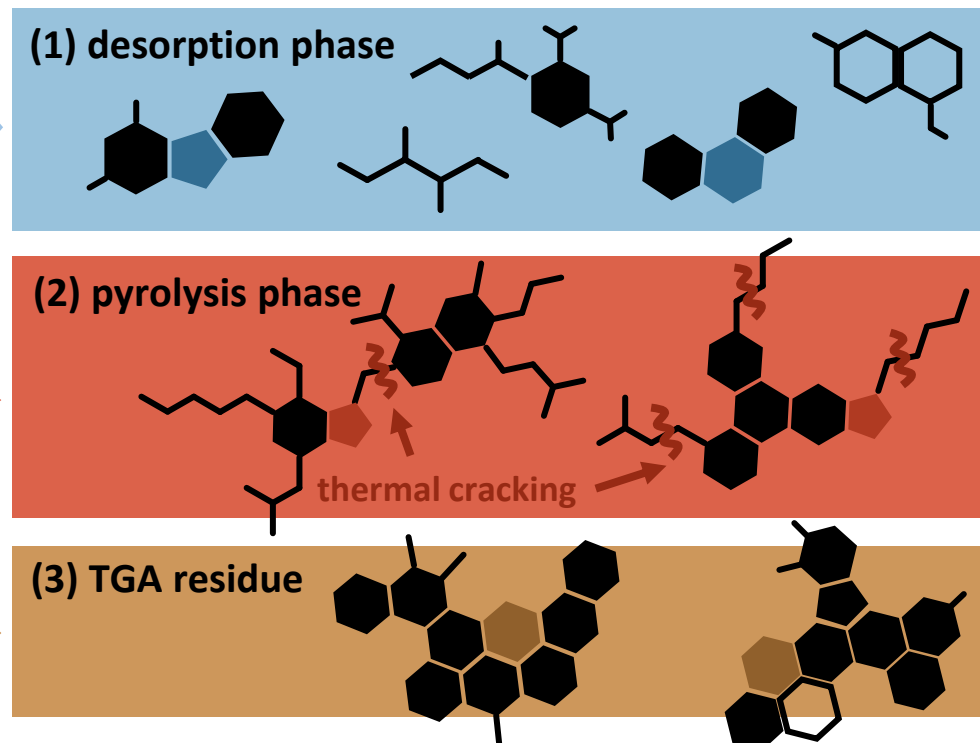
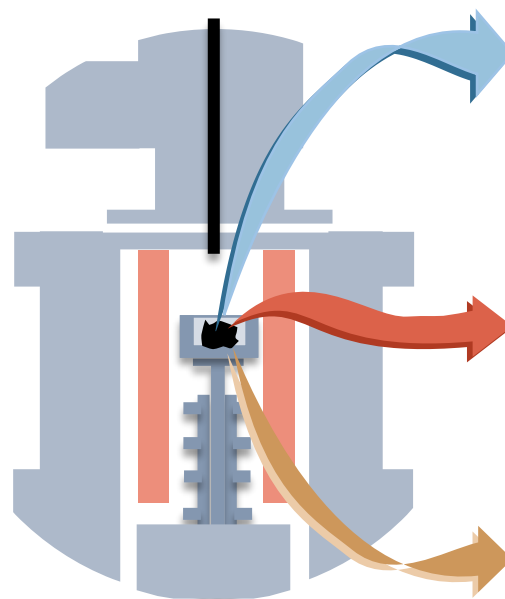
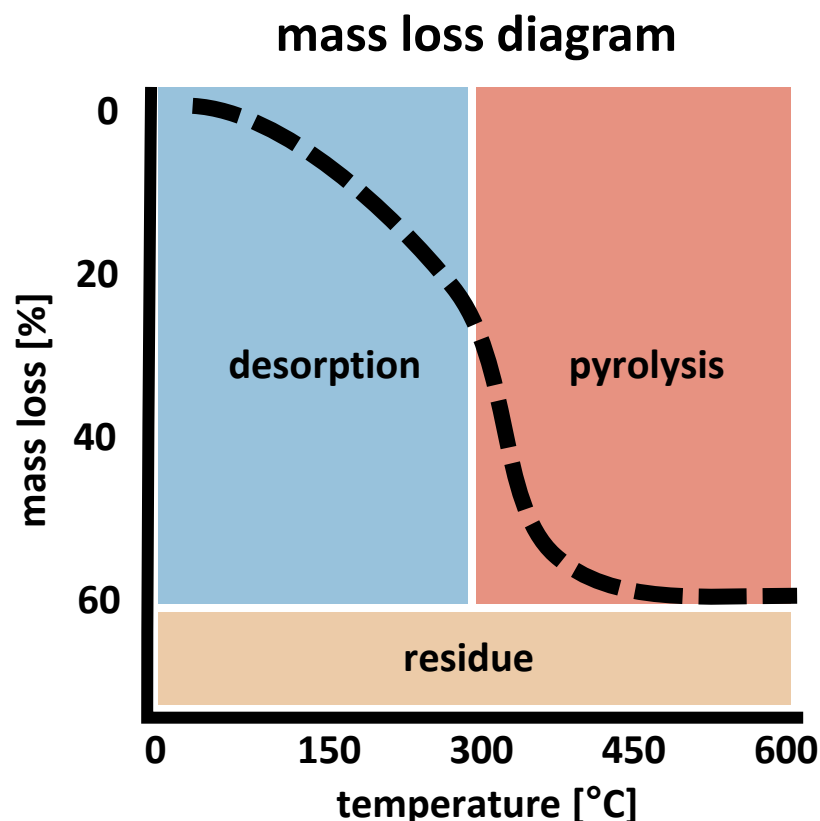
GC/GCxGC-MS accessible



- Petroleum (fossil/biomass/pyrolysis) are ultra complex mixtures → *Petroleomics*

Challenging heavy fractions

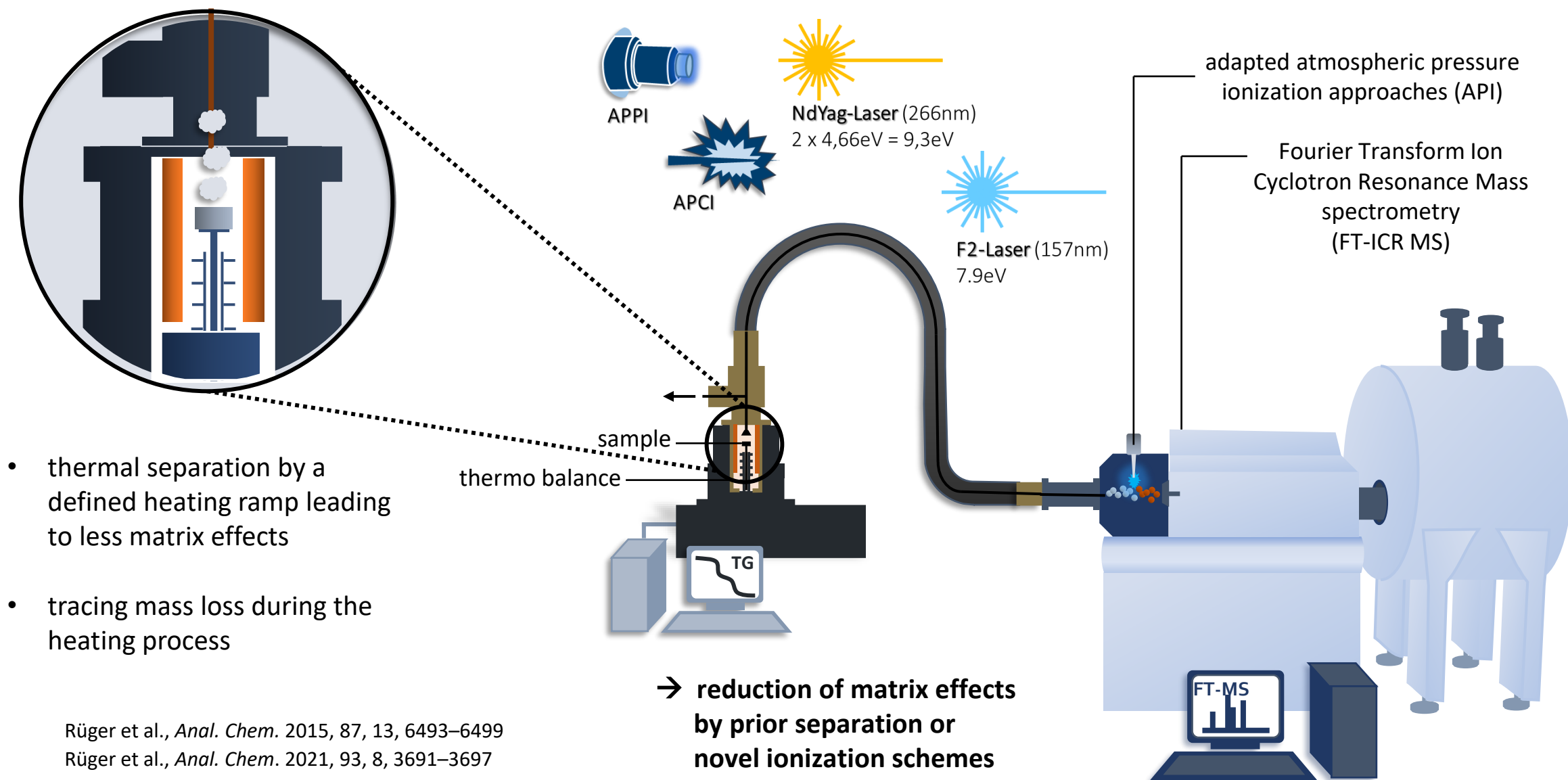




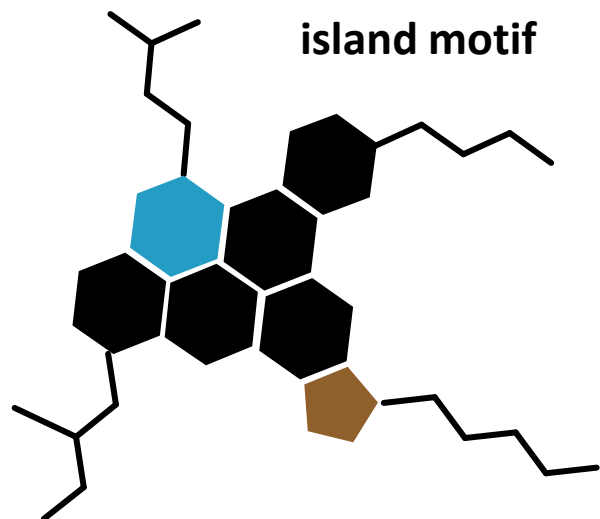
**Three processes occur during the thermogravimetric heating process:**

- (1) Desorption: mostly evaporation of intact compounds
- (2) Pyrolysis: thermal decomposition of high molecular weight species (starting at 300-350 °C)
- (3) Residue: coke formation of high aromatic compounds and highly stable aggregates

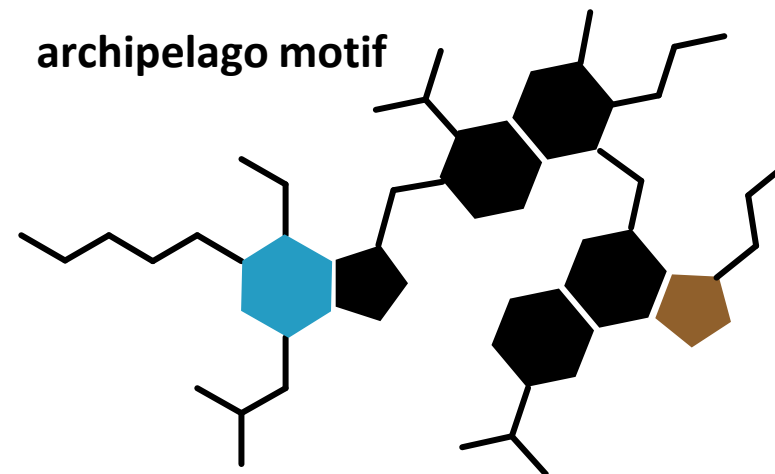
# Mass Spectrometric Platform – FT-ICR MS



## Architecture of asphaltenes strongly discussed in literature



Wyoming Deposit asphaltenes



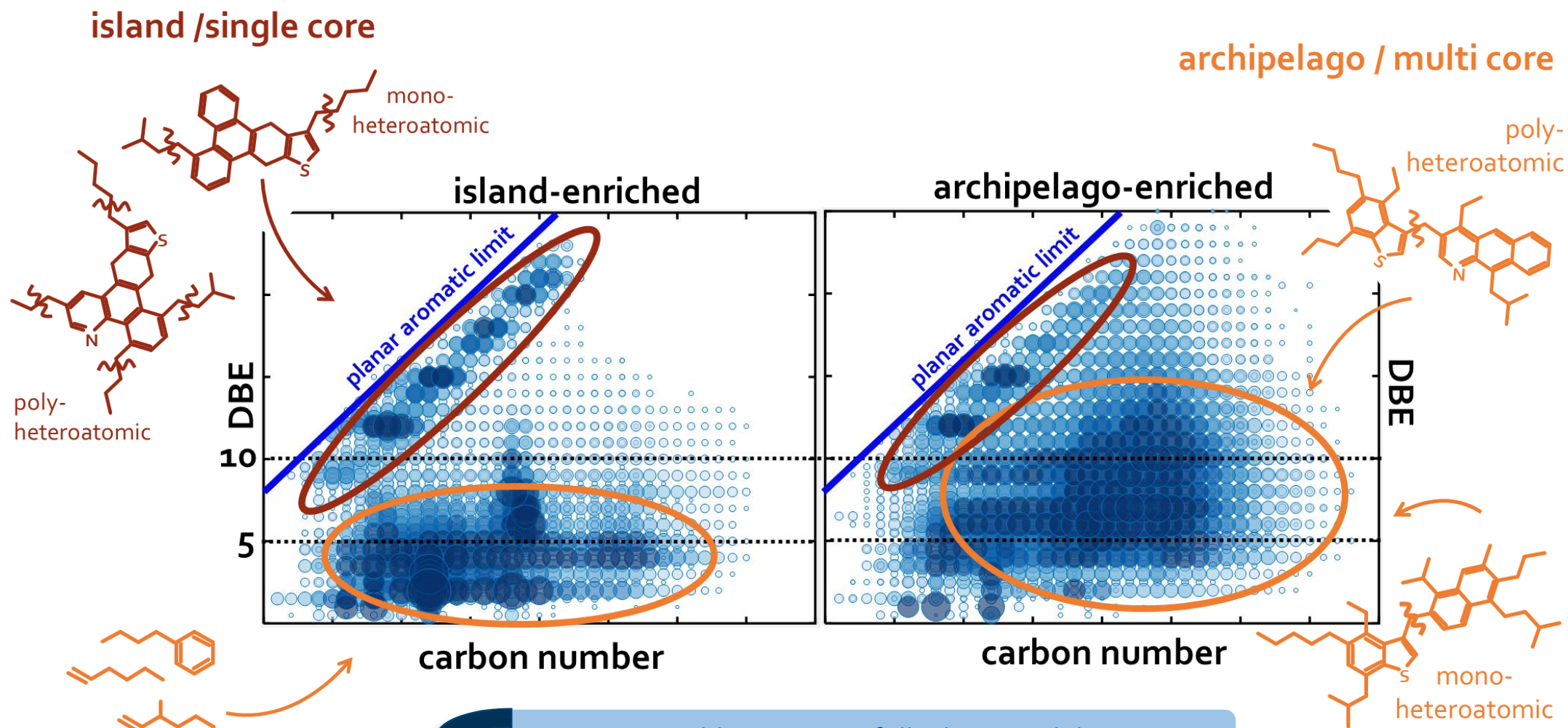
Athabasca Bitumen asphaltenes

Only limited analytical techniques available, which show evidence for both types of asphaltenes.  
→ A structural motif is most often highlighted by a specific analytical technique.

Is TG-FT-ICR MS able to distinguish between both structural motifs?



# Asphaltene Pyrolysis Products

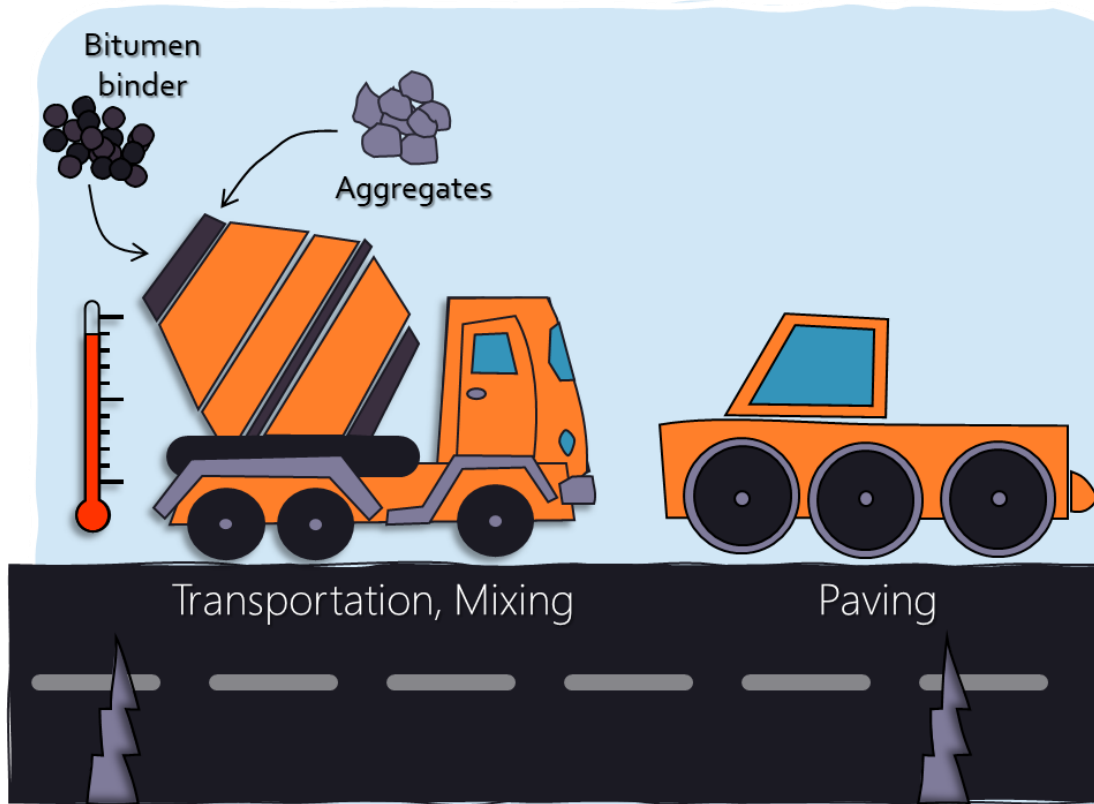


TA-HRMS is able to successfully distinguish between the structural motifs in those ultra-complex mixtures

# Case Study I – Bitumen Aging

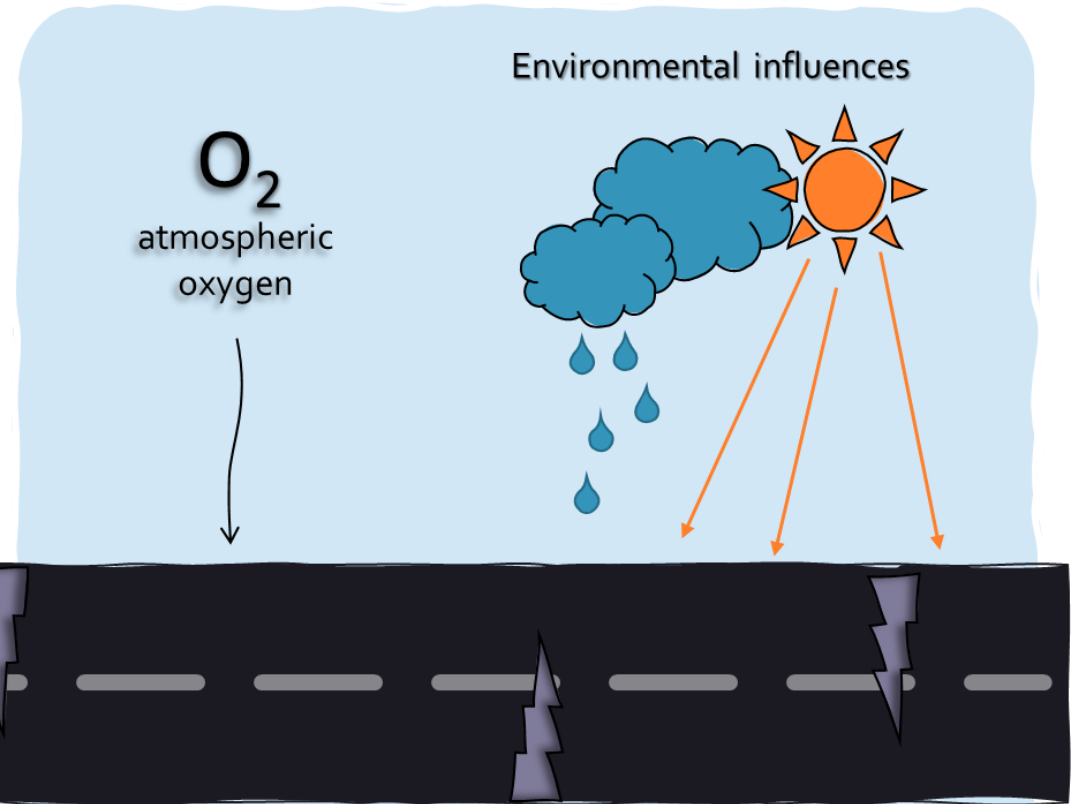
# Bitumen Ageing – Combining FTMS and GCxGC

## SHORT TERM AGING



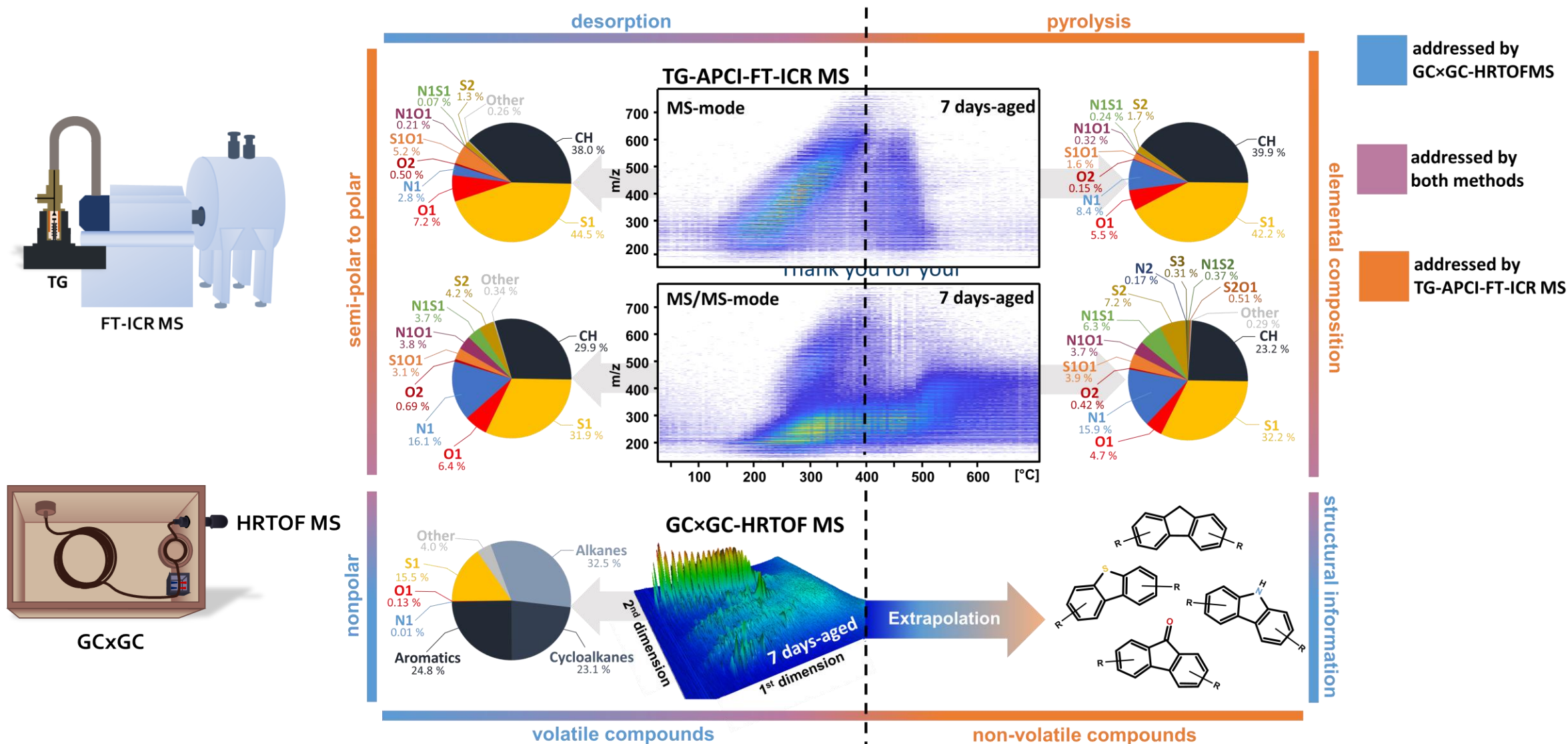
Occurs during **transportation, mixing** and **paving**. It is characterized by **high temperatures** and **high specific surface**.

## LONG TERM AGING



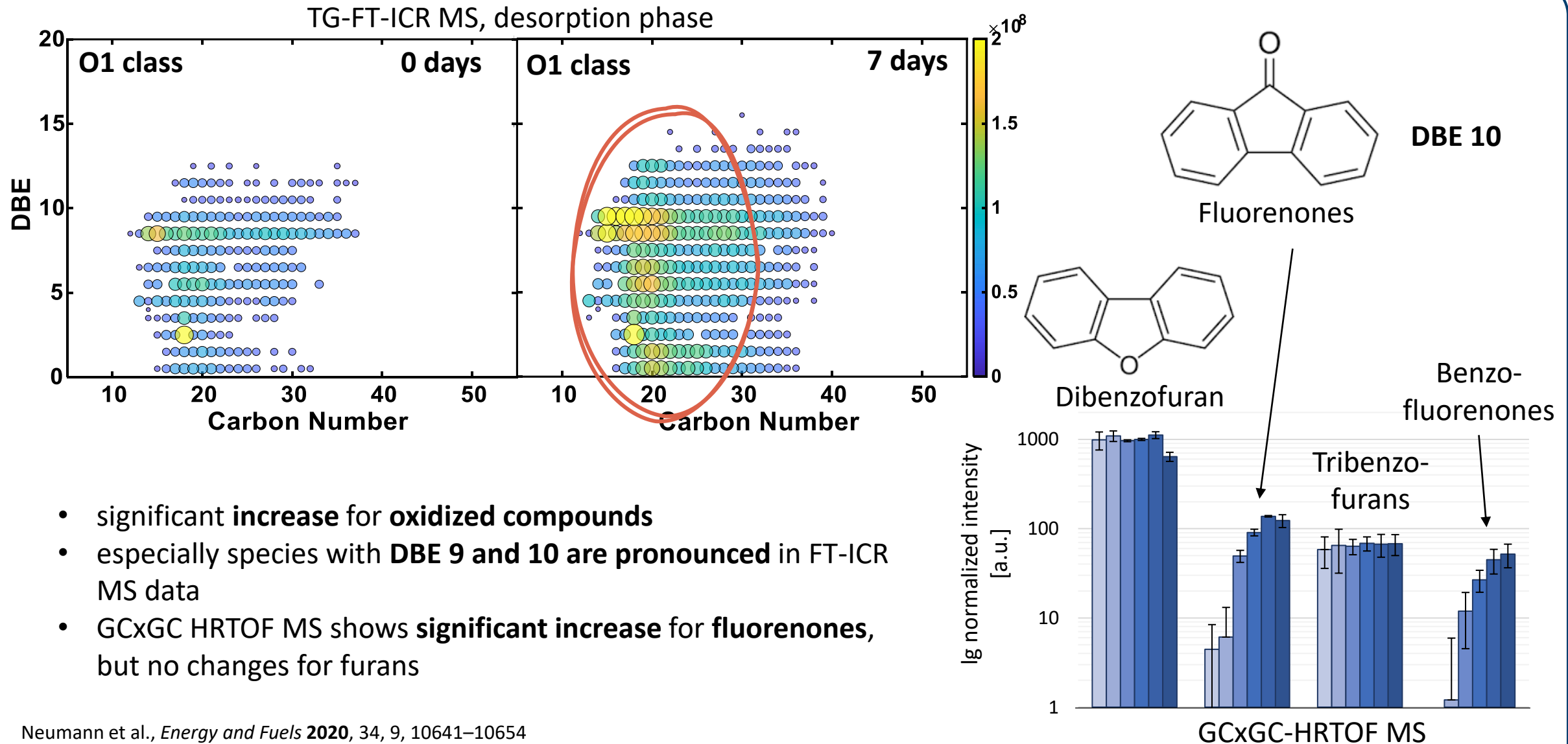
Occurs during the **service time** and is triggered by **climatic conditions, void content, oxidation** and **ultraviolet irradiation**.

# Bitumen Ageing – Combining FTMS and GCxGC

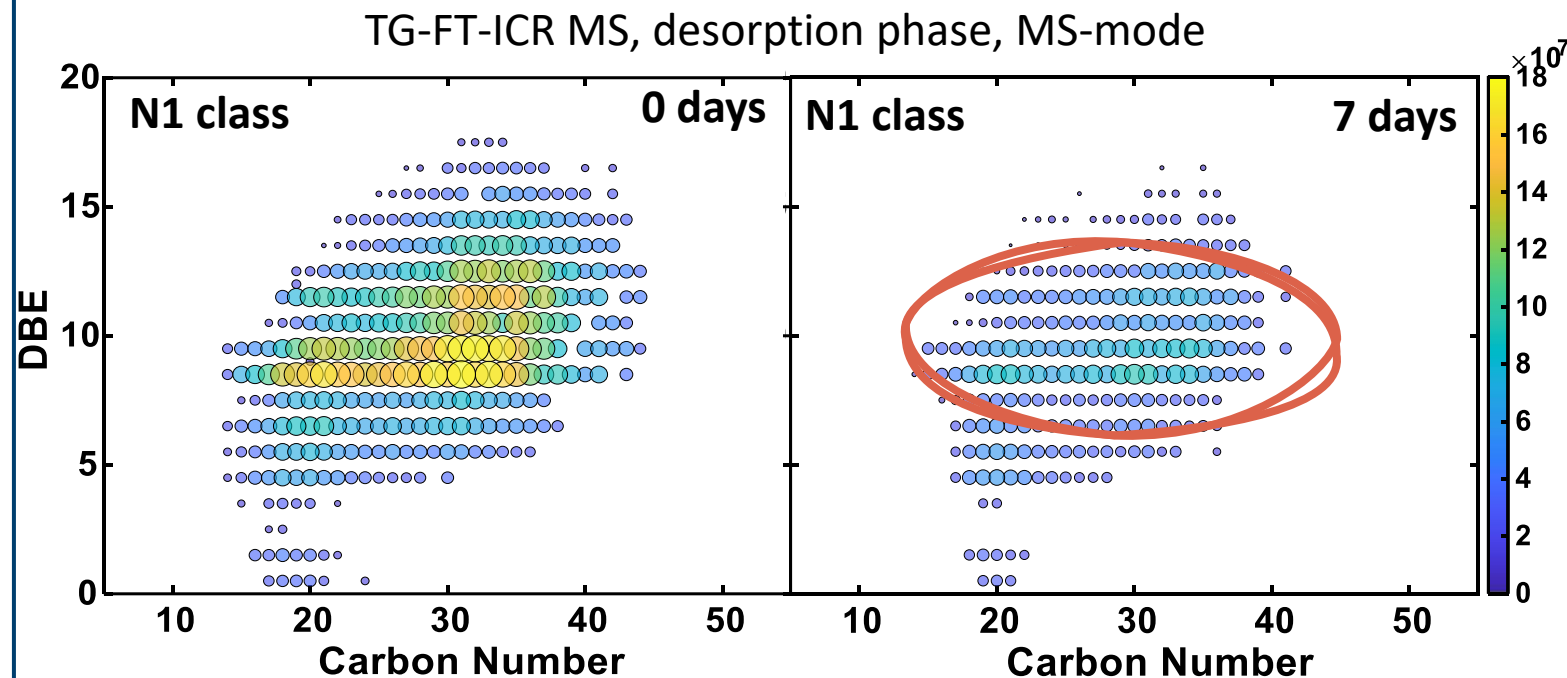


Neumann et al., *Energy and Fuels* **2020**, 34, 9, 10641–10654

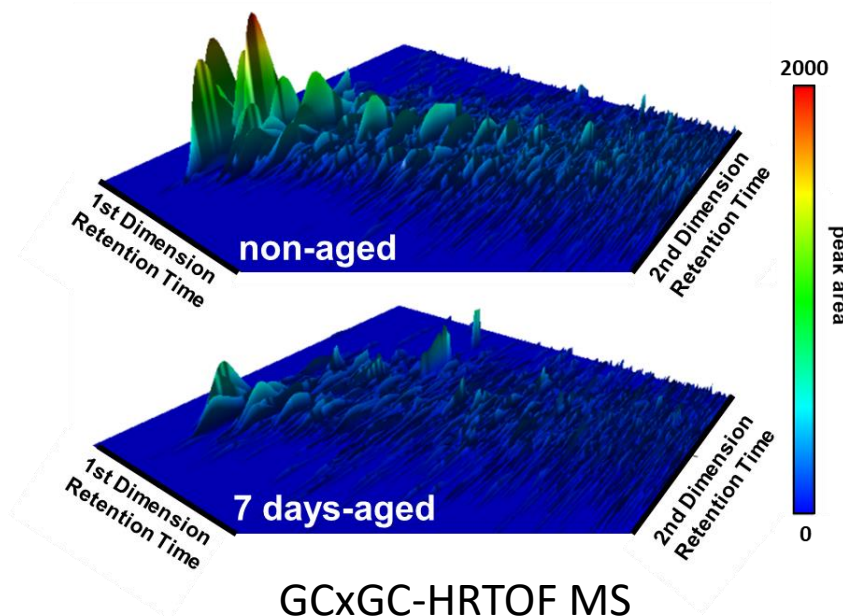
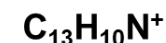
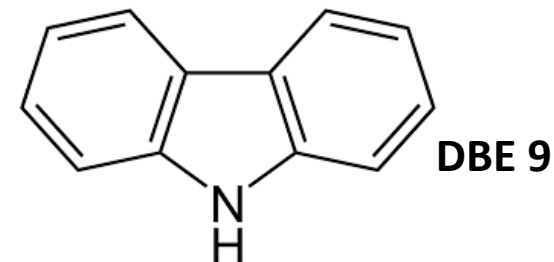




Neumann et al., *Energy and Fuels* **2020**, 34, 9, 10641–10654

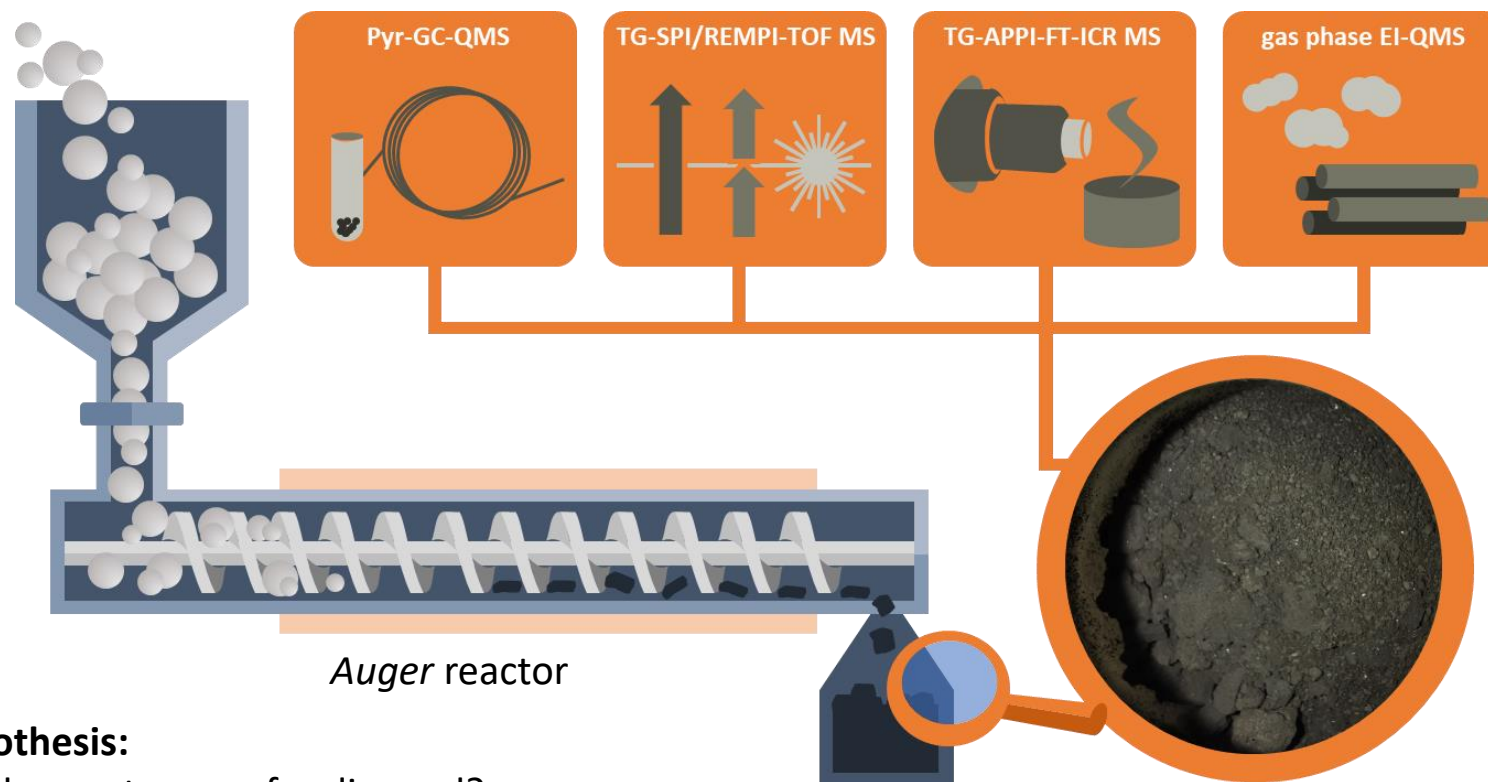


- strong **decrease** for **all N-containing species** revealed by both techniques, especially species with DBE greater than 9 (carbazoles)
- **oxidized N-compounds** are **near detection** limit, but evidence for further oxidation



# Case Study II – Plastic Pyrolysis Chars

## Chemical description of plastic pyrolysis coke residues for improved recycling

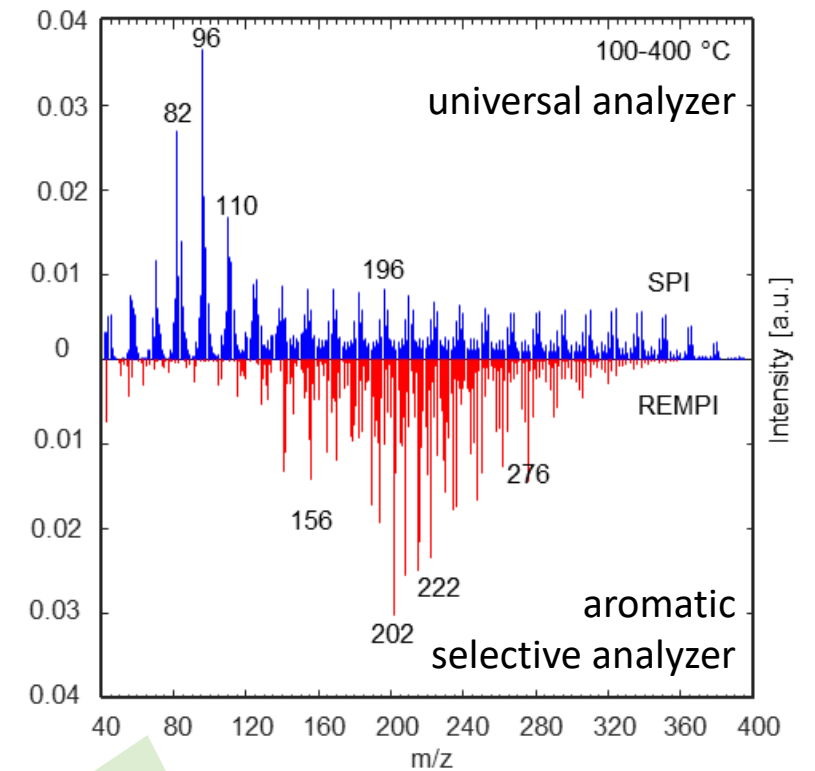
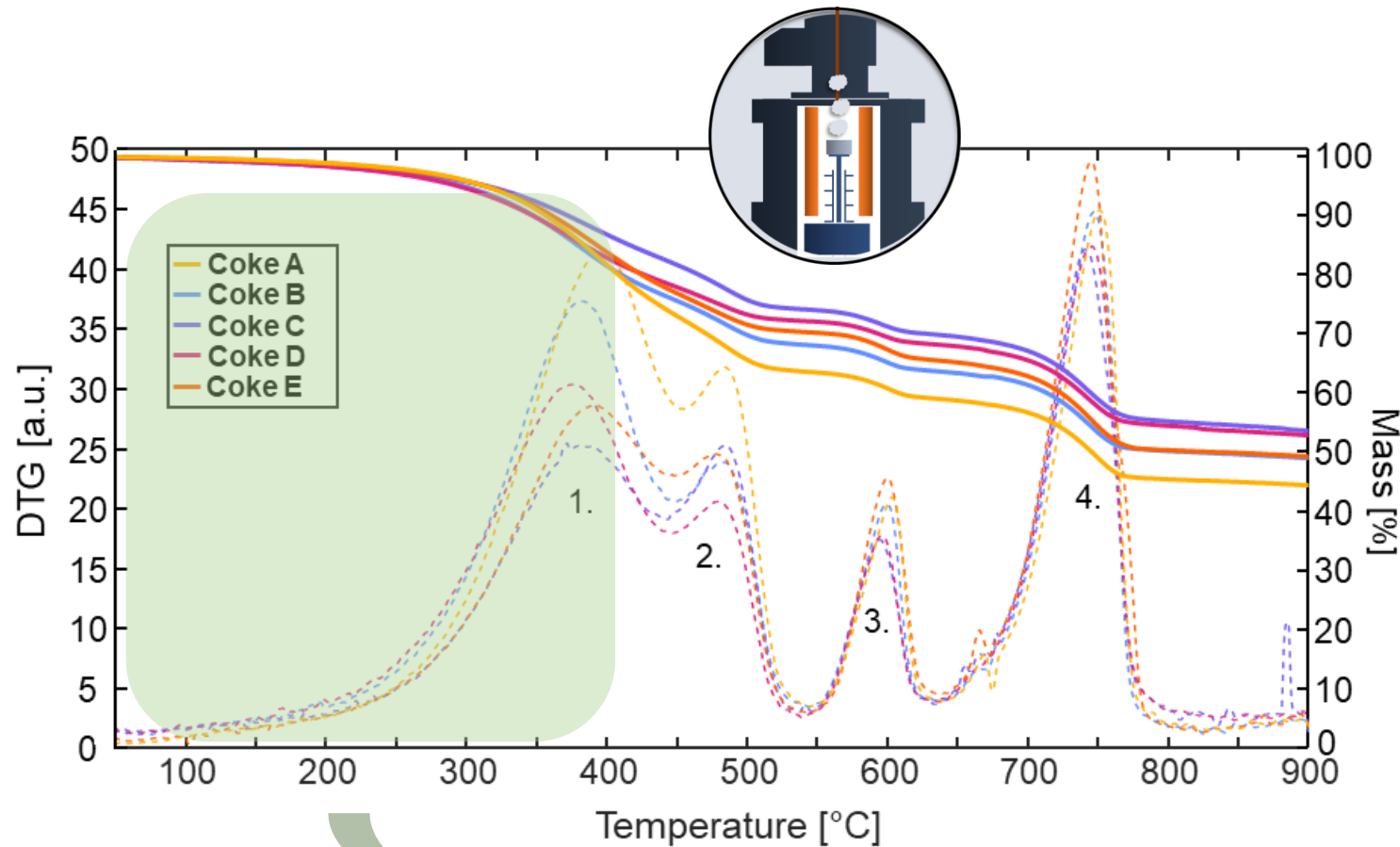


### Main research hypothesis:

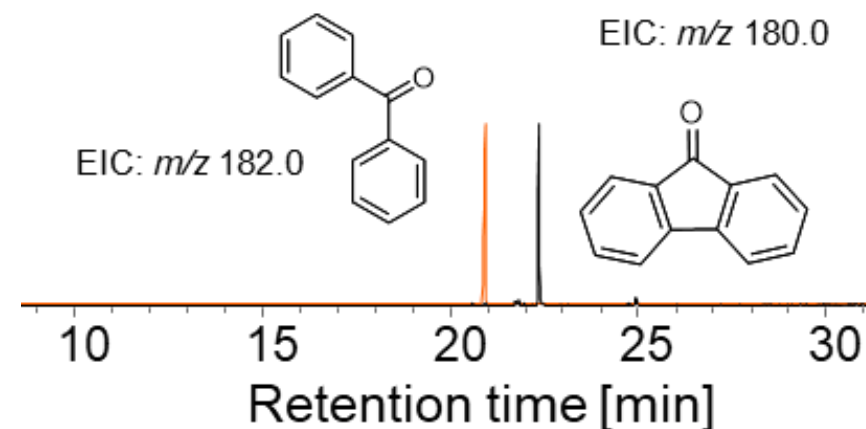
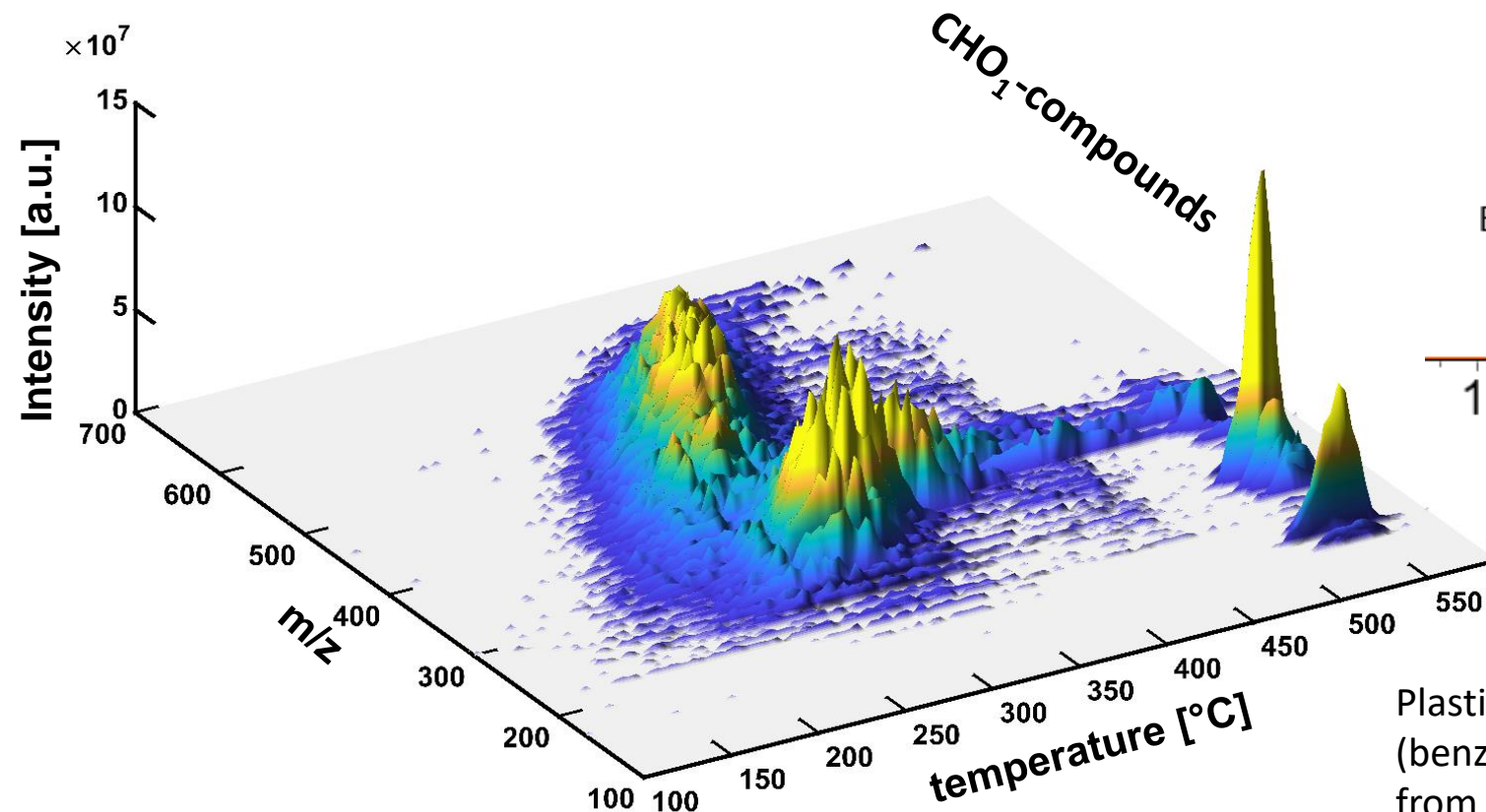
- Can we reduce the waste mass for disposal?
- Can we examine the quality and purity/toxicity of the coke?
- Are valuable chemicals be accessible based on a second pyrolysis step?
- Is there a potential for usage of the residue in material science?



Unique thermogravimetric behavior with several distinct mass loss events



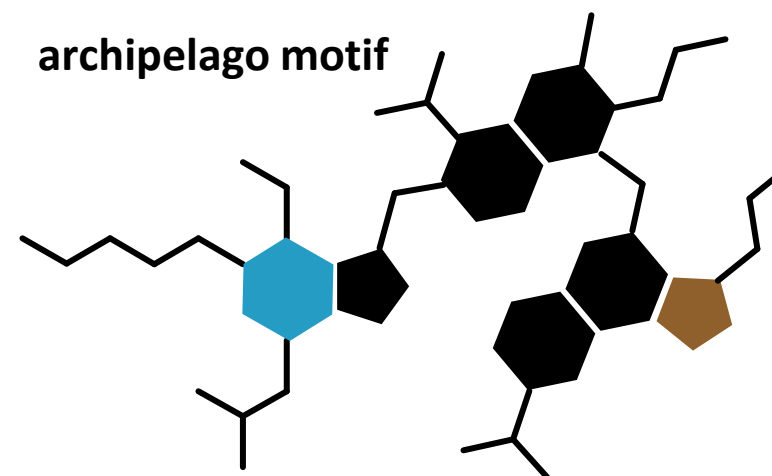
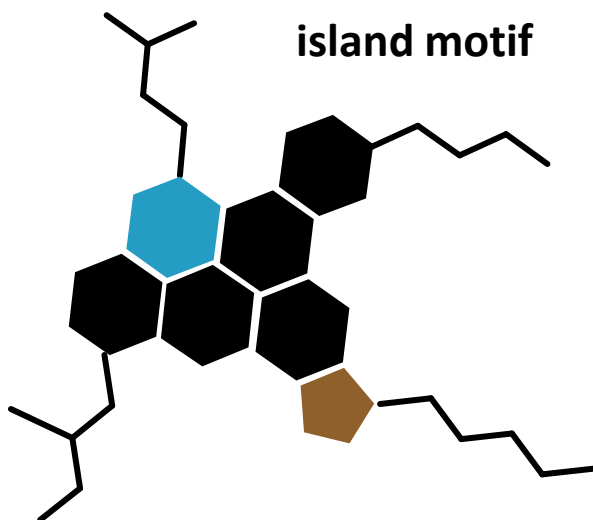
## Polymer additives and plasticizers cause unique pyrolysis chemistry



Survey view of the thermal analysis high-resolution mass spectrometric data

Plasticizer derivatives, such as UV-stabilizers (benzophenones) released at elevated temperatures from the char network (strongly bound and/or integrated in the macromolecular structure).

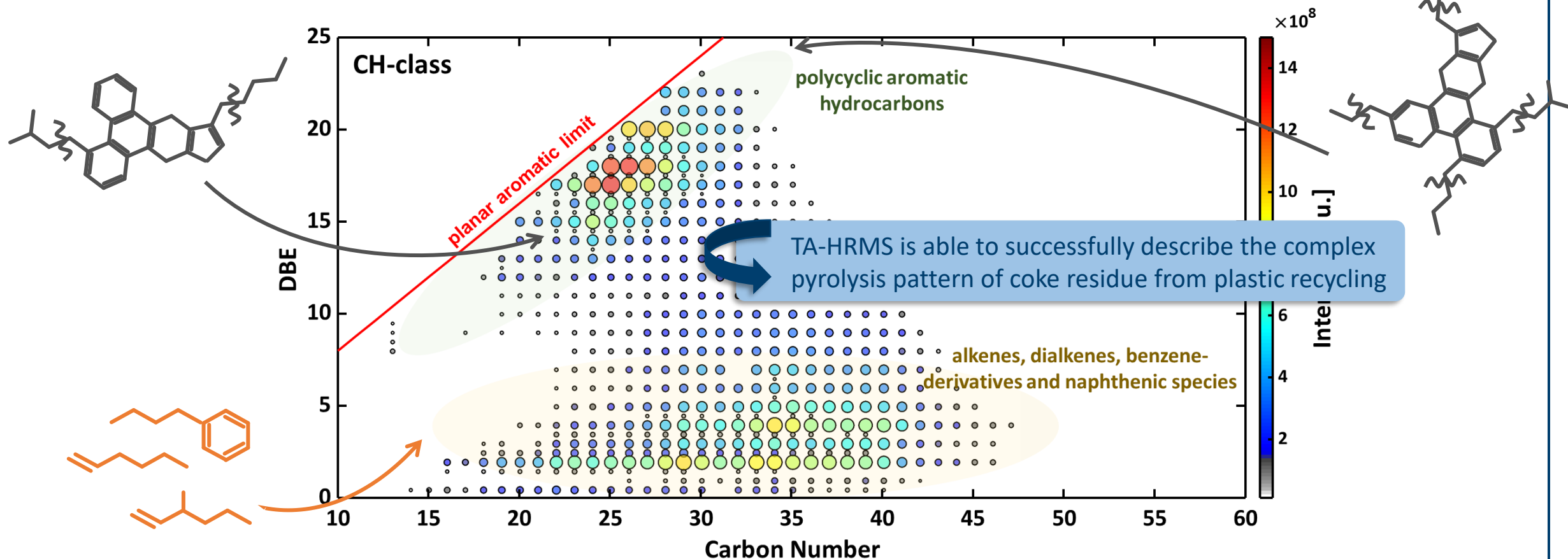
Architecture of graphitized materials strongly discussed in literature...



Is TG-FT-ICR MS able to distinguish between both structural motifs?

Neumann *et. al.*, *Energy Fuels* 2021, 35, 3808-3824; DOI: 10.1021/acs.energyfuels.0c03751

## Island/Archipelago-type structural motives with low heteroatom count



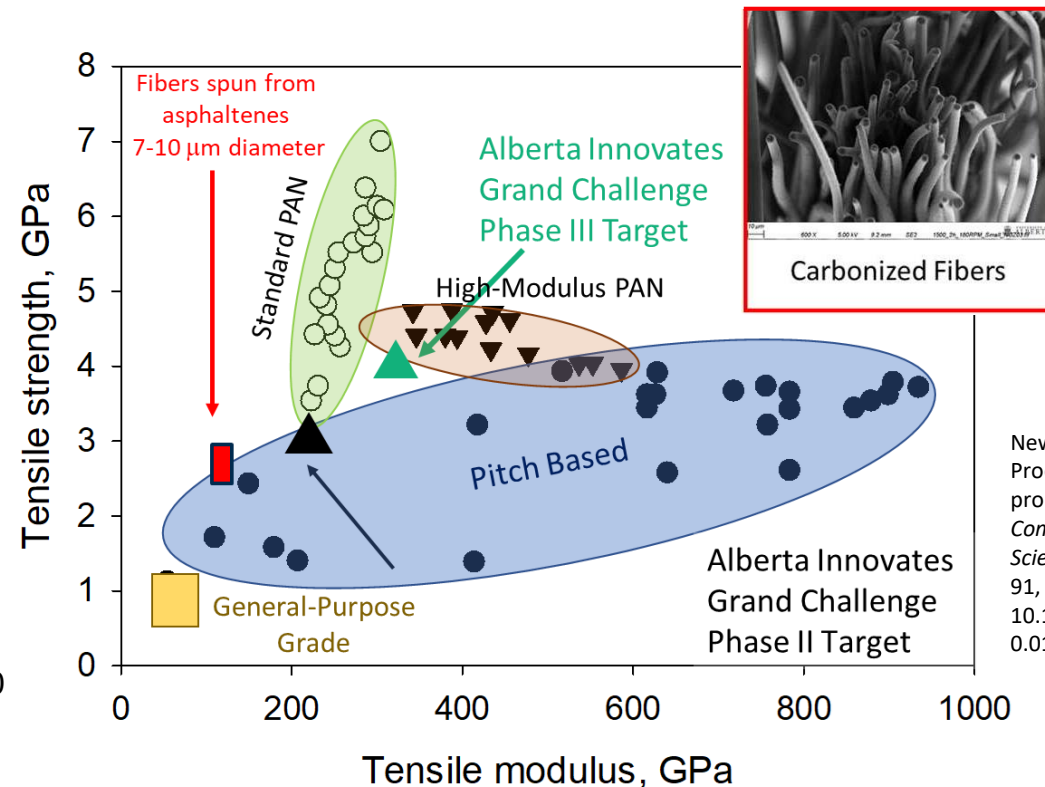
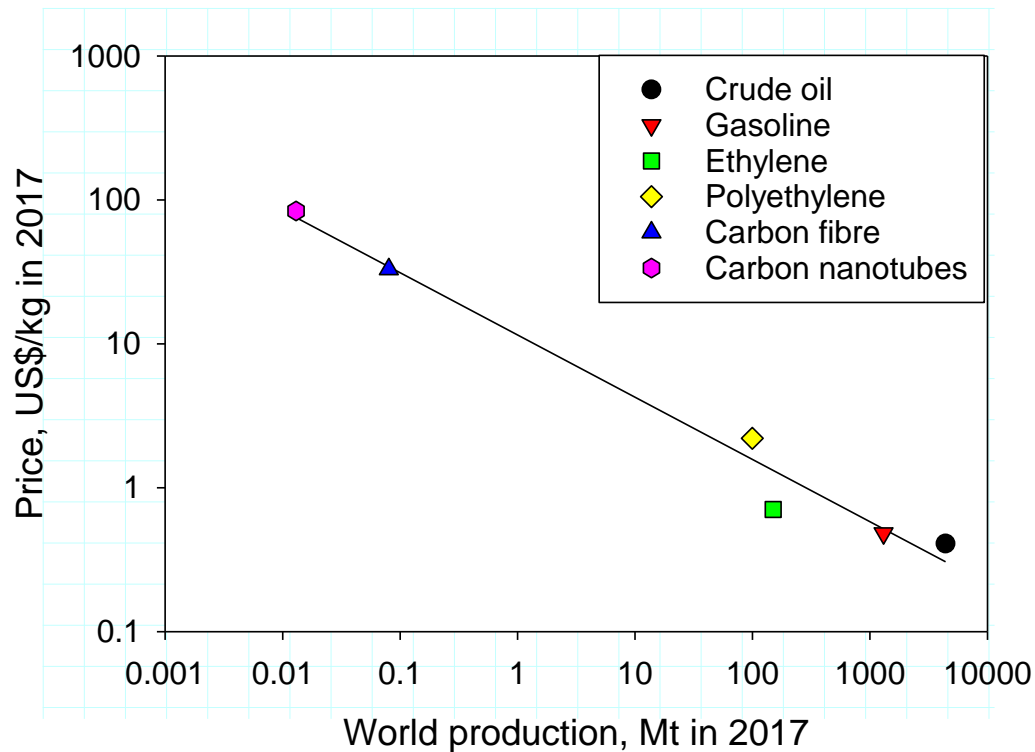
Comprehensive Chemical Description of Pyrolysis Chars from Low-Density Polyethylene by Thermal Analysis Hyphenated to Different Mass Spectrometric Approaches, Friederici et al., *Energy and Fuels*, 2021, DOI 10.1021/acs.energyfuels.1c01994



# Case Study III – Bitumen Carbon Fibers

# Carbon Fibers from Bitumen Feedstocks

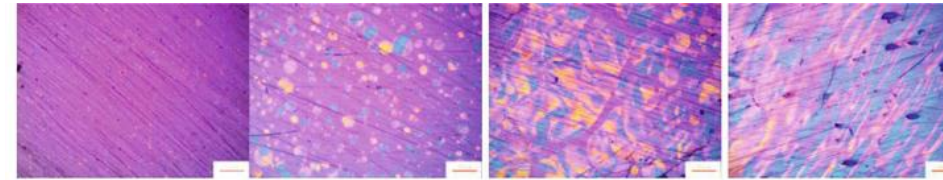
- production of asphalt binder, activated carbon and **carbon fibers** from asphaltene-rich bitumen (15-30 wt%)
- carbon fibers are widely used in applications ranging from aerospace to **blades for wind turbines**
- **light weight, corrosion resistance, and conductivity**
- increasing demand for carbon fibers (58k tons in 2015 to more than 100k tons in 2020)
- PAN serves as the principal precursor (96%) => greenhouse gas emissions for PAN and fiber production



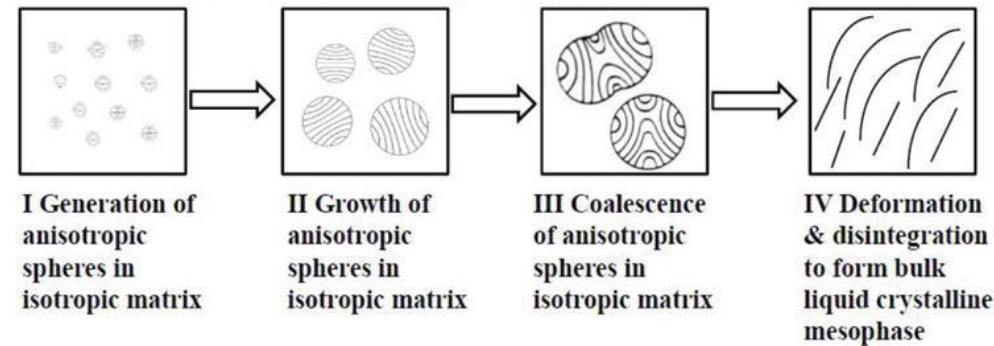
Newcomb, Bradley A. (2016): Processing, structure, and properties of carbon fibers. In: *Composites Part A: Applied Science and Manufacturing* 91, S. 262–282. DOI: 10.1016/j.compositesa.2016.10.018.

# Carbon Fibers from Bitumen Feedstocks

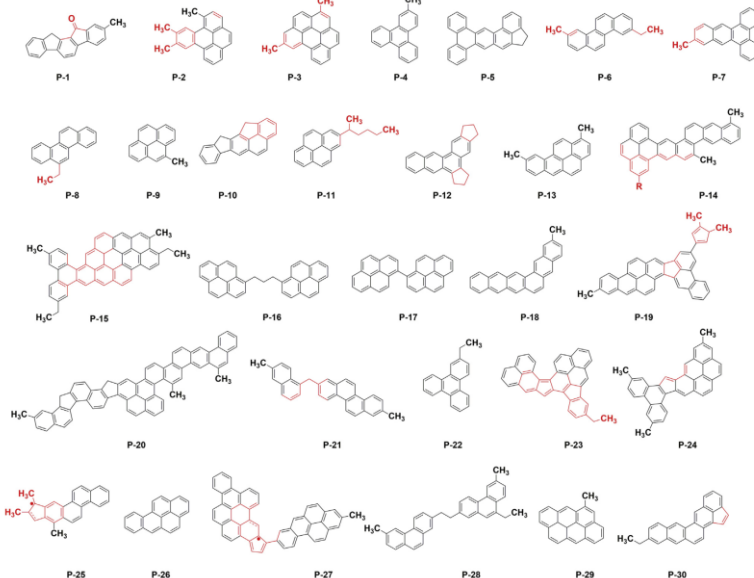
- large aromatics separate to form a liquid crystal phase
- stacking of aromatics detected by diffraction of light
- liquid crystal phase is called mesophase



Sasani Ghamsari, Morteza; Carlescu, Irina (Hg.) (2020): Liquid Crystals and Display Technology: IntechOpen.



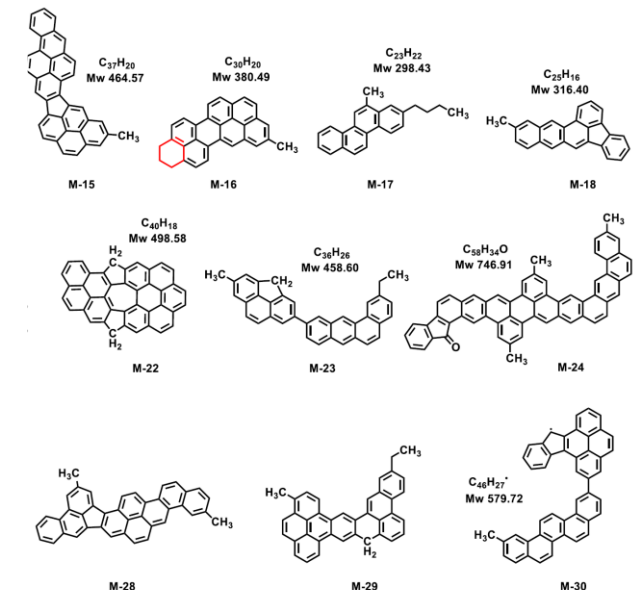
## Isotropic Pitch



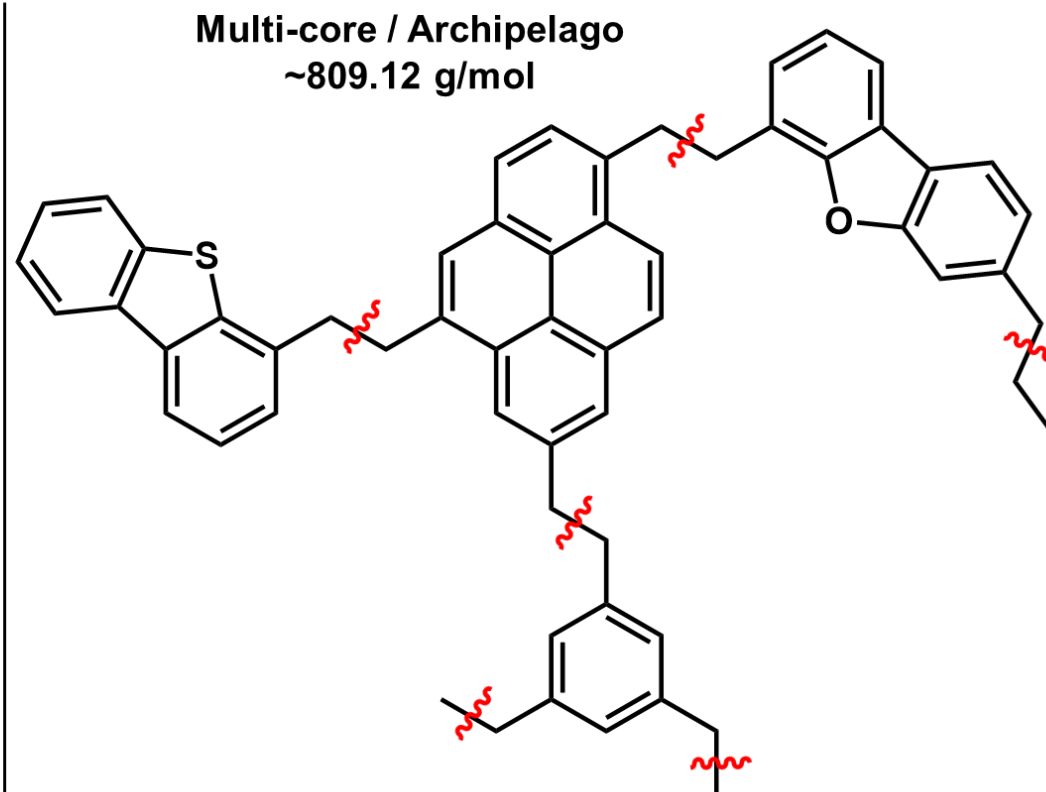
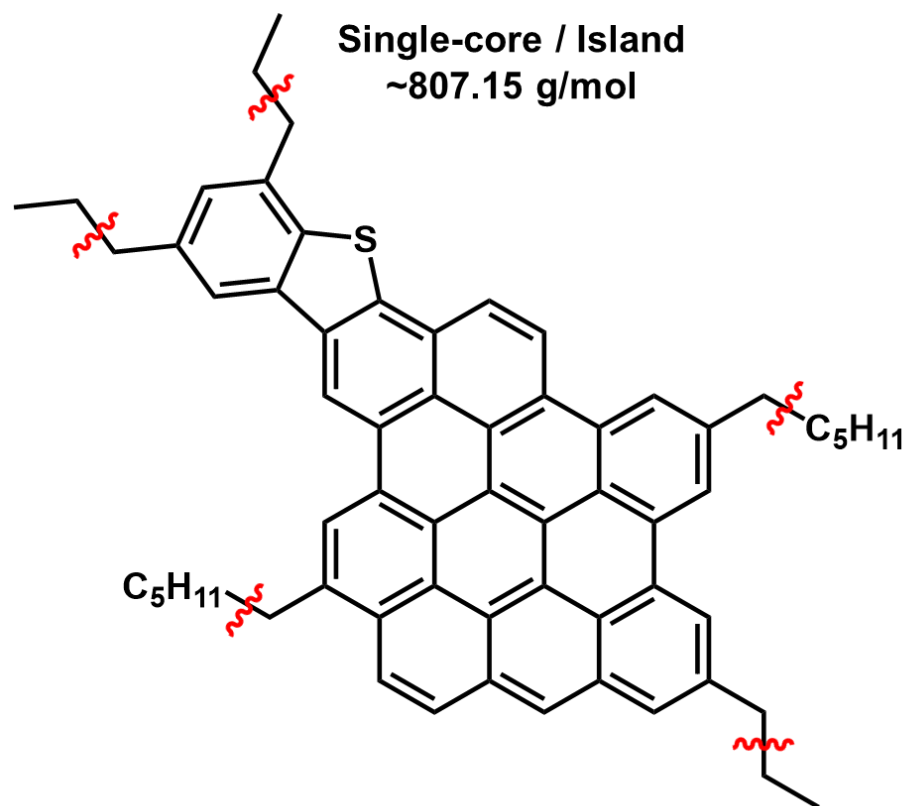
React at 400 °C for 3h

Addition reactions give dimers, trimers etc.  
Methyl groups contribute to addition reactions

## Mesophase Pitch



## Which one makes “good” carbon fibers?





# Carbon Fibres from Bitumen Feedstocks

## Chemistry and Properties of Carbon Fiber Feedstocks from Bitumen Asphaltenes.

Martha L. Chacón-Patiño<sup>1,2\*</sup>, Anika Neumann<sup>3,4\*</sup>, Christopher P. Rüger<sup>2,3,4\*</sup>, Paolo G. Bomber<sup>5</sup>, Lukas Friederici<sup>3,4</sup>, Ralf Zimmermann<sup>3,4</sup>, Erik Frank<sup>6</sup>, Philipp Kretsch<sup>6</sup>, Michael R. Buchmeiser<sup>6,8</sup>, Murray R. Gray<sup>6,7</sup>

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<sup>2</sup>International Joint Laboratory-IC2MC: Complex Matrices Molecular Characterization, TRTG, 76700 Harfleur, France.

<sup>3</sup>Joint Mass Spectrometry Centre (JMSC)/Chair of Analytical Chemistry, University of Rostock, 18059 Rostock, Germany

<sup>4</sup>Department Life, Light & Matter (LLM), University of Rostock, 18059 Rostock, Germany

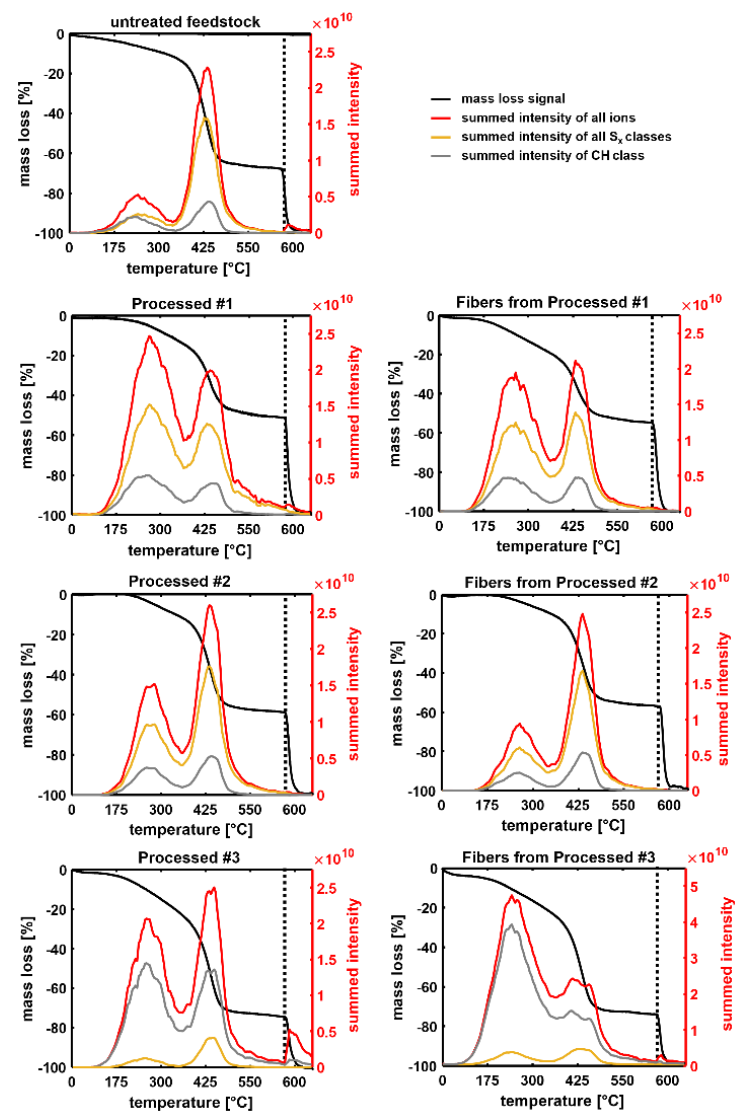
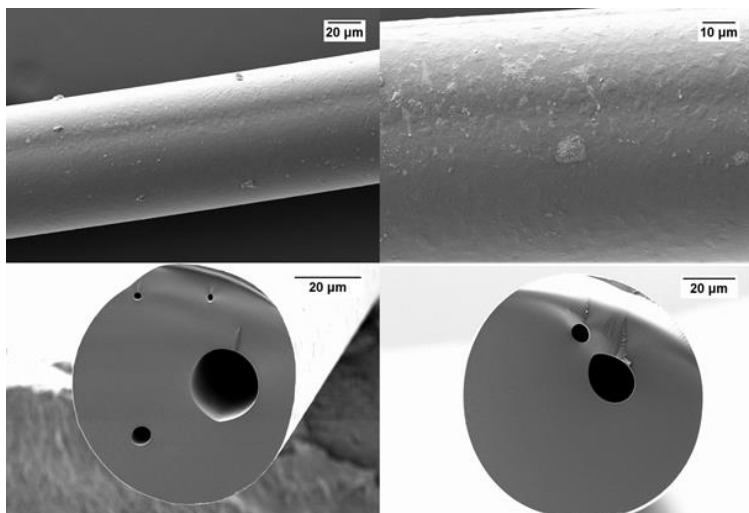
<sup>5</sup>Alberta Innovates Suite 2540, 801-6<sup>th</sup> Ave SW, Calgary AB, T2P 3W2, Canada

<sup>6</sup>German Institutes of Textile- and Fiber Research (DITF) Denkendorf, Körschtalstr. 26, D-73770 Denkendorf, Germany.

<sup>7</sup>Department of Chemical and Materials Engineering, University of Alberta, Edmonton, AB T6G 1H9, Canada

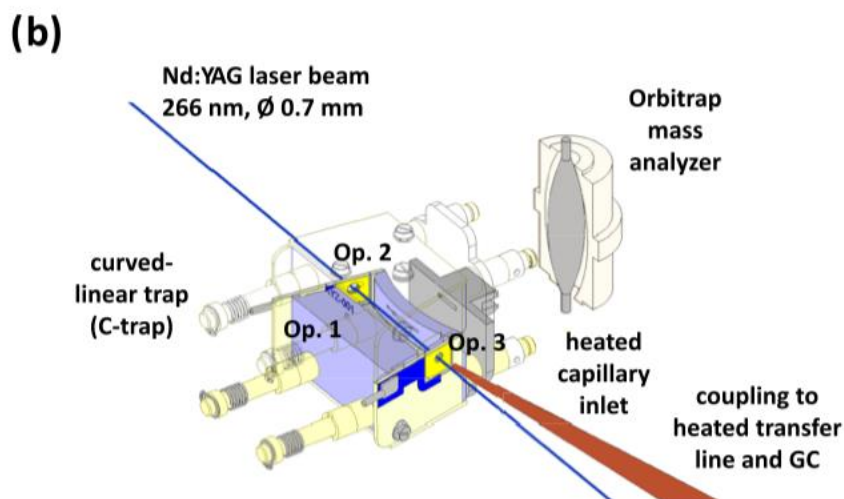
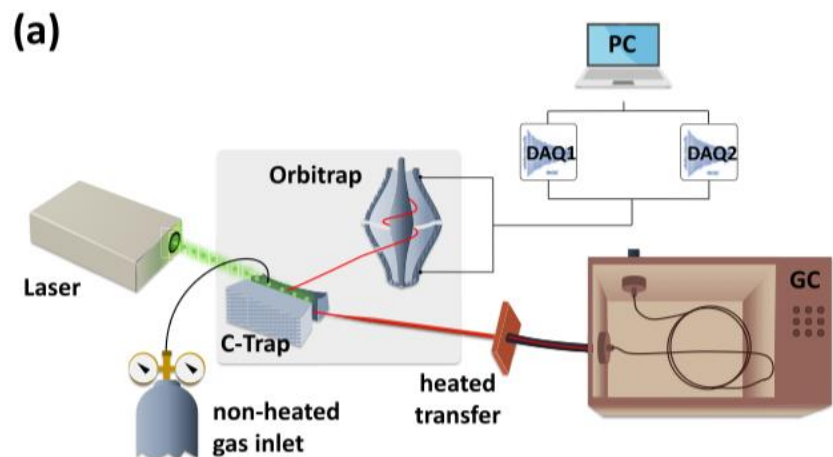
<sup>8</sup>Institute of Polymer Chemistry, University of Stuttgart, Pfaffenwaldring 55, 70569 Stuttgart, Germany

submission to ACS Energy and Fuels



Mass loss data (black) from TG measurements and summed intensity of all ions (red), summed intensity of all  $S_x$ -classes (yellow), and summed intensity of the HC-class (grey) from HRMS measurements for the asphaltene feedstock, the three processed samples, and the fibers. The dotted line indicates the change of the atmosphere from nitrogen to oxygen at 600 °C at the end of the TG measurement.

# Case Study IV – Application on Orbitrap



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## Vacuum Laser Photoionization inside the C-trap of an Orbitrap Mass Spectrometer: Resonance-Enhanced Multiphoton Ionization High-Resolution Mass Spectrometry

Paul Kösling, Christopher P. Rüger,\* Julian Schade, Kyle L. Fort, Sven Ehlert, Robert Irsig, Anton N. Kozhinov, Konstantin O. Nagornov, Alexander Makarov, Martin Rigler, Yury O. Tsybin, Andreas Walte, and Ralf Zimmermann

Cite This: <https://doi.org/10.1021/acs.analchem.1c01018>

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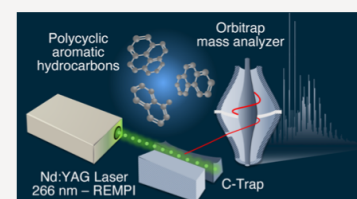
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**ABSTRACT:** State-of-the-art mass spectrometry with ultraviolet (UV) photoionization is mostly limited to time-of-flight (ToF) mass spectrometers with 1000–10 000  $m/\Delta m$  mass resolution. However, higher resolution and higher spectral dynamic range mass spectrometry may be indispensable in complex mixture characterization. Here, we present the concept, implementation, and initial evaluation of a compact ultrahigh-resolution mass spectrometer with gas-phase laser ionization. The concept is based on direct laser photoionization in the ion accumulation and ejection trap (C-trap) of an Orbitrap mass spectrometer. Resonance-enhanced multiphoton ionization (REMPI) using 266 nm UV pulses from a frequency-quadrupled Nd:YAG laser was applied for selective and efficient ionization of monocyclic and polycyclic aromatic hydrocarbons. The system is equipped with a gas inlet for volatile compounds and a heated gas chromatography coupling. The former can be employed for rapid system  $m/z$ -calibration and performance evaluation, whereas the latter enables analysis of semivolatile and higher-molecular-weight compounds. The capability to evaluate complex mixtures is demonstrated for selected petrochemical materials. In these experiments, several hundred to over a thousand compounds could be attributed with a root-mean-square mass error generally below 1 ppm and a mass resolution of over 140 000 at 200  $m/z$ . Isobaric interferences could be resolved, and narrow mass splits, such as 3.4 mDa ( $\text{SH}_4/\text{C}_3$ ), are determined. Single laser shots provided limits of detection in the 20-ppb range for p-xylene and 1,2,4-trimethylbenzene, similar to compact vacuum REMPI-ToF systems.

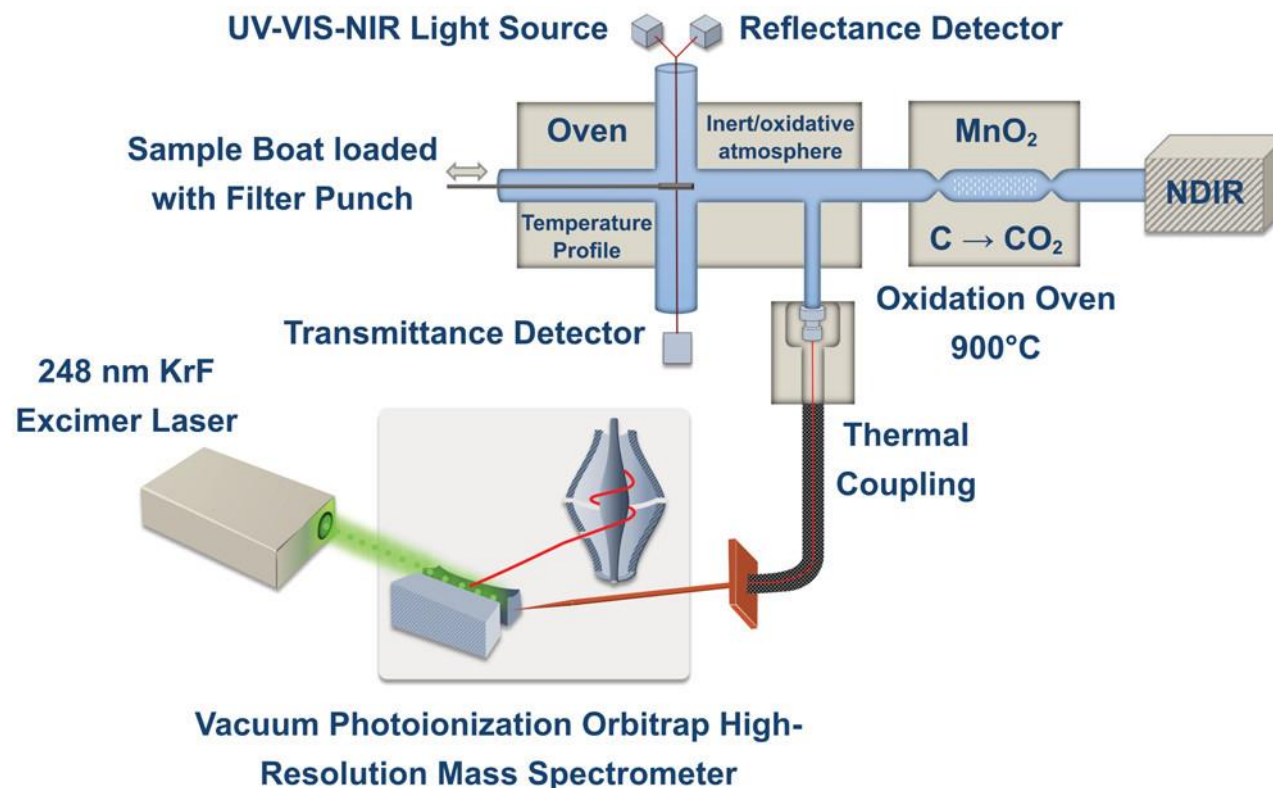
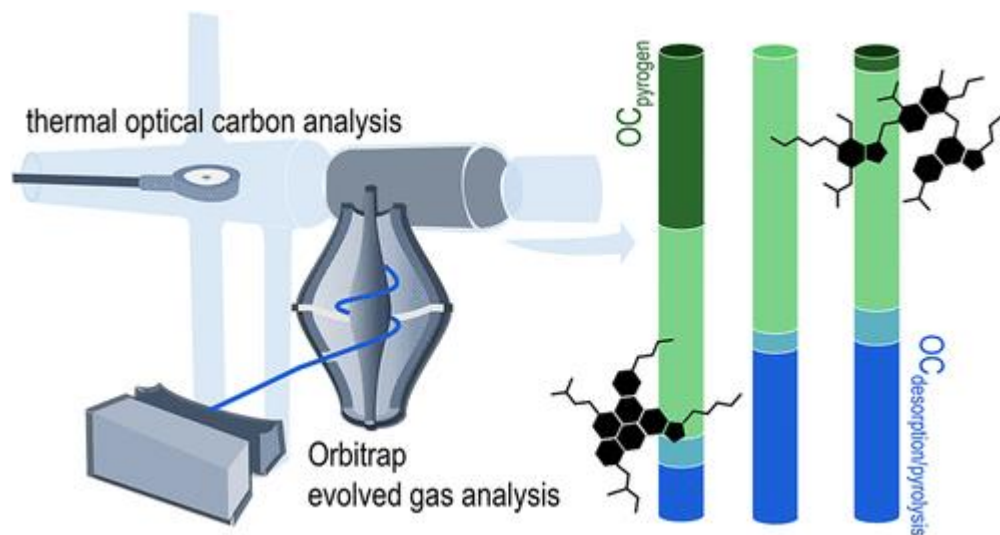


## Addressing Thermal Behavior and Molecular Architecture of Asphaltenes by a Thermal-Optical Carbon Analyzer Coupled to High-Resolution Mass Spectrometry

Christopher P. Rüger,\* Anika Neumann, Paul Kösling, Silvia Juliana Vesga Martínez, Martha Liliana Chacón-Patiño, Ryan P. Rodgers, and Ralf Zimmermann

Cite This: <https://doi.org/10.1021/acs.energyfuels.2c02122>

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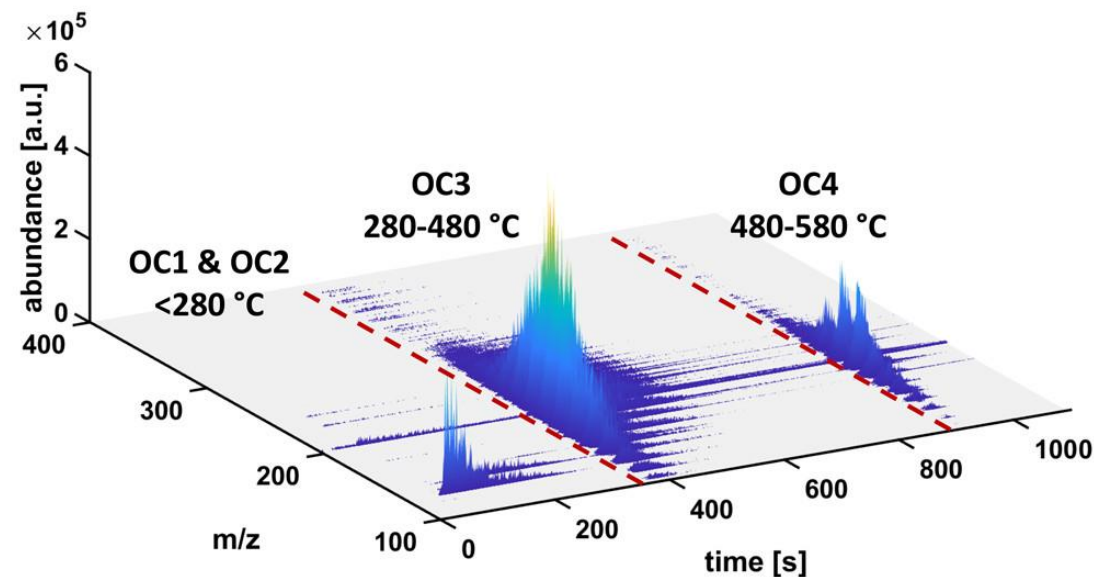
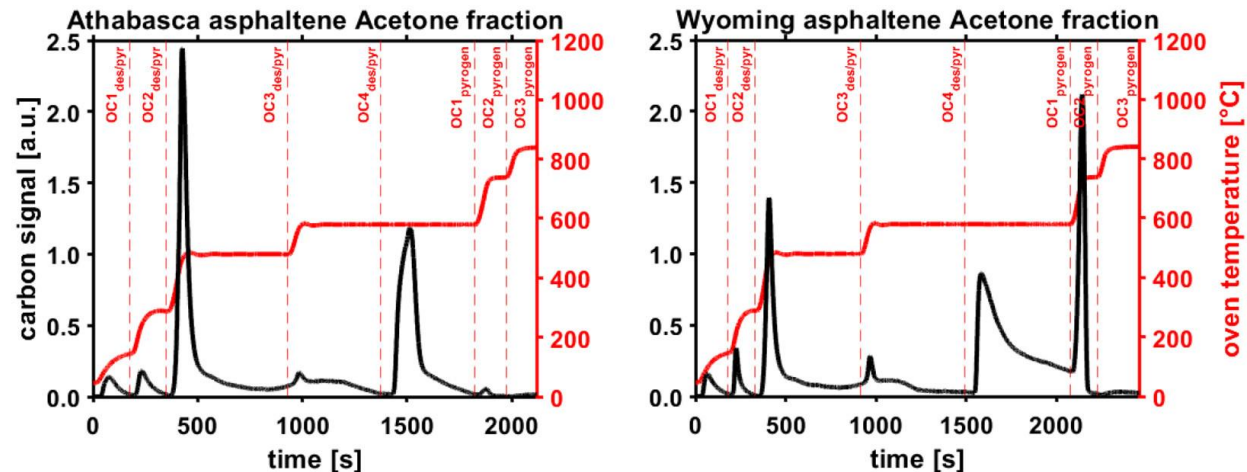
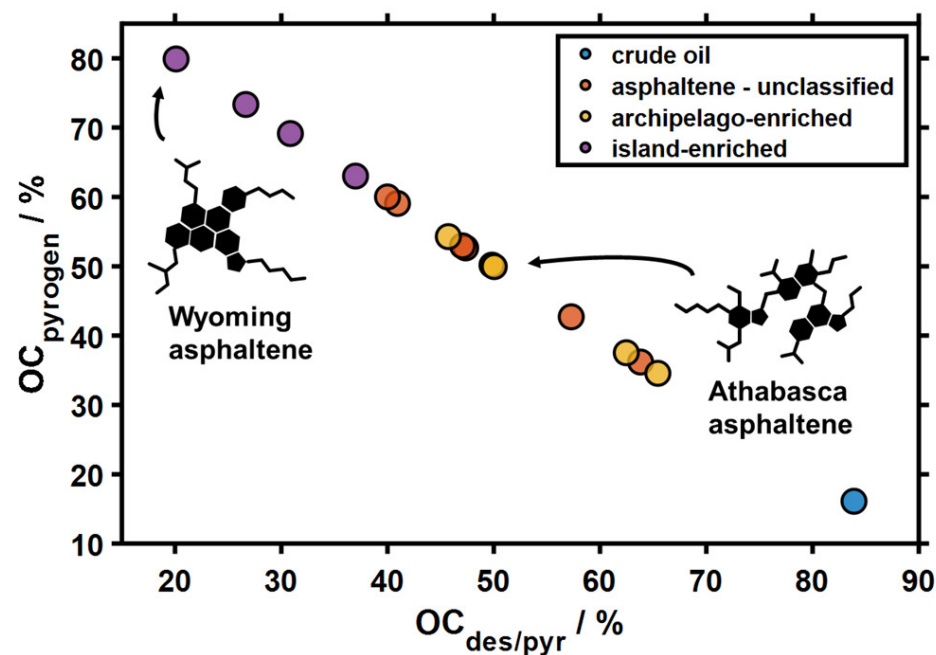


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Cite This: <https://doi.org/10.1021/acs.energyfuels.2c02122>

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## Online Tracing of Ship Emission via High-Resolution Mass Spectrometry

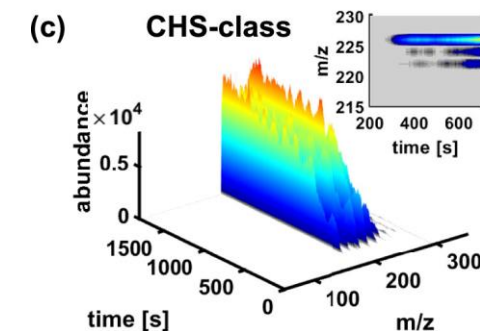
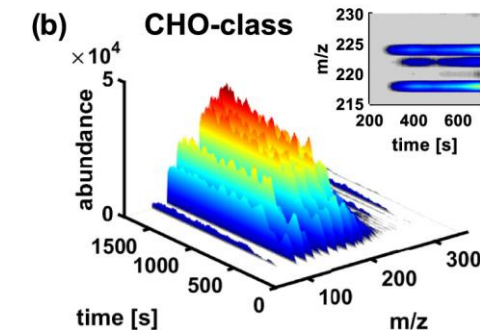
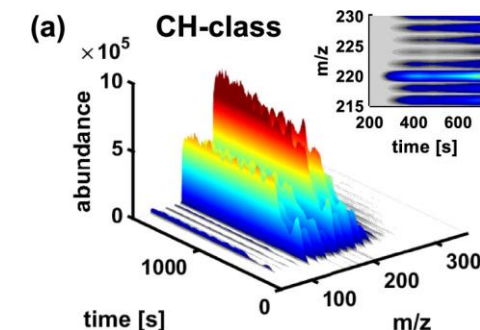
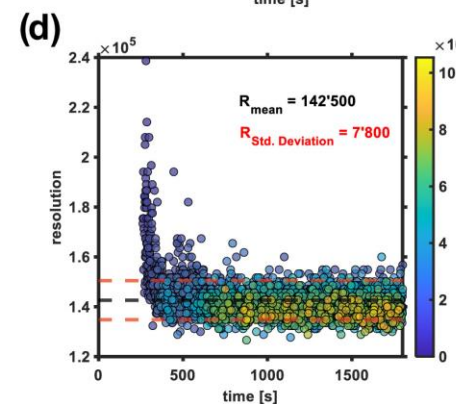
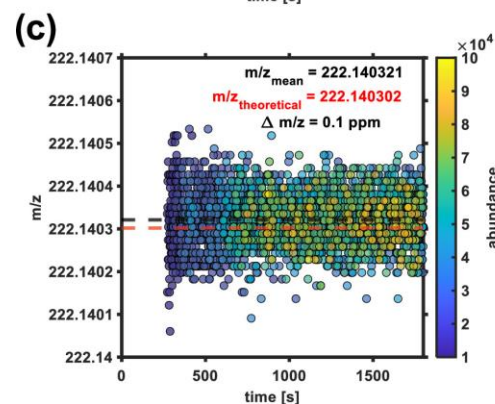
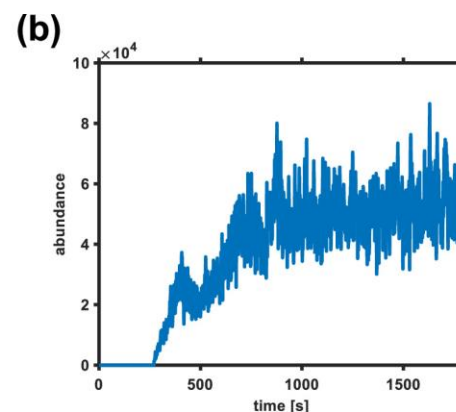
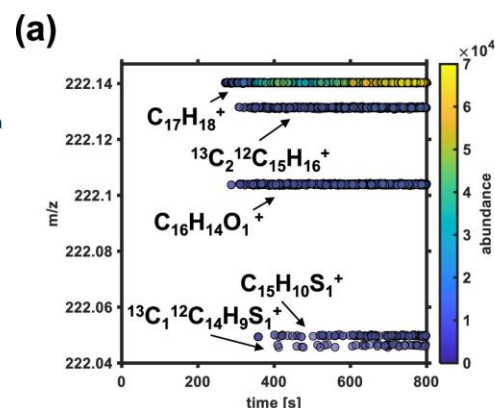
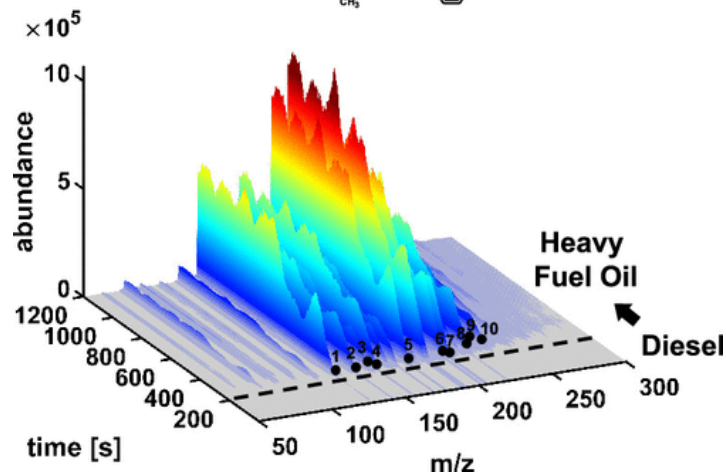
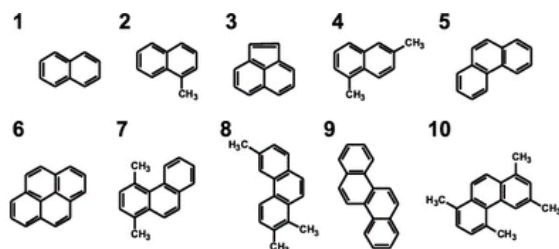
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Article

### Real-Time Investigation of Primary Ship Engine Emissions by Vacuum Resonance-Enhanced Multiphoton Ionization High-Resolution Orbitrap Mass Spectrometry

Paul Kösling, Christopher P. Rüger,\* Julian Schade, Sven Ehlert, Uwe Etzien, Anton N. Kozhinov, Yury O. Tsybin, Martin Rigler, Thomas Adam, Andreas Walte, Bert Buchholz, and Ralf Zimmermann





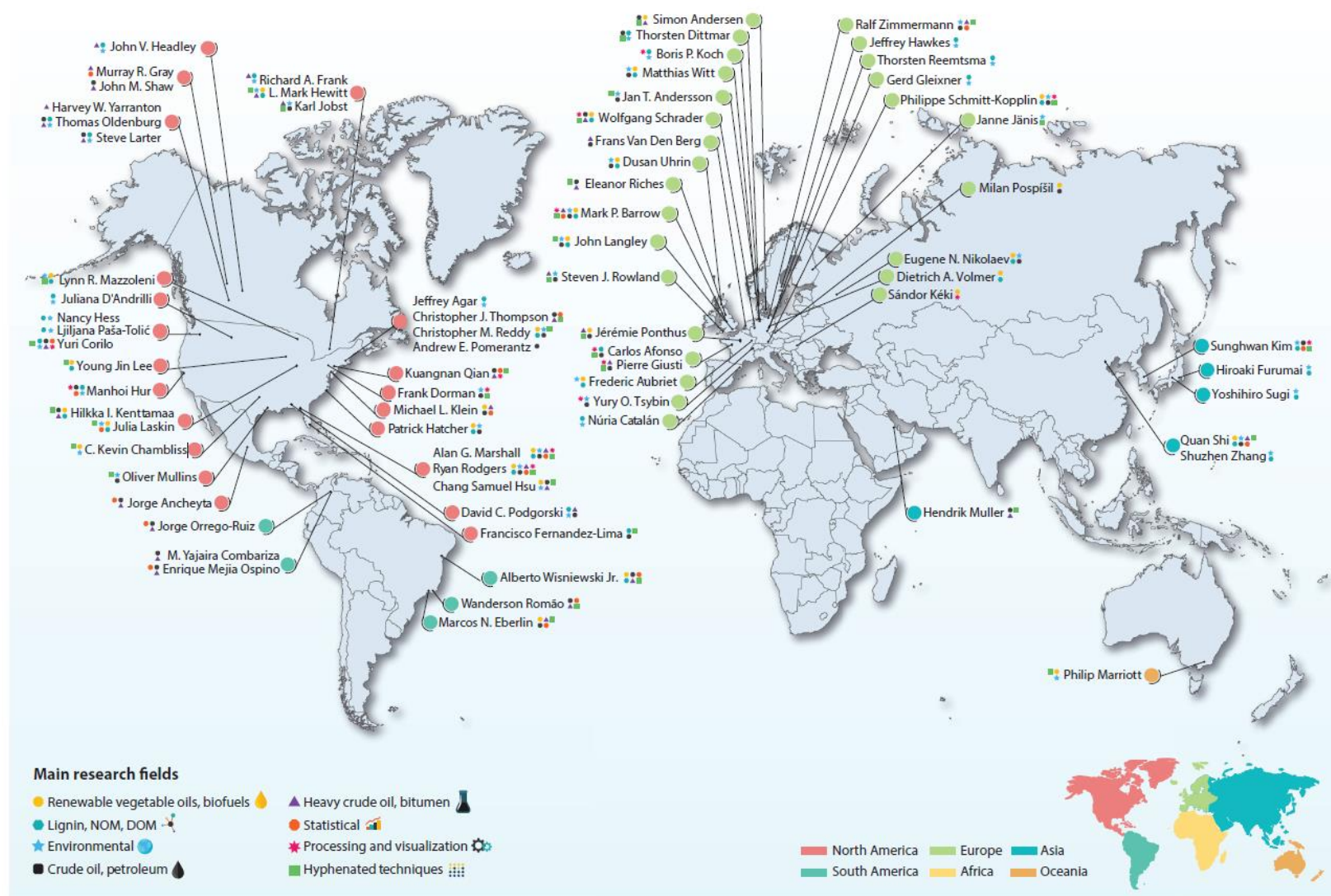
*Annual Review of Analytical Chemistry*

## Petroleomics: Tools, Challenges, and Developments

Diana Catalina Palacio Lozano,<sup>1</sup> Mary J. Thomas,<sup>1,2</sup>  
Hugh E. Jones,<sup>1,2</sup> and Mark P. Barrow<sup>1</sup>

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email: M.P.Barrow@warwick.ac.uk

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Coventry CV4 7AL, United Kingdom



## Petroinformatics

Manhoi Hur, Sunghwan Kim, Chang Samuel Hsu

Studies on petroleomics have been focused on advanced molecular-level characterization of compounds that could not be analyzed by conventional techniques. The next stage of the development would be more discussions on the information obtained and relationships with the properties and functions. The relationship between molecular composition and bulk properties or functions can be explicitly expressed by *petroinformatics*, which utilizes statistics, mathematics, and computational visualization technology to interpret or correlate analytical results with bulk properties and experimental data. This provides explicit or implicit information for underlying science and engineering. In this chapter, several examples of petroinformatics are presented. Statistical methods, such as principle component analysis (PCA) for dimensionality reduction in multivariate analysis, and hierarchical clustering analysis (HCA), have been applied to interpret complex petroleum mass spectra obtained by ultrahigh-resolution Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR MS). The mass spectral peaks were statistically analyzed by Spearman's rank correlation, and by correlation diagrams showing relationships between composition and bulk properties. Additionally, the chapter demonstrates quantitative analyses for petroleum samples by PCA for multivariate analysis and t-tests for univariate analysis. Volcano plots are utilized to visualize the quantitative change or difference between samples in detail.

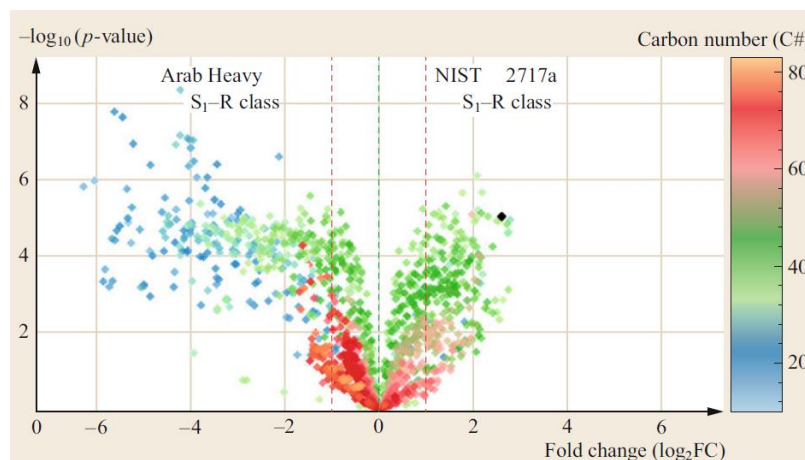
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The software platform, which integrates data from many samples obtained from different analytical instruments, is a very important tool to achieve more comprehensive understanding of complex analytes such as crude oils. The learnings from other research fields, such as metabolomics, genomics, and proteomics, are important and valuable for the next steps of petroinformatics development, i.e., standardization of data and retrieval of its metadata information.

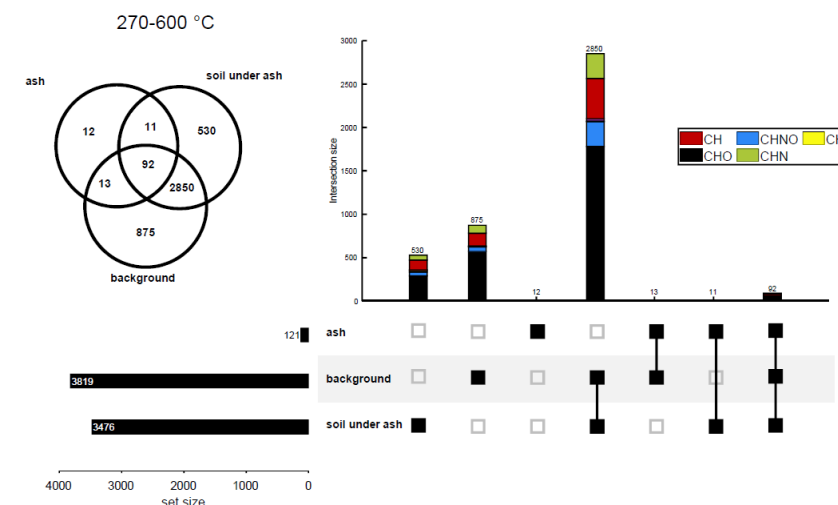
In modern society, information is generated by studies, experiments, surveys, experience and other communications. Information technology, or informatics, vastly improves access to such information by people who support education, research, practice, development, and decision making. Thanks to advances in computer technology, a huge amount of information from various sources can be digitized, integrated, stored, and pro-

cessed by computers with high speed and efficiency. *Petroinformatics*, which utilizes statistics, mathematics, and computational visualization technology to interpret or correlate analytical results with bulk properties and experimental data, is particularly needed for very complex petroleum mixtures, ranging from light gas to solid residue that can contain millions of molecules. Sophisticated analytical techniques generate data that often

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C.S. Hsu, P.R. Robinson (Eds.), *Springer Handbook of Petroleum Technology*, DOI 10.1007/978-3-319-49347-3\_4



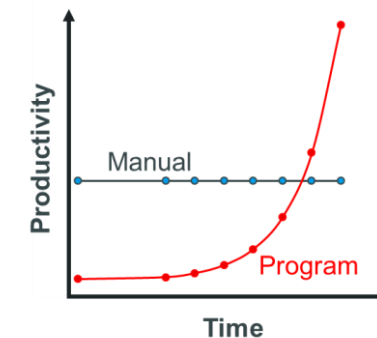
- **supervised and unsupervised multivariate data analysis** for accessing similarities and differences in large attribution data sets
- always take hypothesis tests into account (if data set allows it), such as **ANOVA/MANOVA** for creating powerful and easy significance and fold-change visualizations (Volcano Plot)
- primarily via principle component analysis (**PCA**), clustering approaches (e.g. hierarchical clustering – **HCA**), matrix factorization (NNMF/PMF)
- **Venn plots** as starting point for elucidation shared and unique chemical space (upset plots for nice visualization)

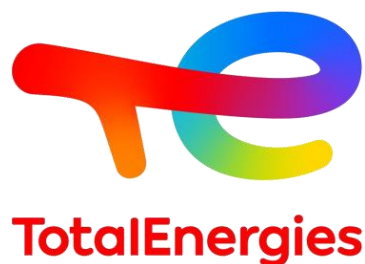




## “Are we there yet?”

- all steps of data processing are important in FT-ICR MS and require attention for pushing to the best results
- but peak-picking and transient processing (FT/non-FT, absorption/magnitude mode, etc.) are very well advanced
- current challenges are primarily for elemental composition attribution with **optimized validation strategies**
- efficient open-code and free workflows **tackling big data from hyphenated solutions** remain rare
- for most complex mixture fields (energy transition, environmental, etc.) we are just at the starting point for **comprehensive data treatment with supervised/non-supervised statistics** (ANOVA, HCA/PCA; NMR
- plenty of room for future (your?) research → learn a programming language
  - it will always help you either in academia or industry





*Complete team of:*

Joint Mass Spectrometry Centre Rostock and Munich - JMSC

International Joint Laboratory Complex Matrices Molecular Characterization- iC2MC



Thank you for the kind attention!

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